PERCHLORATES 23

3. HEALTH EFFECTS

3.1 INTRODUCTION

The primary purpose of this chapter is to provide public health officials, physicians, toxicologists, and other interested individuals and groups with an overall perspective on the toxicology of perchlorates. It contains descriptions and evaluations of toxicological studies and epidemiological investigations and provides conclusions, where possible, on the relevance of toxicity and toxicokinetic data to public health.

A glossary and list of acronyms, abbreviations, and symbols can be found at the end of this profile.

The perchlorate anion forms salts with a wide variety of cations. There are five perchlorate salts that are manufactured in substantial amounts: magnesium, potassium, ammonium, sodium, and lithium perchlorate (see Section 4.1). The potassium, sodium, and ammonium salts are the ones most commonly encountered in the toxicology literature. Therefore, data on potassium, sodium, ammonium, and other perchlorate salts were considered pertinent to the assessment of the perchlorate anion. Perchloric acid was not included because it is a strong acid and its toxicity is dominated by the irritating effects of the hydrogen cation. In the absence of water, the five commercial perchlorates listed above will exist as a solid. In water, perchlorate salts (perchlorates) will rapidly dissolve and completely dissociate into the perchlorate anion, also referred to as perchlorate, and the corresponding metal cation. Potassium, ammonium, and sodium cations are ubiquitous in the environment and are considered spectator ions. Therefore, the species of concern in this document is the perchlorate anion.

Perchlorate has only recently been identified as high priority environmental contaminants, after the development of a new, more sensitive analytical method for its detection in water in April 1997 and its subsequent discovery in drinking water supplies in California, Nevada, and Utah. An expert peer review committee was convened in May 1997 by the EPA to recommend studies that were needed to address the key data gaps in the health effects database for perchlorate. A broad array of studies were recommended, including a 90-day oral subchronic bioassay, a two-generation reproductive toxicity study, a developmental toxicity study, a neurobehavioral developmental study, pharmacokinetic and mechanistic studies, genotoxicity assays, and immunotoxicity studies. Most of these studies have now been completed and some of them have been published in the open literature; others, while completed, have not yet been published in peer-reviewed journals. However, unpublished versions were made available to

interested parties by the EPA. The discussion of health effects provided below is based primarily on the newer literature, which provides information on dose-response relationships for low-dose effects on perchlorate's target organ, the thyroid gland, in adults and developing organisms.

3.2 DISCUSSION OF HEALTH EFFECTS BY ROUTE OF EXPOSURE

To help public health professionals and others address the needs of persons living or working near hazardous waste sites, the information in this section is organized first by route of exposure (inhalation, oral, and dermal) and then by health effect (death, systemic, immunological, neurological, reproductive, developmental, genotoxic, and carcinogenic effects). These data are discussed in terms of three exposure periods: acute (14 days or less), intermediate (15–364 days), and chronic (365 days or more).

Levels of significant exposure for each route and duration are presented in tables and illustrated in figures. The points in the figures showing no-observed-adverse-effect levels (NOAELs) or lowestobserved-adverse-effect levels (LOAELs) reflect the actual doses (levels of exposure) used in the studies. LOAELs have been classified into "less serious" or "serious" effects. "Serious" effects are those that evoke failure in a biological system and can lead to morbidity or mortality (e.g., acute respiratory distress or death). "Less serious" effects are those that are not expected to cause significant dysfunction or death, or those whose significance to the organism is not entirely clear. ATSDR acknowledges that a considerable amount of judgment may be required in establishing whether an end point should be classified as a NOAEL, "less serious" LOAEL, or "serious" LOAEL, and that in some cases, there will be insufficient data to decide whether the effect is indicative of significant dysfunction. However, the Agency has established guidelines and policies that are used to classify these end points. ATSDR believes that there is sufficient merit in this approach to warrant an attempt at distinguishing between "less serious" and "serious" effects. The distinction between "less serious" effects and "serious" effects is considered to be important because it helps the users of the profiles to identify levels of exposure at which major health effects start to appear. LOAELs or NOAELs should also help in determining whether or not the effects vary with dose and/or duration, and place into perspective the possible significance of these effects to human health.

The significance of the exposure levels shown in the Levels of Significant Exposure (LSE) tables and figures may differ depending on the user's perspective. Public health officials and others concerned with appropriate actions to take at hazardous waste sites may want information on levels of exposure associated with more subtle effects in humans or animals (LOAELs) or exposure levels below which no

adverse effects (NOAELs) have been observed. Estimates of levels posing minimal risk to humans (Minimal Risk Levels or MRLs) may be of interest to health professionals and citizens alike.

A User's Guide has been provided at the end of this profile (see Appendix B). This guide should aid in the interpretation of the tables and figures for Levels of Significant Exposure and the MRLs.

3.2.1 Inhalation Exposure

3.2.1.1 Death

No studies were located regarding lethality in humans or animals after inhalation exposure to perchlorate.

3.2.1.2 Systemic Effects

No studies were located regarding respiratory, cardiovascular, gastrointestinal, musculoskeletal, dermal, or ocular effects in humans or animals after inhalation exposure to perchlorate.

The highest NOAEL values for systemic effects from the two occupational studies available are recorded in Table 3-1 and plotted in Figure 3-1.

Hematological Effects. No hematological effects were found in ammonium perchlorate workers (22–31 high-dose and 18–27 low-dose versus 72–150 controls) exposed for 1–27 years (mean=8.3 years) to average perchlorate concentrations of up to 0.63 mg/m³ (Gibbs et al. 1998). The researchers estimated an average cumulative lifetime perchlorate absorbed dose of 38 mg/kg in the high-dose workers in this study, which corresponds to a daily dose of 0.01 mg/kg/day based on the approximate average exposure duration of 9 years for high-dose workers. Oral exposure due to deposition in the mouth and throat was also likely to have occurred. The accuracy of dose estimates from this study is questionable, however, because the researchers estimated the fraction absorbed using a study on an unrelated chemical and did not consider the size of the inhaled ammonium particles in their calculations. Particle size (mean and distribution) is an important determinant of inhaled dose for particulates (EPA 1994). A similar study of 37 ammonium perchlorate workers also found no evidence of hematological effects among the workers (Lamm et al. 1999). The workers were assigned to one of three categories of presumptive exposure based

Table 3-1. Levels of Significant Exposure to Perchlorates - Inhalation

		Exposure/			L	OAEL	
Key t		duration/ frequency	System	NOAEL (mg/m3)	Less serious (mg/m3)	Serious (mg/m3)	Reference Chemical Form
	CHRONIC E	EXPOSURE					
	Systemic						
1	Human	1-27 yr	Hemato	0.63			Gibbs et al. 1998
		(avg=8.3 yr)					NH4CIO4
			Hepatic	0.63			
			Renal	0.63			
			Endocr	0.63			
2	Human	40% over 5	Hemato	0.86			Lamm et al. 1999
_		years					NH4CIO4
			Hepatic	0.86			
			Renal	0.86			
			Endocr	0.86			

^a The number corresponds to entries in Figure 3-1.

endocr = endocrine; hemato = hematological; LOAEL = lowest-observed-adverse-effect level; NOAEL = no-observed-adverse-effect level

Figure 3-1. Levels of Significant Exposure to Perchlorates - Inhalation Chronic (≤365 days)

Systemic

า3 _	Hemale	Hepatie	Renal	Endocr.		
1						
	△2	△2	△2	riangle2		
	∴∠	∴∠	∴∠	∴ £		
	riangle1	△1	△1	Δ 1		
c-Cat	-Humans	f-Ferret	n-Mink o-Other	Cancer Effect Level-Animals	▼ Cancer Effect Level-Humans	■LD50/LC50
d-Dog r-Rat	k-Monkey m-Mouse	j-Pigeon e-Gerbil s-Hamster g-Guinea P	o-Other	◆ Cancer Effect Level-Animals ● LOAEL, More Serious-Animals ● LOAEL, Less Serious-Animals ○ NOAEL - Animals	▼Cancer Effect Level-Humans ▲ LOAEL, More Serious-Humans ▲ LOAEL, Less Serious-Humans △ NOAEL - Humans	Minimal Risk L for effects other than Cancer
p-Pig q-Cow	a-Sheep	g-Guinea F	ig	ONOAEL - AHIIIIdis	△NOAEL - ΠυΠΙΙΙΙΙΝ	Cancer

on visible dust generated. The average airborne exposure for the high-exposure group was 8.6 mg/day (respirable fraction; particle size 0.1– $10 \mu m$) or 59.4 mg/day (total particulate perchlorate). Dividing by the default inhalation volume of 10 m^3 /day results in a respirable concentration of 0.86 mg/m^3 . The absorbed oral dose per shift was calculated using urinary perchlorate measurements and the assumption that the absorbed dose that is excreted is 95%. In the low-, medium-, and high-exposure categories, the absorbed doses were estimated to be 4, 11, and 34 mg perchlorate/day, respectively. Assuming a body weight of 70 kg, the 34 mg/kg oral dose corresponds to about 0.5 mg perchlorate/kg/day. Measures of cumulative exposure were not considered in this study.

No studies were located regarding hematological effects in animals after inhalation exposure to perchlorate.

Hepatic Effects. No effects on serum enzymes indicative of liver toxicity were found in the ammonium perchlorate workers studied by Gibbs et al. (1998) or among those studied by Lamm et al. (1999) (see Hematological Effects above for further details on these studies). No further relevant information was located.

No studies were located regarding hepatic effects in animals after inhalation exposure to perchlorate.

Renal Effects. No effects on serum enzymes indicative of kidney toxicity or in serum creatinine and blood urea nitrogen (BUN) were found in the ammonium perchlorate workers evaluated by Gibbs et al. (1998) or Lamm et al. (1999) (see Hematological Effects above for further details on these studies).

No studies were located regarding renal effects in animals after inhalation exposure to perchlorate.

Endocrine Effects. No significant effects on serum levels of TSH, total serum thyroxine (TT4), T3, or free T4 index (FTI) were found among the ammonium perchlorate workers studied by Gibbs et al. (1998). The mean airborne concentration of perchlorate to which the workers were exposed ranged from 0.02 to 0.63 mg/m³. The researchers estimated that exposure to airborne perchlorate provided an average cumulative lifetime absorbed dose of up to 0.01 mg perchlorate/kg/day for high-exposure workers. Comparison of pre- and post-shift serum thyroid hormone measurements for individual workers failed to find any evidence of a transient effect associated with daily exposure. However, it appeared that the thyroid values of the unexposed workers fell in between those of the low-exposure and high-exposure groups, suggesting that important confounding was present. In the occupational-exposure study

conducted by Lamm et al. (1999), there were also no significant alterations in serum TSH, T3, T4, FTI, thyroid hormone binding ratio, or thyroid peroxidase antibody concentrations among the workers. In this study, it was estimated that the high-exposure workers, who were exposed to an average of 0.86 mg of respirable airborne perchlorate particles/m³, absorbed doses of approximately 0.5 mg perchlorate/kg/day (see above under Hematological Effects for further details on these studies). A study conducted in the same manufacturing facility studied by Lamm et al. (1999) found that intermittent, high exposure to perchlorate for many years did not induce goiter or any evidence of hypothyroidism among the workers as judged by no significant alterations in serum TSH or thyroglobulin even though iodine uptakes were decreased during the work shift (Braverman et al. 2005). The median estimated absorbed dose was 0.167 mg/kg/day, equivalent to drinking approximately 2 L of water containing 5 mg perchlorate/L.

No studies were located regarding endocrine effects in animals after inhalation exposure to perchlorate.

No studies were located regarding the following effects in humans or animals after inhalation exposure to perchlorate:

- 3.2.1.3 Immunological and Lymphoreticular Effects
- 3.2.1.4 Neurological Effects
- 3.2.1.5 Reproductive Effects
- 3.2.1.6 Developmental Effects
- 3.2.1.7 Cancer
- 3.2.2 Oral Exposure
- 3.2.2.1 Death

Several cases of human deaths were reported among hyperthyroid patients treated with potassium perchlorate (Barzilai and Sheinfeld 1966; Fawcett and Clarke 1961; Gjemdal 1963; Hobson 1961; Johnson and Moore 1961; Krevans et al. 1962). Deaths were due to aplastic anemia or severe agranulocytosis and were considered to be causally related to potassium perchlorate. The lethal doses in these patients were in the low-to-moderate range of doses employed in thyrotoxicosis therapy: 600–1,000 mg potassium perchlorate/day, or roughly 5–12 mg perchlorate/kg/day. The patients had received treatment for anywhere between 2 and 8 months. All of the deaths were females (Graves' disease, the

most common cause of hyperthyroidism, is far more common in women than in men) and their ages ranged from 24 to 82 years.

Gauss (1972) reported an LD_{50} dietary concentration of 3.55% (approximately 3,621 mg perchlorate/kg/day) for potassium perchlorate in mice exposed for up to 30 days. The first deaths occurred within 4 days of the start of treatment.

The LD_{50} value for mice is recorded in Table 3-2 and plotted in Figure 3-2.

3.2.2.2 Systemic Effects

The highest NOAEL values and all LOAEL values from each reliable study for systemic effects in each species and duration category are recorded in Table 3-2 and plotted in Figure 3-2.

Respiratory Effects. No studies were located regarding respiratory effects in humans after oral exposure to perchlorate. The only relevant information in animals is that from a study by Siglin et al. (2000) in which no significant effects on lung weight and no gross or microscopic alterations were found in the lungs from rats administered up to 8.5 mg perchlorate/kg/day (as ammonium perchlorate) in the drinking water for up to 90 days.

Cardiovascular Effects. No studies were located regarding cardiovascular effects in humans after oral exposure to perchlorate.

Absolute and relative heart weights were significantly decreased in rats treated with 2% potassium perchlorate (approximately 2,327 mg perchlorate/kg/day) in the drinking water for 6 weeks (MacDermott 1992). No gross or microscopical alterations were observed in the heart of rats administered ammonium perchlorate in the drinking water at doses of up to 8.5 mg perchlorate/kg/day for up to 90 days (Siglin et al. 2000); the weight of the heart was also not affected by exposure to perchlorate.

Gastrointestinal Effects. No information was located regarding gastrointestinal effects of perchlorate in healthy humans. Symptoms of gastrointestinal distress, including nausea and vomiting, have been reported in a small percentage of cases of hyperthyroid patients treated with potassium

Table 3-2. Levels of Significant Exposure to Perchlorates - Oral

		Exposure/				LOAE	iL	
Key to ^a figure	Opcoics	duration/ frequency (Specific route)	System	NOAEL (mg/kg/day)	Less se (mg/k		Serious (mg/kg/day)	Reference Chemical Form
	ACUTE E	EXPOSURE						
	Death							
	Mouse (NMRI)	30 d ad libitum (F)					3621 F (LD50)	Gauss 1972 KCIO4
	Systemic							
2	Human	14 d (W)	Hemato Hepatic Renal Endocr	0.5 0.5 0.5 0.007 ^b	0.1	(42% inhibition of		Greer et al. 2002 KClO4
						radioiodine uptake by the thyroid)		
3	Human	14 d (W)	Hemato	0.14				Lawrence et al. 2000 KCIO4
		(**)	Hepatic	0.14				
			Renal	0.14				
			Endocr	0.14				
4	Human	14 d	Endocr	0.04				Lawrence et al. 2003
		(W)						KCIO4
	Rat (Sprague-	14 d ad libitum	Endocr		0.1	(increased serum TSH in females, decreased T4 in		Caldwell et al. 1995 NH4ClO4
	Dawley)	(W)				males and females, and decrease T3 in females)		
			Bd Wt	39.9				
			Other	39.9				

Table 3-2. Levels of Significant Exposure to Perchlorates - Oral (continued)

		Exposure/ duration/				LOAEL	
Key to		frequency	System	NOAEL (mg/kg/day)	Reference Chemical Form		
6	Rat (Sprague- Dawley)	4 d ad libitum (W)	Endocr	1.4 M	7.2 M (approximately 20% decrease in T3 and decrease inT4)		Mannisto et al. 1979 KCIO4
7	Rat (Sprague-	14 d ad libitum	Resp	8.5			Siglin et al. 2000 NH4ClO4
	Dawley)	(W)	0 "	0.5			
			Cardio	8.5			
			Gastro	8.5			
			Hemato	8.5			
			Musc/skel	8.5			
			Hepatic	8.5			
			Renal	8.5			
			Endocr		0.009 M (approximately 20% decreased serum T males)	6 3 in	
			Dermal	8.5			
			Ocular	8.5			
			Bd Wt	8.5			
			Other	8.5			
8	Rat (Sprague- Dawley)	14 d ad libitum (W)	Endocr		0.09 M (increased TSH and decreased serum T		Yu et al. 2002 NH4ClO4
9	Mouse (B6C3F1)	14 d ad libitum (W)	Endocr	0.05 F	0.2 F (significant decreas serum T4 levels; non-significant incre in TSH)		BRT 2000 NH4ClO4

Table 3-2. Levels of Significant Exposure to Perchlorates - Oral (continued)

		Exposure/ duration/			LOAE	EL	
Key to		frequency (Specific route)	System	NOAEL (mg/kg/day)	Less serious (mg/kg/day)	Serious (mg/kg/day)	Reference Chemical Form
	Mouse	14 d	Hemato	25.5 F			DoD 1999
	(B6C3F1)	ad libitum					NH4CIO4
		(W)	Hepatic	25.5 F			
			Renal	25.5 F			
			Endocr		2.6 F (15% decrease serum T4)		
			Bd Wt	25.5 F			
			Other	25.5 F			
	Immunol	ogical/Lymphor	eticular				
11	Mouse	14 d			0.05 F (increased response to		BRT 2000
	(B6C3F1)	ad libitum			sensitizer		NH4CIO4
		(W)			2,4-dinitrochlorobenzene)		
12	Mouse	14 d		25.5 F			DoD 1999
	(B6C3F1)	ad libitum					NH4CIO4
		(W)					
	Neurolog	ical					
	Rat	14 d		8.5			Siglin et al. 2000
	(Sprague-	ad libitum					NH4ClO4
	Dawley)	(W)					
	Reproduc	ctive					
	Rat	Gd 2-8		532 F			Brown-Grant 1966
	(Wistar)	ad libitum					KCIO4
		(W)					
15	Rat	Gd 1-13		1752 F			Brown-Grant and
	(Wistar)	ad lib					Sherwood 1971
		(W)					KCIO4

Table 3-2. Levels of Significant Exposure to Perchlorates - Oral (continued)

a		Exposure/ duration/				LOAEL	
Key to ⁴ figure	Species (Strain)	frequency (Specific route)	System	NOAEL (mg/kg/day)	Less serious (mg/kg/day)	Serious (mg/kg/day)	Reference Chemical Form
(9	Rat Sprague- Pawley)	14 d ad libitum (W)		8.5			Siglin et al. 2000 NH4ClO4

Table 3-2. Levels of Significant Exposure to Perchlorates - Oral (continued)

_		Exposure/ duration/		_	LOAI	EL	
Key to ^a figure	Opecies	frequency (Specific route)	System	NOAEL (mg/kg/day)	Less serious (mg/kg/day)	Serious (mg/kg/day)	Reference Chemical Form
	INTERM	EDIATE EXPO	SURE				
	Systemic						
17	Human	4 wk (IN)	Endocr		9 M (decreased thyroid I and serum TSH)		Brabant et al. 1992 KClO4
18	Rat (Wistar)	19 wk ad libitum	Hepatic	64 M			Hiasa et al. 1987 KClO4
		(F)	Endocr			64 M (thyroid weight doubled; 24% decrease in serum T4 100% increase in TSH)	;
			Bd Wt	64 M			
19	Rat (Wistar)	6 wk ad lib	Cardio		2327 M (decreased heart weight)		MacDermott 1992 KCIO4
		(W)	Musc/skel		2327 M (decreased membrane potential and intracellular K+ activity)		
			Endocr			2327 M (thyroid weight more than doubled; 71% decrease in serum T4)	
			Bd Wt			2327 M (27% reduced weight at 6 weeks)	
20		25 d	Endocr			175 M (thyroid weight tripled;	Ortiz-Caro et al. 198
	(Wistar)	ad libitum (W)				undetectable levels of T3 and T4 in serum; 50% increase in TSH)	KClO4
			Bd Wt			175 M (40% reduction in weight gain)	
			Metab		175 M (decreased alpha-GPD activity)		

Table 3-2. Levels of Significant Exposure to Perchlorates - Oral (continued)

	_	Exposure/ duration/				LOAE	LOAEL		
Key to		frequency (Specific route)	System	NOAEL (mg/kg/day)		serious kg/day)	Serious (mg/kg/day)	Reference Chemical Form	
21	Rat (Wistar)	45 d Metab 1x/d (G)		Metab		(decrease glucose and increase urea in serum: increased activity of aldolase, LDH, arginase; decrease G-6-Pase)		Sangan and Motlag 1986 KCIO4	
22	Rat (Wistar)	3 mo ad lib (W)	Hemato				1059 M (decreased hematopoeisis)	Shevtsova et al. 199 KCIO4	
	Rat (Sprague- Dawley)	90 d ad libitum (W)	Resp	8.5				Siglin et al. 2000 NH4ClO4	
		(**)	Cardio	8.5					
			Gastro	8.5					
			Hemato	8.5					
			Musc/skel	8.5					
			Hepatic	8.5					
			Renal	8.5					
			Endocr		0.009	(significant decreases in T4 and T3 in both males and females)			
			Dermal	8.5					
			Ocular	8.5					
			Bd Wt	8.5					
			Other	8.5					
24	Rat (Wistar)	45 d 1x/d	Metab		406 N	I (decreased activity of lipase, phospholipase A;		Vijayalakshmi and Motlag 1989a	
		(G)				decreased free fatty acids; increased cholesterol, triglycerides, phospholipids)		NaClO4	

Table 3-2. Levels of Significant Exposure to Perchlorates - Oral (continued)

		Exposure/ duration/				LOAE	L	
Key to	Op00.00	frequency (Specific route)	System	NOAEL (mg/kg/day)		serious g/day)	Serious (mg/kg/day)	Reference Chemical Form
	Rat (Sprague- Dawley)	>19 wk ad libitum (W)	Endocr	0.26	2.6	(increased absolute and relative thyroid weight in both sexes; hypertrophy and hyperplasia in males; increased TSH)		York et al. 2001a NH4ClO4
			Bd Wt Other	25.5 25.5				
	Rat (Sprague- Dawley)	ad libitum	Endocr	0.26	2.6	(hypertrophy/hyperplasia of the thyroid)		York et al. 2001a NH4ClO4
	•	(W)	Bd Wt Other	25.5	0.26 M	(decreased water consumption)		
	Rat (Sprague- Dawley)	15 d premating Gd 1-21 ad libitum	Endocr		0.009	(increase serum TSH and decrease T4)		York et al. 2003 NH4ClO4
		(W)	Bd Wt	25.5				
	Rat (Sprague- Dawley)	31 d Gd 0-21 Pnd 1-10 ad libitum	Bd Wt	8.5				York et al. 2004 NH4ClO4
		(W)						
	Mouse (B6C3F1)	90 d ad libitum (W)	Endocr	0.02 F	0.05 F	(17% increase in serum TSH)		BRT 2000 NH4ClO4

Table 3-2. Levels of Significant Exposure to Perchlorates - Oral (continued)

		Exposure/ duration/				LOAE	L		
(ey to [°] figure	Opecies	frequency (Specific route)	System	NOAEL (mg/kg/day)	Less s (mg/k	erious g/day)	Serious (mg/kg/day)		Reference Chemical Form
	Mouse (B6C3F1)	90 d ad libitum	Hemato	25.5 F					DoD 1999 NH4ClO4
		(W)	Hepatic	25.5 F					
			Renal	25.5 F					
			Endocr	0.09 F	0.85 F	(significant decrease in serum T4 levels)	25.5 F	(colloid depletion; intrafollicular capillary congestion)	
			Bd Wt Other	25.5 F 25.5 F					
	Mouse (BALB/c)	3 mo ad libitum	Hemato				1750 M	(decreased hematopoiesis)	Shevtsova et al. 199 KCIO4
		(W)							
32	Gn Pig	30 60 90d	Endocr				531 F	(thyroid weight almost	Postel 1957
	(NS)	ad libitum (W)						tripled; thyroid hyperplasia and colloid depletion)	KCIO4
	Dog (NS)	3 wk 1x/d	Gastro		3811	(mucosal irritation)			Selivanova and Vorobieva 1969
		(G)	Hemato				3811	(inhibition of hematopoeisis in bone marrrow)	NH4ClO4
			Endocr				3811	(inhibited thyroid function)	
	Immunol	ogical/Lymphor	eticular						
34	Rat (Sprague-	90 d ad libitum		8.5					Siglin et al. 2000 NH4ClO4
	Dawley)	(W)							
35	Mouse	90 d		0.02 F	0.05 F	(increased sensitization			BRT 2000
	(B6C3F1)	ad libitum (W)				response to 2,4-dinitrochlorobenzene)			NH4ClO4

Table 3-2. Levels of Significant Exposure to Perchlorates - Oral (continued)

		Exposure/ duration/				LOA	EL		
Key to	Opcoics	frequency (Specific route)	System	NOAEL (mg/kg/day)		serious kg/day)	Serious (mg/kg/day)	Reference Chemical	
	Mouse (B6C3F1)	90 d ad libitum		25.5 F				DoD 1999 NH4ClO4	
		(W)							
	Neurolog	ical							
	(Sprague-	90 d ad libitum		8.5				Siglin et al. NH4ClO4	2000
	Dawley)	(W)							
	Reproduc	ctive							
	Rat (Sprague- Dawley)	90 d ad libitum (W)		8.5				Siglin et al. NH4ClO4	2000
	Rat (Sprague-	16 wk ad libitum		25.5				York et al. 2 NH4ClO4	2001a
	Dawley)	(W)							
	Developn	nental							
40	Rat (Sprague- Dawley)	2 wk premating Gd 1-21 PND 1-10 ad libitum		8.5				Bekkedal et NH4ClO4	t al. 2000
		(W)							
	Rat (Sprague- Dawley)	31 d Gd 2-21 PND 1-10 ad libitum			1	(increased TSH and decreased T4 in pups exposed via maternal milk)		Mahle et al. NH4ClO4	. 2003
		(W)							
42	Rat (Sprague- Dawley)	16 wk ad libitum (W)		0.26	2.6	(thyroid hypertrophy and hyperplasia in F1 females and in F2 males and females)		York et al. 2 NH4ClO4	2001a

Table 3-2. Levels of Significant Exposure to Perchlorates - Oral (continued)

V 0	2	Exposure/ duration/				LOAE	<u>L</u>		_
Key to figure	Opcoics	frequency (Specific route)	System	NOAEL (mg/kg/day)		serious kg/day)	Seriou (mg/kg/		Reference Chemical Form
	Rat (Sprague- Dawley)	15 d premating Gd 1-21 ad libitum			0.009	(17.6% decreased serum T3 in pups)			York et al. 2003 NH4ClO4
		(W)							
	Rat (Sprague- Dawley)	31 d Gd 0-21 Pnd 1-10 ad libitum		0.09	0.9	(reduced T3 in pups on PND 5)			York et al. 2004 NH4ClO4
		(W)							
	Gn Pig (NS)	Gd 21-48 ad libitum					531	(increased weight of fetal thyroid and hyperplasia in	Postel 1957 KClO4
		(W)						fetal thyroid)	
	Rabbit (NS)	Gd 1-28 ad libitum					72	(significantly enlarged fetal thyroid and histological	Lampe et al. 1967 KClO4
		(F)						changes in fetal thyroid)	
	Rabbit (New Zealand)	22 d Gd 6-28 ad libitum		85					York et al. 2001b NH4ClO4
		(W)							
	Cancer								
	Rat (Wistar)	12 mo ad libitum					928	(CEL: thyroid follicular adenoma)	Florencio Vicente KCIO4
		(W)							
	Mouse (NMRI)	160 d ad libitum					1020 F	(CEL: thyroid adenoma)	Gauss 1972 KClO4
		(F)							
	Mouse (BALB/c)	46 wk ad libitum					2573 F	(CEL: thyroid follicular cell carcinoma in 5/6)	Pajer and Kalisnik 1991

Table 3-2. Levels of Significant Exposure to Perchlorates - Oral (continued)

	_	Exposure/ duration/	System	NOAEL (mg/kg/day)	LOAEL				<u> </u>
Cey to		frequency (Specific route)			Less serious (mg/kg/day)		Serious (mg/kg/day)		Reference Chemical Form
	CHRONIC	EXPOSURE							
	Systemic								
51	Rat (Wistar)	24 mo ad libitum	Endocr				956 M	(thyroid fibrosis)	Kessler and Kruskemper 1966
		(W)	Bd Wt	956 M					KCIO4
52	Rat (Wistar)	15 mo ad libitum	Endocr				928	(thyroid hypertrophy and hyperplasia)	Toro Guillen 1991 KClO4
		(W)	Bd Wt		928	(unspecified decreased weight gain)			
	Cancer								
53	Rat (Wistar)	24 mo ad libitum					956 M	(CEL: incr papillary and/or follicular adenomas in	Kessler and Kruskemper 1966
		(W)						thyroid)	KCIO4
	Rat (Wistar)	15 mo ad libitum (W)					928	(CEL: follicular and papillar carcinoma of thyroid)	y Toro Guillen 1991 KClO4

^{*}The number corresponds to entries in Figure 3-2.

Bd Wt = body weight; Cardio = cardiovascular; CEL = cancer effect level; d = day(s); Endocr - endocrine; (F) = food; F = female; gastro = gastrointestinal; (GW) = gavage in water; gd = gestation day; (IN) = ingestion; Hemato = hematological; LOAEL = lowest-observable-adverse-effect level; M = male; Metab = metabolic; Musc/skel = musculoskeletal; NOAEL = no-observable-adverse-effect level; Resp = respiratory; (W) = water; wk = week(s); x = times; yr = year(s)

^b ATSDR has adopted the NAS chronic RfD of 0.0007 mg/kg/day for the chronic oral MRL. The RfD was calculated by dividing the NOEL of 0.007 mg/kg/day by an uncertainty factor of 10 (for the protection of sensitive populations).

HEALTH EFFECTS

PERCHLORATES

mac. Cardiovascular Castrointestinal mg/kg/day 10000 ■1m ○15r 1000 **014r** 100 ○5r **⊘**5r ○10m ○10m ○10m ○10m ○10m ○12m 10 **○7r** ○7r ○7r **○7r ○7r** ○7r O7r O7r ○7r ○7r **○7r** ○13r ○16r €6r **10m** ○6r 1 △2 △2 △2 **1**9m ∆3 ∆3 ∆з **1**5r **1**8r 0.1 **11m** 0.01 △2 0.001 LD50/LC50
Minimal Risk Level
for effects
other than Cancer Effect Level-Animals
LOAEL, More Serious-Animals
LOAEL, Less Serious-Animals
NOAEL - Animals ▼Cancer Effect Level-Humans ▲ LOAEL, More Serious-Humans ▲ LOAEL, Less Serious-Humans △ NOAEL - Humans c-Cat -Humans f-Ferret n-Mink j-Pigeon e-Gerbil d-Dog k-Monkey o-Other r-Rat m-Mousé

Figure 3-2. Levels of Significant Exposure to Perchlorates - Oral Acute (≤14 days)

Systemic

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p-Pig q-Cow

h-Rabbit

a-Sheep

s-Hamster

g-Guinea Pig

Cancer

Figure 3-2. Levels of Significant Exposure to Perchlorates - Oral (*Continued*) Intermediate (15-364 days)

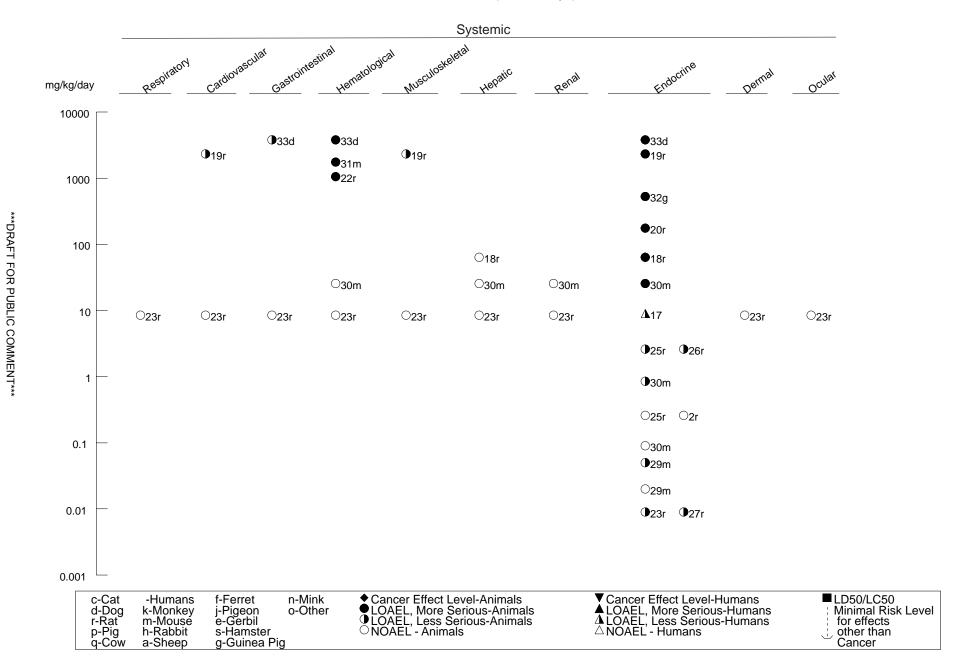


Figure 3-2. Levels of Significant Exposure to Perchlorates - Oral (*Continued*)

Intermediate (15-364 days)

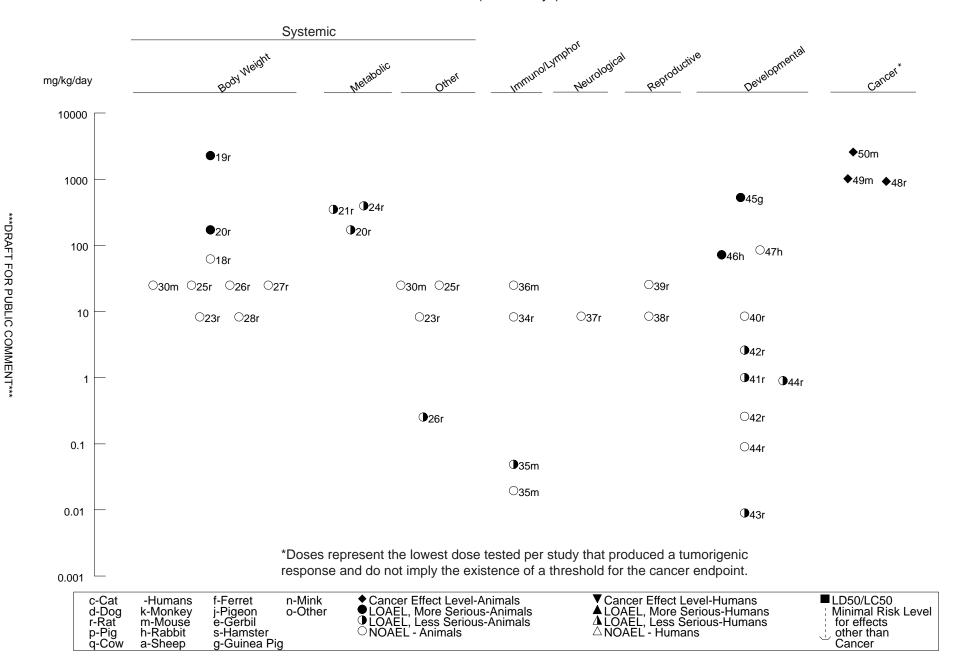
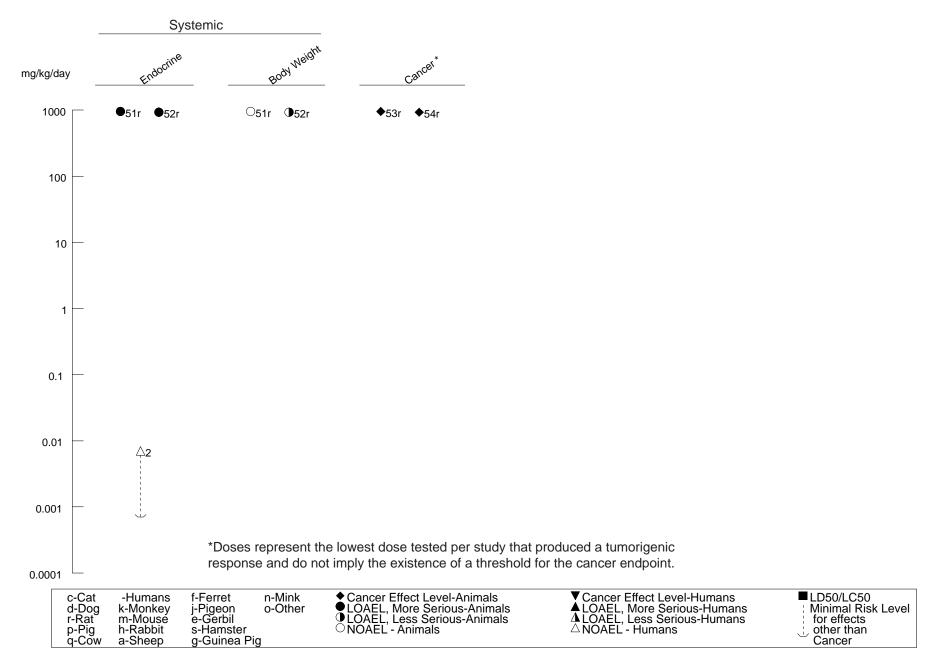


Figure 3-2. Levels of Significant Exposure to Perchlorates - Oral (*Continued*) Chronic (≥365 days)



DRAFT FOR PUBLIC COMMENT

perchlorate (Crooks and Wayne 1960; Godley and Stanbury 1954). In a review of 250 cases, the incidence of nausea was 1.5% (3/200) among patients given 600 or 1,000 mg potassium perchlorate/day (approximately 6 or 10 mg perchlorate/kg/day) and 4% (2/50) among patients given 1,500 or 2,000 mg potassium perchlorate/day (approximately 15 or 20 mg perchlorate/kg/day) (Crooks and Wayne 1960). Although gastrointestinal distress was limited to nausea in most cases, there was one case of a 22-year-old anorectic female Graves' disease patient who experienced burning epigastric discomfort and frequent vomiting within days of starting perchlorate treatment (600 mg potassium perchlorate/day or 6 mg perchlorate/kg/day), and developed a ruptured duodenal ulcer a week later (Godley and Stanbury 1954).

Irritation of the gastric mucosa was reported in dogs given 3,811 mg perchlorate/kg/day as ammonium perchlorate by gavage for 3 weeks (Selivanova and Vorobieva 1969). In rats administered up to 8.5 mg perchlorate/kg/day as ammonium perchlorate in the drinking water for up to 90 days, there was no evidence of gross or histological alterations of any section of the gastrointestinal tract (Siglin et al. 2000).

Hematological Effects. Two recent controlled acute exposure studies in euthyroid volunteers provide information of hematological effects of perchlorate in humans. No alterations in hematological parameters (complete blood count and routine chemistries) were observed in a group of nine male subjects who consumed once a day for 14 consecutive days a solution of potassium perchlorate that provided 10 mg of perchlorate/day (Lawrence et al. 2000). Blood tests were repeated on days 7 and 14 of dosing and 14 days after perchlorate was discontinued. Assuming a body weight of 70 kg, the perchlorate intake was approximately 0.14 mg/kg/day. Similar lack of hematological alterations was recently reported among a group of 37 volunteers who ingested up to 0.5 mg of perchlorate/kg/day for 14 days (Greer et al. 2002).

Hematological parameters were evaluated in an epidemiological study of school-age children from three cities with different concentrations of perchlorate in drinking water in northern Chile (Crump et al. 2000). The city with the highest perchlorate concentration was Taltal, 100–120 µg perchlorate/L (ppb), water from the city of Chañaral had 5–7 µg/L, and perchlorate was not detected in water from the city of Antofagasta. Assuming a default consumption of 1–2 L of water/day and a measured body weight of approximately 25 kg, the children in Taltal may have consumed up to 0.004–0.008 mg perchlorate/kg/day. The study comprised 162 children 6–8 years of age of which 127 had resided continuously in their respective city since conception. There was nearly an equal number of boys and girls. Analysis of complete blood counts showed no significant differences between the three groups of children whether the analysis included all of the children or only the lifelong residents.

Severe hematological effects were found in several cases of hyperthyroid patients treated with potassium perchlorate. Some patients developed aplastic anemia, characterized by drastic reductions in circulating granulocytes, erythrocytes, and thrombocytes, and a lack of erythropoietic and granulopoietic cells in the bone marrow (Barzilai and Sheinfeld 1966; Fawcett and Clarke 1961; Gjemdal 1963; Hobson 1961; Johnson and Moore 1961; Krevans et al. 1962). Aplastic anemia was the cause of death in most of the documented fatalities associated with potassium perchlorate treatment of thyrotoxicosis. In other patients, the decrease in formed blood elements was limited to the granulocytes (agranulocytosis) and/or thrombocytes (thrombocytopenia). Agranulocytosis was fatal in at least one case (Barzilai and Sheinfeld 1966), although other patients survived this condition (Barzilai and Sheinfeld 1966, Crooks and Wayne 1960; Southwell and Randall 1960; Sunar 1963). The doses in patients who developed agranulocytosis and aplastic anemia were mostly in the low-to-moderate range of doses employed in thyrotoxicosis therapy: 600–1,000 mg potassium perchlorate/day, or roughly 5–12 mg perchlorate/kg/day. Cases of agranulocytosis were found within 14 days to 3 months of the start of potassium perchlorate treatment. Although aplastic anemia was found after 2 months of treatment in one case, in most cases, it was only found after 4-8 months. All of the documented cases of aplastic anemia and agranulocytosis were females (Graves' disease, the most common cause of hyperthyroidism, is far more common in women than in men), with ages ranging from 24 to 82 years.

Inhibition of hematopoiesis in the bone marrow has also been reported in dogs given 3,811 mg perchlorate/kg/day as ammonium perchlorate by gavage for 3 weeks (Selivanova and Vorobieva 1969), and in rats and mice exposed to 1% potassium perchlorate in the drinking water for 3 months (approximate doses of 1,059 and 1,750 mg perchlorate/day, respectively) (Shevtsova et al. 1994). No significant alterations in hematological parameters were reported following administration of ammonium perchlorate in a drinking water study in mice at doses up to 25.5 mg perchlorate/kg/day for 14 or 90 days (DoD 1999). Similarly, a recent study in rats found no evidence of hematotoxicity after administration of up to 8.5 mg perchlorate/kg/day in the drinking water for 90 days (Siglin et al. 2000). The investigators evaluated routine hematology and clinical chemistry parameters.

Musculoskeletal Effects. No studies were located regarding musculoskeletal effects in humans after oral exposure to perchlorate.

MacDermott (1992) observed a decrease in membrane potential and in intracellular potassium ion activity in skeletal muscle from rats treated with 2% potassium perchlorate (approximately 2,327 mg

perchlorate/kg/day) in the drinking water for 6 weeks. The observed changes are consistent with a decrease in the number of sodium-potassium pump units in the muscle. No alterations in gross or microscopic appearance of skeletal muscle were reported in rats exposed to doses up to 8.5 mg perchlorate/kg/day as ammonium perchlorate in the drinking water for 90 days (Siglin et al. 2000).

Hepatic Effects. No evidence of liver toxicity, as judged by blood chemistry tests, was observed in a group of nine volunteers who ingested approximately 0.14 mg of perchlorate/kg/day as potassium perchlorate for 14 consecutive days (Lawrence et al. 2000). Similar results were reported by Greer et al. (2002) in a study of 37 volunteers who consumed up to 0.5 mg of perchlorate/kg/day also for 14 days. In the study by Crump et al. (2000) of 162 school-age children from three cities in northern Chile with different perchlorate concentration in the drinking water (up to 100–120 μg/L), there were no indications of altered liver function among the children as measured by serum aspartate aminotransferase (AST), alkaline phosphatase (AP), and lactate dehydrogenase (LDH) activities.

Godley and Stanbury (1954) reported no evidence of liver toxicity in a series of 24 hyperthyroid patients treated with potassium perchlorate (600 mg, or approximately 6 mg perchlorate/kg/day) for up to 52 weeks. However, it is not clear what tests were conducted to monitor effects on the liver or how frequently such tests may have been conducted.

A 0.1% concentration of potassium perchlorate in the diet (about 64 mg perchlorate/day) for 19 weeks had no effect on liver weight in rats (Hiasa et al. 1987). A more recent study in rats found that administration of ammonium perchlorate in the drinking water at doses up to 8.5 mg perchlorate/kg/day for up to 90 days caused no significant alterations in liver weight, in the gross or microscopic appearance of the liver, or in serum transaminase activities (Siglin et al. 2000). No effects on liver weight were reported in 14- and 90-day studies in mice administered up to 25.5 mg perchlorate/kg/day in the drinking water as the ammonium salt (DoD 1999).

Renal Effects. Limited information exists regarding renal effects of perchlorate in humans. Two studies in euthyroid volunteers who ingested up to 0.5 mg of perchlorate/kg/day as the potassium salt for 14 days found no evidence of renal effects as judged by standard clinical chemistry tests (Greer et al. 2002; Lawrence et al. 2000). Also, no alterations in BUN or in serum creatinine levels were observed in a group of 60 school-age children from northern Chile exposed to perchlorate in their drinking water at concentrations up to $100-120 \mu g/L$ (Crump et al. 2000).

In a case report, a patient with severe hyperthyroidism who was treated with an average of 1,068 mg sodium perchlorate/day (approximately 12 mg perchlorate/kg/day) for 3.5 months developed nephrotic syndrome, as diagnosed by albuminuria, decreased serum albumin, and increased serum cholesterol. The effects subsided after treatment was stopped, and were considered by the researchers to probably have been treatment-related (Weber and Wolf 1969).

There is also limited information on the renal effects of perchlorate in animals. In 14- and 90-day drinking water studies in rats, doses of up to 8.5 mg/kg/day produced no significant alterations in kidney weight or in gross or microscopical appearance of the kidneys (Siglin et al. 2000). In addition, kidney function, monitored by measurements of BUN and serum creatinine, was not affected by exposure to perchlorate (Siglin et al. 2000). A similar study in mice also found no effects of ammonium perchlorate on kidney weight following 14 or 90 days of exposure to up to 25.5 mg perchlorate/kg/day, but kidney function tests were not performed (DoD 1999).

Endocrine Effects. The findings of groundwater contamination with perchlorate in western areas of the United States has triggered considerable research on the effects of this anion on the thyroid gland, its main target organ, in efforts to describe dose-response relationships at low doses and to define no-effectlevel of exposure. For example, Lawrence et al. (2000) evaluated serum TSH, free thyroxine index (FTI), total serum triiodothyronine (TT3), and radioactive iodine uptake (RAIU); serum and 24-hour urine perchlorate; and 24-hour urinary iodide excretion in volunteers who ingested approximately 0.14 mg perchlorate/kg/day in drinking water for 14 days. Tests were conducted pre-dosing, on day 7 and 14, and 14 days after perchlorate ingestion was discontinued. The only significant finding was a significant decrease in 4-, 8-, and 24-hour RAIU values by a mean of about 38% relative to baseline on day 14 of dosing. Fourteen days later, RAIU had recovered to a mean of 25% above baseline values. Greer et al. (2002) conducted a similar study in volunteers administered 0.007, 0.02, 0.1, or 0.5 mg perchlorate/kg/day in drinking water for 14 days. RAIU was measured on exposure days 2 and 14, and 15 days after dosing ceased. As a percentage of baseline RAIU, inhibition in the 0.007, 0.02, 0.1, and 0.5 mg/kg/day dose groups was 1.8, 16.4, 44.7, and 67.1%, respectively. There were no significant differences between the RAIU values measured on day 2 and 14. Fifteen days after perchlorate treatment was discontinued, RAIU values were slightly higher than baseline values. Greer et al. (2002) also found no significant effects of perchlorate treatment on serum TSH, free T4, TT4, and TT3, and on serum antithyroid peroxidase levels; serum antiglobulin levels were below detection levels in all samples tested. The National Academy of Sciences (NAS 2005) derived a chronic RfD of 0.0007 mg/kg/day for

perchlorate based of the findings of Greer et al. (2002). ATSDR has adopted the NAS RfD for the chronic oral MRL.

Other earlier studies in healthy human subjects also showed that perchlorate administered in doses between 7 and 10 mg/kg/day reduced thyroid iodide uptake, increased serum iodide levels, and increased urinary iodide excretion (Brabant et al. 1992; Bürgi et al. 1974; DeGroot and Buhler 1971; Faure and Dussault 1975).

Epidemiological studies evaluating adults, children, and newborns have also been conducted (studies of children and newborns are summarized in Section 3.2.2.6, Developmental Effects). In a study of the general population, Li et al. (2001) examined the prevalence of thyroid diseases in Nevada Counties with respect to perchlorate in drinking water. The cohort consisted of all users of the Nevada Medicaid program during the period of January 1, 1997 to December 31, 1998. Disease prevalence in residents from Clark County (Las Vegas), whose drinking water had 4–24 µg/L of perchlorate (0.0001–0.0007 mg perchlorate/kg/day), were compared with those from another urban area of similar size (Reno, Washoe County), but with no perchlorate in the water, and also with those from all other counties, also with no perchlorate exposure. Patients were defined as those having one or more of the following diagnoses of thyroid disease: simple and unspecified goiter, nontoxic nodular goiter, thyrotoxicosis with or without goiter, congenital hypothyroidism, acquired hypothyroidism, thyroiditis, other disorders of the thyroid, or malignant neoplasm of the thyroid gland. Analysis of the data showed no statistically significant periodprevalence rate difference between Clark County and Washoe County. For acquired hypothyroidism, the prevalence was lower in Clark County than in other counties (opposite to what would be expected). Li et al. (2001) acknowledged that their analysis was a crude analysis since age- and sex-adjusted prevalence could not be calculated because of lack of information on age and sex distributions of the Medicaideligible population in each county.

A recent study of pregnant women from three cities (Antofagasta, Chañaral, and Taltal) in northern Chile found no significant association between levels of perchlorate in the drinking water and serum levels of TSH, T4, or thyroglobulin measured early (16.1 weeks) or late (32.4 weeks) during pregnancy (Téllez et al. 2005). The mean concentrations of perchlorate in the drinking water from Chañaral and Taltal were 5.8 and 113.9 µg/L, respectively; drinking water from Antofagasta had <0.4 µg/L of perchlorate.

Stanbury and Wyngaarden (1952) found that a single oral dose of 100 mg of potassium perchlorate (approximately 1 mg perchlorate/kg) dramatically reduced uptake of iodide by the thyroid gland in

Graves' disease patients. Subsequent to this finding, potassium perchlorate became an accepted treatment for hyperthyroidism, and was widely used for this purpose for several years (Connell 1981; Crooks and Wayne 1960; Godley and Stanbury 1954; Morgans and Trotter 1960). The use of perchlorate for the treatment of hyperthyroidism came to a virtual stop due to the appearance of cases of aplastic anemia (see Hematological Effects).

Studies in laboratory animals have described the thyroid effects of perchlorate in great detail. Reported findings have included reduced thyroid iodide uptake, increased levels of iodide in serum, decreased serum T4 and T3, increased serum T5H, increased thyroid size and weight, and hypertrophy and hyperplasia of thyroid cells, eventually leading to fibrosis and tumor development (see Cancer section), (Fernandez Rodriguez et al. 1991; Florencio Vicente 1990; Gauss 1972; Hartmann et al. 1971; Hiasa et al. 1987; Kapitola et al. 1971; Kessler and Kruskemper 1966; Logonder-Mlinsek et al. 1985; MacDermott 1992; Mannisto et al. 1979; Matsuzaki and Suzuki 1981; Ortiz-Caro et al. 1983; Pajer and Kalisnik 1991; Postel 1957; Schonbaum et al. 1965; Selivanova and Vorobieva 1969; Spreca and Musy 1974; Tarin-Remohi and Jolin 1972; Toro Guillen 1991; Wyngaarden et al. 1952). In general, many studies conducted in the early 1990s and before used relatively high doses of perchlorate, and/or only one dose level was tested, thus precluding establishing dose-response relationships that defined no-effect dose levels. Perchlorate doses reported to produce the effects mentioned above ranged from 7 to 3,811 mg/kg/day after durations ranging from 1 day to 2 years.

Studies conducted within the past 5 years in adult nonpregnant animals have used much lower doses of perchlorate. For example, Caldwell et al. (1995) conducted a pilot 14-day drinking water study in rats. The animals were exposed to one of seven doses of perchlorate ranging from 0.1 to 39.9 mg perchlorate/kg/day. Perchlorate administration induced dose-related increases in TSH and decreases in T4 and T3 in both males and females, but females appeared to be more sensitive than males. The lowest administered dose, 0.1 mg/kg/day, increased TSH and decreased T4 and T3 in females roughly by 15, 12, and 34%, respectively, relative to controls. An additional 14-day study in rats reported a significant increase in serum TSH and a nonsignificant decrease in T3 at perchlorate doses 0.09 mg/kg/day, the lowest level tested (Yu et al. 2002). A more comprehensive 14-day study in rats was conducted by Siglin et al. (2000). Perchlorate was administered in the drinking water as the ammonium salt in doses of 0, 0.009, 0.04, 0.17, 0.85, or 8.5 mg perchlorate/kg/day. At the end of the exposure period, blood TSH was significantly increased in males at ≥0.17 mg/kg/day (23%) and in females at ≥0.04 mg/kg/day (17%). Blood T4 showed a decreasing trend with increasing perchlorate doses, the differences relative to controls achieved statistical significance in both males (23% decrease) and females (18% decrease) only at the

highest dose level. Blood T3 was significantly decreased (dose-related) in all male groups (21% at the lowest dose), but was not significantly affected in any female group. Both absolute and relative thyroid weights were significantly increased in males from the highest dose group, no significant effects were seen in females. Histological alterations in the thyroid were observed only at the high dose ranging in severity classified as minimal, mild, or moderate. Minimal or mild lesions were seen in 7/10 high dose females and 3/10 high dose males. Moderate lesions were seen in 7/10 males at 8.5 mg/kg/day and consisted of follicular cell hypertrophy with microfollicle formation and colloid depletion. There was no evidence of focal hyperplasia.

In a 14-day study in mice exposed to 0, 0.09, 0.85, 2.6, or 25.5 mg perchlorate/kg/day, serum T4 was significantly decreased at 2.6 and 25.5 mg/kg/day (14 and 22%, respectively) (DoD 1999). T3 was lower than controls, although not significantly, in all treated groups except the 0.85 mg/kg/day group. There was no clear pattern of change in TSH levels. Morphological evaluation of the thyroid showed colloid depletion, intrafollicular capillary congestion, and mildly hypertrophied follicular epithelium in mice from the highest dose group. An additional 14-day study in mice reported a significant decrease in serum T4 levels at \geq 0.2 mg perchlorate/kg/day and a significant increase in TSH at \geq 1.7 mg/kg/day; serum T3 was not measured (BRT 2000). Microscopical examination of the thyroid revealed colloid depletion and hypertrophy in 5 out of 5 mice dosed with 42.5 mg/kg/day, but no significant alterations at the next lower dose level, 1.7 mg/kg/day.

A 90-day study was conducted in rats exposed to 0, 0.009, 0.04, 0.17, 0.85, or 8.5 mg perchlorate/kg/day in the drinking water (Siglin et al. 2000). Following the exposure period, the rats were provided uncontaminated drinking water for an additional 30-day period. After the 90 days of exposure to perchlorate, relative to controls TSH was significantly increased in males at ≥0.17 mg/kg/day (17% increase) and in females at 8.5 mg/kg/day (21% increase). Blood T4 was significantly decreased in both males and females from all treated groups (dose-related) (decreases ranged from 14 to 43% in males). The effect of perchlorate on blood T3 was similar to that on T4 (12–35% decrease in males). At 120 days, hormone levels approached control levels except for T4 in males and TSH in females. Both absolute and relative thyroid weights were significantly increased in males and females at 8.5 mg/kg/day at 90 days but returned to near control values at 120 days. Histological alterations in the thyroid ranged in severity from minimal to mild and were seen only at the 8.5 mg/kg/day dose level in both male and female rats. The lesions consisted of follicular cell hypertrophy with microfollicle formation and colloid depletion. There was no evidence of focal hyperplasia. No abnormal pathology was seen in the thyroid after 120 days. In a 2-generation reproductive study in rats, the F1 generation was exposed directly to

perchlorate (0.26, 2.6, or 25.5 mg/kg/day) from weaning to 19 weeks of age, at which time, the animals were killed (York et al. 2001a). In these adult rats, a significant increase in absolute and relative thyroid weight was seen in males at 2.6 and 25.5 mg/kg/day and in all female groups (dose-related). Hypertrophy and hyperplasia of the thyroid also occurred at 2.6 and 25.5 mg/kg/day in males and in high-dose females. TSH increased only in high-dose males and females and T4 decreased in high-dose males (26% decrease); T3 levels were not significantly affected. Hypertrophy and hyperplasia of the thyroid was reported at ≥2.6 mg perchlorate/kg/day in the paternal generation of rats in the 2-generation study mentioned above in which the rats were exposed for a period that included premating, pregnancy, and lactation (York et al. 2001a); the NOAEL was 0.26 mg/kg/day. The highest dose tested, 25.5 mg/kg/day, induced a significant increase in TSH and a decrease in T4 in males.

In a developmental study in rats in which dosing with ammonium perchlorate at doses of 0, 0.009, 0.09, 0.85, and 25.5 mg perchlorate/kg/day began 14 days premating and continued to gestation day 21, TSH and T4 were significantly increased and decreased, respectively, in a dose-related manner in all dosed groups of dams (York et al. 2003). In an additional developmental study in rats in which exposure started 14 days before mating and continued until postnatal day 10, treatment with up to 8.5 mg perchlorate/kg/day caused no maternal toxicity as judged by clinical observations, body and thyroid weights, and thyroid histology (York et al. 2004).

BRT (2000) evaluated serum TSH and T4 levels and thyroid histology in mice in a 90-day study. The exposure levels were 0, 0.02, 0.05, 0.2, 1.7, or 42.5 mg perchlorate/kg/day. Treatment with ammonium perchlorate decreased serum T4 levels, and the magnitude of the difference relative to controls achieved statistical significance at the 1.7 mg/kg/day dose level (18% decrease). The decrease in T4 was dose-related at ≥0.2 mg/kg/day and higher. Serum TSH was significantly elevated at ≥0.05 mg/kg/day relative to controls (17% increase at the 0.05 mg/kg/day dose level). Microscopical examination of the thyroid revealed hypertrophy in 3 out of 15 mice at 1.7 mg/kg/day, and in 4 out of 5 high-dose mice. Colloid depletion was present in 5 out of 5 mice dosed with 42.5 mg/kg/day. No significant treatment-related differences were observed between the other groups and controls.

In a developmental study in New Zealand rabbits, exposure to up to 85 mg perchlorate/kg/day on gestation days 6–28 did not significantly alter absolute or relative thyroid weight (York et al. 2001b). However, hypertrophy of the follicular epithelium was seen in the does at \geq 8.5 mg/kg/day, and the incidence was dose-related. Neither serum TSH nor T3 levels were significantly affected by treatment

with perchlorate. Serum T4 was significantly reduced at 25.5 and 85 mg/kg/day; T4 was also reduced at 0.85 and 8.5 mg/kg/day, but not significantly.

Other endocrine effects reported in perchlorate-treated animals included pituitary hypertrophy and hyperplasia (Pajer and Kalisnik 1991), reduced serum growth hormone levels (Ortiz-Caro et al. 1983), and reduced serum insulin (Tarin-Remohi and Jolin 1972). All of these effects were accompanied by thyroid effects in the same studies.

Dermal Effects. No reports were found of adverse dermal effects of perchlorate in healthy humans. Skin rash was the most frequent side effect of potassium perchlorate therapy in thyrotoxicosis patients, occurring primarily in patients receiving doses at the high end of the therapeutic range. Rash was observed in 10% (5/50) of patients treated with 1,500 or 2,000 mg (approximately 15 or 20 mg perchlorate/kg/day) by Crooks and Wayne (1960), and in 15% (10/67) of patients treated with 1,200 or 1,600 mg (approximately 12 or 16 mg perchlorate/kg/day) by Morgans and Trotter (1960). However, rash was seen in only 0.5% (1/200) patients treated with 600 or 1,000 mg (approximately 6 or 10 mg perchlorate/kg/day) by Crooks and Wayne (1960), and in none of the 24 patients treated with 600 mg (approximately 6 mg perchlorate/kg/day) by Godley and Stanbury (1954). The observed rash was characterized as maculopapular by Crooks and Wayne (1960), and was attributed by these authors to a hypersensitivity reaction. Hemorrhagic skin lesions were frequently noted in cases with severe hematological effects (Barzilai and Sheinfeld 1966; Fawcett and Clarke 1961; Gjemdal 1963; Hobson 1961; Johnson and Moore 1961; Krevans et al. 1962; Southwell and Randall 1960). The lesions, which were variously described as punctate erythema, hemorrhagic pustulae, purpuric rash, skin hemorrhage, bleeding into the skin, and petecchiae, apparently occurred secondary to the hematological effects.

In rats administered up to 8.5 mg perchlorate/kg/day as ammonium perchlorate in the drinking water for 90 days, no significant gross or microscopical alterations in the skin were found throughout the study (Siglin et al. 2000).

Ocular Effects. No studies were located regarding ocular effects in humans after oral exposure to perchlorate. Opthalmological examinations on rats dosed with up to 8.5 mg of perchlorate/kg/day for up to 90 days revealed no treatment-related effects (Siglin et al. 2000).

Body Weight Effects. No studies were located regarding body weight effects in humans after oral exposure to perchlorate.

In acute-duration rat studies, reduced growth was reported at an estimated dose of 1,830 mg perchlorate/kg/day (Arieli and Chinet 1985), but not at doses of 1,500 mg perchlorate/kg/day or below (Caldwell et al. 1995; Kapitola et al. 1971; Matsuzaki and Suzuki 1981; Schonbaum et al. 1965; Siglin et al. 2000). In longer-term studies, there are reports of reduced body weight gain in rats at doses of 175 mg perchlorate/kg/day for 25 days (Ortiz-Caro et al. 1983), 1,362 mg/kg/day for 18 days (Tarin-Remohi and Jolin 1972), 928 mg/kg/day for 12 months (Florencio Vicente 1990), 2,327 mg/kg/day for 6 weeks (MacDermott 1992), and 928 mg/kg/day for 15 months (Toro Guillen 1991). Treatment of rats for 90 days with up to 8.5 mg of perchlorate/kg/day in the drinking water did not result in significant effects on growth (Siglin et al. 2000), nor did treatment with 64 mg/kg/day for 19 weeks (Hiasa et al. 1987). Also, in a 2-generation reproduction study in rats, no significant effects on body weight were seen in F1 animals treated directly with up to 25.5 mg perchlorate/kg/day from weaning to 19 weeks of age in addition to being exposed perinatally (York et al. 2001a); no significant effects on body weight were seen in the paternal generation also in that study. A study in mice also found no alterations in body weight or weight gain following 14 or 90 days of exposure to ammonium perchlorate in the drinking water in doses up to 25.5 mg perchlorate/kg/day (DoD 1999). Where present, reduced growth is considered secondary to hypothyroidism produced by perchlorate.

Metabolic Effects. No studies were located regarding metabolic effects in humans after oral exposure to perchlorate.

Researchers in India conducted a number of studies investigating the metabolic effects of perchlorate in rats given 500 mg/kg/day of potassium, sodium, or ammonium perchlorate by daily gavage for 45 days (Sangan and Motlag 1986, 1987; Vijayalakshmi and Motlag 1989a, 1989b, 1990, 1992). They found that perchlorate increased protein metabolism (increased liver arginase activity and serum urea levels), altered carbohydrate metabolism (decreased serum glucose and increased liver and kidney glycogen levels, reflecting increased activity of aldolase, lactate dehydrogenase, and glycogen synthase, and decreased activity of glucose-6-phosphatase and glycogen phosphorylase), and modified lipid metabolism (increased cholesterol, triglyceride, and phospholipid, and decreased free fatty acid levels, reflecting decreased activity of lipase and phospholipase). They also found that perchlorate reduced the activities of mitochodrial enzymes involved in cellular respiration, apparently due to changes in lipid composition of mitochodrial membranes (increased cholesterol and decreased phospholipid) reducing membrane fluidity.

Other studies reported only a small (11%), nonsignificant decrease in serum glucose levels in rats exposed to potassium perchlorate (1,362 mg perchlorate/kg/day) in the drinking water for 18 days (Tarin-Remohi and Jolin 1972), and no effect on serum glucose in rats exposed to 0.1% potassium perchlorate (approximately 175 mg perchlorate/kg/day) in the drinking water for 25 days (Ortiz-Caro et al. 1983). No effects were observed on serum glucose levels in rats exposed to up to 8.5 mg of perchlorate/kg/day as ammonium perchlorate in the drinking water for up to 90 days (Siglin et al. 2000).

Ortiz-Caro et al. (1983) observed a significant decrease in the activity of α -glycerophosphate dehydrogenase (α -GPD) in hepatic mitochondria in their study that was considered secondary to hypothyroidism produced by perchlorate. However, Arieli and Chinet (1985) found no effect on cytoplasmic α -GPD in brown fat in rats that received 2% potassium perchlorate (1,830 mg perchlorate/kg/day) in the drinking water for 2 weeks.

Other Systemic Effects. No studies were located regarding other systemic effects in humans after oral exposure to perchlorate.

Eskin et al. (1975) observed reduced iodide uptake, decreased weight, and dysplastic histopathological lesions in the mammary gland of rats treated with 459 mg perchlorate/kg/day as sodium perchlorate in the drinking water for 8 weeks. Mammary gland dysplasia was also seen in ovariectomized rats given estrogen replacement and then dosed with 494 mg perchlorate/kg/day for 8 weeks (Eskin et al. 1976).

Water consumption was not significantly altered in rats administered up to 40 mg of perchlorate/kg/day for 14 days in the drinking water (Caldwell et al. 1995). Neither food or water consumption were affected in rats exposed to perchlorate via drinking water in doses up to 8.5 mg/kg/day for up to 90 days (Siglin et al. 2000). In a 2-generation reproduction study in rats, paternal males exposed for 16 weeks showed a reduction in absolute and relative water consumption at 0.26 and 25.5 mg perchlorate/kg/day, but no at 2.6 mg/kg/day (York et al. 2001a). In that same study, no significant effects were seen on water consumption in the F1 generation exposed directly to up to 25.5 mg perchlorate/kg/day in the drinking water from weaning to 19 weeks of age. No significant effects on water consumption were also reported in 14- and 90-day studies in mice given ammonium perchlorate in the drinking water in doses of up to 25.5 mg perchlorate/kg/day (DoD 1999).

3.2.2.3 Immunological and Lymphoreticular Effects

No reports were found of perchlorate-induced alterations in immune system parameters in healthy humans. Two cases of lymphadenopathy (not further described) were reported among a series of 247 hyperthyroid patients treated with potassium perchlorate (Morgans and Trotter 1960). Both cases occurred in patients treated with 1,200 or 1,600 mg potassium perchlorate/day (roughly 12 or 16 mg perchlorate/kg/day). Lymphoreticular effects were not reported in other case studies.

Spreca and Musy (1974) found increases in the proportion of degranulated mast cells in the thyroid, skin, liver, and lungs of rats treated with potassium perchlorate (approximately 323 mg perchlorate/kg/day) for 1 day. The effect was greatest in the thyroid (27% decrease) and skin (21% decrease). Degranulation of mast cells is typically associated with exposure to an allergen; degranulation releases pharmacological mediators of immediate hypersensitivity responses (histamine, heparin, etc.), leading to allergy symptoms. Clinical signs of hypersensitivity response were not monitored in this study. There was also an increase in the number of mast cells in the thyroid and small decreases in mast cell numbers in the skin, liver, and lung. The researchers suggested that the increase in the thyroid was associated with hyperplasia in this tissue, and that the decrease in the other tissues may reflect loss of cells by degranulation. An increase in mast cell numbers in the thyroid of mice treated with sodium perchlorate (1.2% in drinking water, or roughly 2,622 mg perchlorate/kg/day) for 64 days was also reported by Logonder-Mlinsek et al. (1985). The extent of mast cell degranulation was not reported in this study.

More recent studies in animals have tested much lower doses of perchlorate and conducted a more complete evaluation of the immune system. For example, 14- and 90-day studies in rats administered ammonium perchlorate in the drinking water in doses up to 8.5 mg perchlorate/kg/day reported no significant effects on spleen weight and no gross or microscopic alterations in lymph nodes, spleen, and thymus; no tests of immunocompetence were conducted in these studies (Siglin et al. 2000). DoD (1999) evaluated a series of immunological end points in 14- and 90-day studies in mice exposed to ammonium perchlorate in the drinking water in doses of 0, 0.09, 0.9, 2.6, and 25.5 mg perchlorate/kg/day. End points evaluated included thymus and spleen weight and cellularity, CD4/CD8 splenocyte and thymocyte subpopulations, stem cell progenitors (90-day), melanoma tumor incidence (90-day), natural killer (NK) cell activity, delayed-type hypersensitivity (DTH), cytotoxic T cell activity, response to challenge with *Listeria monocytogenes* (90-day), peritoneal macrophage phagocytosis and nitrite production, and specific IgM and IgG response to cell dependent sheep red blood cell (SRBC) challenge. Significant findings in the 14-day study included an increase in the percent of CD4-/CD8+ thymic lymphocytes at 0.09 and

0.9 mg/kg/day, decreased macrophage phagocytosis at 0.9 and 25.5 mg/kg/day, and increased DTH response at 25.5 mg/kg/day. In the 90-day study, NK cell activity was increased in the highest-dose group, macrophage phagocytosis was decreased in all treated groups, the DTH response was also increased at 25.5 mg/kg/day, and increased resistence to the challenge with *Listeria* in the high-dose group when challenged only with high immunization levels. Overall, because only a few immunological parameters were affected and resistance to the challenge with *Listeria* was not decreased, the results of this study do not suggest an immunosuppressive function for perchlorate at the doses tested.

Additional 14- and 90-day drinking water studies exposed mice to 0, 0.02, 0.05, 0.2, 1.7, or 42.5 mg perchlorate/kg/day examined the plaque forming cell (PFC) response following sheep red blood cells (SRBC) immunization and the ability of mice to generate a hypersensitivity response (local lymph node assay [LLNA]) to 2,4-dinitrochlorobenzene (DNCB), a known sensitizing chemical (BRT 2000). No significant effects were seen on the PFC response after 14 days of treatment, but an increased response was seen after 90 days in the 1.7 and 42.5 mg/kg/day dose groups when the results were expressed as number of response per spleen and only at 42.5 mg/kg/day when the responses were expressed per number of spleen cells. In the LLNA assay, perchlorate increased the sensitizing potential of DNCB at all doses except 1.7 mg/kg/day in the 14-day experiment, whereas in the 90-day experiment, perchlorate increased the sensitizing potential of DNCB at 0.05 and 0.2 mg/kg/day, had no effect at 0.02 or 1.7 mg/kg/day, and decreased it at 42.5 mg/kg/day. It should be mentioned, however, that cyclophosphamide, the positive control, did not abolish the sensitizing effect of DNCB alone, calling into question the reliability of the experiment. The physiological relevancy of the enhancement of the LLNA is unclear. Further research in this area is needed to determine whether perchlorate is a contact sensitizer.

NOAEL and LOAEL values for immune system effects from the rodent studies are shown in Table 3-2 and Figure 3-2.

3.2.2.4 Neurological Effects

No studies were located regarding neurological effects in humans after oral exposure to perchlorate and limited information is available in animals. No gross or microscopic alterations were observed in the brain from rats treated with ammonium perchlorate in drinking water in doses up to 8.5 mg perchlorate/kg/day for 14 or 90 days (Siglin et al. 2000). Brain weight was also not significantly altered by exposure to perchlorate.

Neurodevelopmental effects resulting from perinatal exposure to perchlorate are discussed in Section 3.2.2.6, Developmental Effects.

3.2.2.5 Reproductive Effects

No studies were located regarding reproductive effects in humans after oral exposure to perchlorate.

Exposure to 1% potassium perchlorate (roughly 532 mg perchlorate/kg/day) in the drinking water on days 2 through 8 of gestation had no effect on the number of live litters produced, mean litter size, or duration of pregnancy in rats (Brown-Grant 1966). Nor was there any effect on the number or weight of implantation sites in lactating pregnant female rats that received approximately 1,752 mg perchlorate/kg/day in the drinking water on days 1 through 13 of gestation (Brown-Grant and Sherwood 1971).

A more recent 14-day study in male and female rats administered ammonium perchlorate in the drinking water at doses up to 8.5 mg perchlorate/kg/day found no alterations in absolute weight of the uterus, testes, or ovaries (Siglin et al. 2000). Also, there were no gross or microscopic alterations in the testes, prostate, epididymis, uterus, ovaries, or mammary glands. Examination of these end points following 90 days of exposure to the same doses also revealed no significant effects (Siglin et al. 2000). In addition, the 90-day study showed no significant effects on sperm motility, concentration, count, or morphology.

The results of a 2-generation reproductive study in rats recently became available (York et al. 2001a). Male and female rats (P generation) were exposed to ammonium perchlorate in the drinking water at target doses of 0, 0.26, 2.6, or 25.5 mg perchlorate/kg/day for 10 weeks before mating and during pregnancy and lactation. Males were sacrificed after 13 weeks of exposure and females were sacrificed on postpartum day (PPD) 21. Offspring (F1) were dosed from weaning to 19 weeks of age. Mating of F1 generation females and males produced the F2 generation. Male and female mating and fertility parameters were not affected by perchlorate; estrous cycling (before cohabitation) was also not altered by exposure to perchlorate. No significant effects were seen on number of dams delivering litters, duration of gestation, implantations, any litter parameter, lactation index, or sex ratios. In the F1 generation, there were no effects on mating and fertility or in sperm parameters; in F1 females, there were no effects on estrous cycling, fertility, sexual maturation, or in delivery and litter observations. The NOAEL for reproductive effects of perchlorate in this study was 25.5 mg/kg/day.

The NOAEL values for reproductive effects in these studies are shown in Table 3-2 and Figure 3-2.

3.2.2.6 Developmental Effects

Several recent developmental studies of perchlorate in humans have focused on the evaluation of neonatal thyroid parameters. Lamm and Doemland (1999) examined rates of congenital hypothyroidism in seven counties of Nevada and California with perchlorate contamination in the drinking water (4–16 μ g/L [ppb]) (0.0001–0.0005 mg/kg/day). The investigators analyzed data from the neonatal screening programs of the two states for any increased incidence of congenital hypothyroidism in those counties. The rates for the California births were adjusted for Hispanic ethnicity, which was known to be a risk factor for congenital hypothyroidism. During 1996 and 1997, nearly 700,000 newborns were screened. The risk ratio in the seven counties was 1.0 (95% confidence interval [CI] 0.9–1.2) (249 cases observed/243 expected). The risk ratios for the individual counties relative to statewide expected rates ranged from 0.6 to 1.1. While the results showed no increase in rates of congenital hypothyroidism, a stronger analysis could have been performed if information on age at blood sampling and the county-specific levels of perchlorate had been available.

Kelsh et al. (2003) also found no relationship between congenital hypothyroidism and exposure to perchlorate through the drinking water in a study of newborns (n=15,348) whose mothers resided in the community of Redlands, California, during the period 1983 through 1997 and who were screened by the California Newborn Screening Program. Perchlorate was detected in the water system serving the community at a concentration of up to 9 μ g/L (mean, <1 μ g/L). Two adjacent communities with no detectable perchlorate in their water systems, San Bernardino and Riverside (n=695,967), served as comparison groups. The majority of the newborns had blood collected for TSH assay 18 hours or more after birth. Cases were defined as infants diagnosed with congenital hypothyroidism or whose TSH screening concentrations were >25 μ U/mL or sometimes >16 μ U/mL. Covariates included in the model were age at specimen collection, sex, race, ethnicity, birth weight, multiple birth status, and calendar year of birth. Analysis of the results showed an adjusted prevalence ratio for congenital hypothyroidism of 0.45 (95% CI, 0.06–1.64) and an odds ratio for elevated TSH of 1.24 (95% CI, 0.89–1.68) among all newborns screened and 0.69 (95% CI, 0.27–1.45) for newborns whose age at screening was ≥18 hours. Limitations of the study include the fact that data from a single year were used to characterize exposures over the entire 15 years of the study.

Li et al. (2000b) compared mean monthly neonatal T4 levels for days 1–4 of life for newborns from the city of Las Vegas and Reno, both in Nevada. Las Vegas has perchlorate in its drinking water, whereas Reno does not. The cohorts consisted of 17,308 newborns in Las Vegas and 5,882 newborns in Reno evaluated during the period of April 1998 through June 1999; the analysis was restricted to newborns whose birth weights were between 2,500 and 4,000 grams. Perchlorate was detected in the drinking water from Las Vegas during 7 of the 15 months of the study period at levels of 9–15 μg/L (0.0003–0.0004 mg/kg/day). Analyses were performed comparing serum T4 levels of children born during the 7 months in which perchlorate was detected in the drinking water (period A) and children born during the months in which perchlorate was not detected in the drinking water (period B). The mean T4 levels were compared in a univariate analysis both crude and stratified by time period. In a multivariate analysis, T4 was the outcome variable, city and time period were the main effect variables, and gender, birth weight, and age and time of blood collection were the covariates. There was no significant difference in mean T4 level between Las Vegas and Reno in the crude analysis or when data were stratified by time period (period A or B). Gender, birth weight, and age and time of blood collection were significant covariates.

The same group of investigators also evaluated blood TSH levels in newborns in their first month of life from Las Vegas (n=4,070) and Reno (n=133) from December 1998 to October 1999 (Li et al. 2000a). TSH levels were measured on screening samples that were below the 10th percentile of T4 daily measurements in blood samples collected throughout the state. The analysis was restricted to birth weights between 2,500 and 4,500 grams, adjusted for gender and age at screen (days 2–7 vs. 8–30). The mean TSH levels of the two cities did not differ significantly, whether crude or stratified by age or sex. Multiple linear regression analysis showed that the TSH level was significantly affected by age at which the sample was collected (higher at earlier age) and by sex (higher for males), but not by location. These findings suggested that neonatal TSH levels were not affected by living in areas where drinking water contained up to 15 µg/L of perchlorate (0.0004 mg/kg/day).

A similar study of newborn TSH levels was conducted by Brechner et al. (2000). TSH levels were compared between two cities in Arizona, Yuma and Flagstaff, representing areas of exposure and nonexposure, respectively. The study covered a 3-year period between October 1994 and December 1997. Exposure data for the study period were not available. However, measurements done by EPA in 1999 showed perchlorate at $6 \mu g/L$ (0.0002 mg/kg/day) in Yuma and nondetectable levels in Flagstaff. Since the water processing facilities had not changed, and perchlorate persists in water for a long time, Brechner et al. (2000) assumed that comparable differences in perchlorate levels existed during the study

period. The final analysis comprised 1,099 newborns from Yuma and 443 from Flagstaff. The study controlled for age at screen and Hispanic ethnicity, but not for gender, gestational age, or birth weight. The median first TSH level in Yuma was significantly higher than in Flagstaff (19.9 mU/L vs. 13.4 mU/L); this difference occurred in both non-Hispanics and Hispanics. A residual confounding by age may have persisted in the analysis due to the higher percentage of newborns screened in the first 24 hours (when TSH levels peak) in Yuma (11%) compared with Flagstaff (3.1%). Lamm (2003) reanalyzed the study and compared TSH neonatal values of Yuma and two cities near Yuma, Somerton and San Luis, which get their water from a different source than the city of Yuma. The water from Somerton and San Luis is assumed to have no perchlorate contamination. Lamm's analysis showed no significant difference in TSH values between newborns from Yuma and Somerton/San Luis, suggesting that the results of Brechner et al. (2000) reflected regional differences, possibly related to the difference in altitude (7,000 feet) between Yuma and Flagstaff.

In an additional, unpublished, study of newborns, Schwartz (2001) evaluated T4 and TSH levels for all newborns in California during 1996. All infants were screened for serum T4 and TSH levels, and the samples with a low T4 (≤9 mg/dL and the next lowest 5% of the values in each tray of samples) were further analyzed for TSH levels. Information on the concentration of perchlorate in tap water was not available for this study. Therefore, perchlorate exposure was estimated using the mother's postal zip code, concentration of perchlorate in undergroundwater sources measured between February 1997 and June 2000, water source production, water purchases and sales, and characteristics of the water distribution system. Ultimately, four categories of exposure were made: 0 (n=255,382), 1-2 µg/L (n=127,041), $3-12 \mu g/L$ (n=131,483), and $\geq 13 \mu g/L$ (n=1,945). Using default values for daily water consumption and for body weight, a concentration of 13 µg of perchlorate/L would provide doses of approximately 0.0004 mg perchlorate/kg/day. This study used an analysis of the covariance model. After controlling for age at screening, gender, single versus multiple birth, and ethnicity, a statistically significant declining trend for T4 was observed with increasing perchlorate exposure. Infants in the low, medium, and high exposure groups had 0.97, 1.12, 1.82 µg/dL lower T4 levels, respectively, than unexposed infants. Log transformed TSH values showed a significant increase trend with perchlorate exposure (0.029, 0.03, and 0.128 ln μ U/mL, in the low, medium, and high exposure groups, respectively). Although significant associations were found, Schwartz (2001) noted that 90% of the variability in the infants' hormone levels remained unexplained by perchlorate exposure, gender, multiple birth, birth weight, and blood sample age. Schwartz (2001) also noted that no adjustment was made in the study for gestational age and laboratory measurement variability, two strong predictors of T4 and TSH.

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As previously mentioned, Crump et al. (2000) conducted a study of school-age children from three cities with different concentrations of perchlorate in drinking water in northern Chile. The city with the highest perchlorate concentration was Taltal, 100–120 µg perchlorate/L (ppb), water from the city of Chañaral had 5-7 µg/L, and perchlorate was not detected in water from the city of Antofagasta. The study comprised 162 children 6-8 years of age, of which 127 had resided continuously in their respective city since conception. The children underwent examination of the thyroid gland and a blood sample was taken for analysis of TSH, T4, FTI, T3, and antiperoxidase antibody. After adjusting for sex, age, and urinary iodide excretion, the children from Taltal and Chañaral had slightly lower TSH levels than children from Antofagasta (opposite to expected), but the differences were not statistically significant. Serum T4 levels in the city with the highest perchlorate levels were significantly higher than in the city with no perchlorate (opposite to expected). Analysis of all of the children included in the study revealed a small nonsignificant increase risk of goiter in the cities with perchlorate compared with the city without perchlorate, however, there was virtually no difference in risk when only lifelong residents were analyzed. The study also found that lifelong residents of Taltal (high perchlorate) were >5 times more likely to report a family history of thyroid disease compared with lifelong residents of Antofagasta (no perchlorate). Assuming a reference daily consumption of water of 1–2 L and using a body weight for the children of approximately 25 kg (measured in the study), a concentration of perchlorate in the drinking water of 100 µg/L would provide doses of approximately 0.004–0.008 mg perchlorate/kg/day.

Crump et al. (2000) also evaluated TSH levels in neonates from the three cities in northern Chile mentioned above. A total of 9,784 neonatal records were analyzed for TSH levels, sex, and date of screening for infants born between February 1996 and January 1999. The study did not control for iodine intake, ethnicity, or birth weight. The rate of congenital hypothyroidism detected in Chile from 1992 to 1999 was 1 per 3,484 cases (based on 773,440 newborns screened). In their study, Crump et al. (2000) detected seven cases of presumptive congenital hypothyroidism corresponding to a rate of 1 per 1,270 newborns. All of these cases originated in the city with no detectable levels of perchlorate. Linear regression comparisons of TSH by city showed a statistically significant decline in TSH with increasing perchlorate concentration in the drinking water, opposite to the known effect of perchlorate. The magnitude of the differences in TSH concentrations did not seem to be clinically significant.

The results of an additional study of neonates from the same cities in northern Chile mentioned above recently became available (Téllez et al. 2005). The study comprised approximately 60 women and their newborns per city. End points evaluated included neonatal weight, length, head circumference, gestational length, and FT4, T3, thyroglobulin, and perchlorate in cord serum. The evaluation showed no

significant differences between the three cities regarding indicators of fetal development or in FT4 or TSH. T3 and thyroglobulin were significantly lower among neonates from Chañaral (low perchlorate, $5.8 \mu g/L$) than in the other two cities. T3 and thyroglobulin were not significantly different between newborns from Antofagasta (no perchlorate) and Taltal (high perchlorate, $114 \mu g/L$).

Chang et al. (2003) evaluated the potential association between exposure to perchlorate via the drinking water and the incidence of attention-deficit-hyperactivity disorder (ADHD) and autism among children less than 18 years of age who were recipients of Medicaid in Nevada. The study included subjects from Clark County, which includes Las Vegas and in which the concentration of perchlorate in the public water supply ranged from undetected to 23.8 µg/L (mean, 10.9 µg/L), as measured in 1997–2001; subjects from Washoe County, which includes Reno, with no detectable perchlorate in the water supply served as an unexposed comparison group, and the remainder of Nevada served as a rural control. No perchlorate was detected in public water supplies from the rural areas. Analysis of the data from the Nevada Medicaid program showed that the rates for ADHD and for autism in the area with perchlorate in the drinking water did not exceed the rates in the areas without perchlorate in the drinking water. Furthermore, there was no difference between the three groups regarding overall fourth-grade school performance. No control was made in the analysis for age, sex, race, or ethnicity.

Studies in laboratory animals have shown that maternal exposure to relatively high doses of perchlorate during pregnancy and/or lactation leads to reduced thyroid function. Pups of rats exposed to 1% sodium perchlorate in the drinking water (about 1,300 mg perchlorate/kg/day) throughout gestation and lactation had reduced growth, increased thyroid weight, drastically decreased serum T4 and T3 levels, and markedly increased serum TSH levels compared with controls (Golstein et al. 1988). These effects are the typical indicators of hypothyroidism in juvenile and adult rats treated with perchlorate directly. In a study in guinea pigs, near-term fetuses from dams treated with 1% potassium perchlorate (about 531 mg perchlorate/kg/day) in the drinking water during the latter half of gestation had thyroid hyperplasia and a dramatic 15-fold increase in relative thyroid weight compared with controls, while maternal thyroids were unaffected (Postel 1957). This suggests that perchlorate may have entered the fetal circulation and directly affected the fetal thyroid gland. Similar fetal effects were seen in rabbits dosed with 72 mg perchlorate/kg/day in the diet throughout gestation, but in this study, effects on the maternal thyroid, although considerably less intense than in the fetuses, were observed (Lampe et al. 1967). Rat pups exposed to perchlorate only for 10 days during lactation had body weights similar to controls, but significantly increased relative thyroid weights (Brown-Grant and Sherwood 1971). The dams in this study, which were pregnant with a new litter while nursing these pups, had received an approximate dose

of 1,752 mg perchlorate/kg/day, and showed an increase in relative thyroid weight of similar magnitude to the pups.

Several developmental studies in animals have focused on the effects of perchlorate on the thyroid and also on neurodevelopmental effects following perinatal exposure to relatively low doses of perchlorate. Information on developmental effects of perchlorate is available in the 2-generation reproduction study in rats by York et al. (2001a) previously described in Section 3.2.2.5, Reproductive Effects. Perchlorate doses were 0, 0.26, 2.6, and 25.5 mg perchlorate/kg/day, and exposure started 10 weeks before mating and continued during pregnancy and lactation. The F1 generation was dosed from weaning (21 days old) to 19 weeks of age, but some pups were sacrificed on PND 21. The second generation (F2) was sacrificed at 3 weeks of age. F1 and F2 generations were exposed in utero, via maternal milk, and through maternal water. Exposure to perchlorate had no significant effect on pup weight. High-dose F1 pups killed on PND 21 showed a significant increase in thyroid weight (males and females) and in spleen weight (females). Significant hypertrophy and hyperplasia of the thyroid was seen in high-dose males and females and in mid-dose females. Also, there was a significant reduction in serum T3 in highdose females, TSH was reduced in low- and mid-dose males, and serum T4 was increased in low-dose females. Thyroid weight from high-dose F2 female pups was significantly increased, and both male and female from the mid- and high-dose group exhibited hyperplasia and hypertrophy of the thyroid. TSH, T3, and T4 levels were not significantly altered in F2 pups, although T3 was somewhat lower in highdose females. On the basis of morphological alterations in the thyroid observed in mid- and high-dose pups, the 0.26 mg/kg/day dose level is considered a developmental NOAEL. Parental (F0) effects were restricted to the thyroid and consisted mainly in hypertrophy and hyperplasia of the thyroid in the midand high-dose groups and significantly increased serum TSH levels in high-dose males.

A subsequent study in rats by York et al. (2003) examined the developmental effects of ammonium perchlorate in doses of 0, 0.009, 0.09, 0.85, and 25.5 mg perchlorate/kg/day. Dosing began 14 days premating and continued to GD 21, at which time, all rats were sacrificed. Satellite groups of rats were treated similarly and were used for collection of blood and thyroid tissues. The rats were observed for clinical signs, abortions, premature deliveries, and deaths. Body weights and food and water consumption were also monitored. At sacrifice, gravid uterine weights were recorded, and the uterus was examined for pregnancy, number and distribution of implantations, live and dead fetuses, and early and late resorptions. In addition, the number of corpora lutea in each ovary was recorded; the placenta was also examined. The fetuses were weighed and examined for gross alterations; one-half was examined for soft tissues alterations and the other half was examined for skeletal alterations and cartilage development. There

were no maternal deaths and all clinical observations were considered unrelated to the test material. There were no significant effects on body weights, weight gains, and gravid uterine weights. There were no treatment-related effects on absolute or relative food or water consumption values. Cesarean sectioning and litter parameters were not affected by exposure to perchlorate. Evaluation of the fetuses showed that the average number of ossification sites per litter for sternal centers and for forelimb phalanges was significantly reduced in the 25.5 mg/kg/day exposure group. Examination of the satellite group of pups showed a statistically significant and dose-related decrease in T3 in all dosed groups. No developmental NOAEL is identified in this study and the 0.009 mg/kg/day dose level is a developmental LOAEL.

An additional study was conducted in rats given ammonium perchlorate via the drinking water that provided doses of 0, 0.09, 0.9, 2.6, and 8.5 mg perchlorate/kg/day (York et al. 2004). Exposure began on gestation day (GD) 1 and continued for additional 10 days postpartum (PPD10). Dams were sacrificed on postnatal day (PND) 10 or 22 (12-day recovery period). Four subsets of pups were formed: subset 1 was sacrificed on PND 12 for neurohistological examination; subset 2 was used for neurobehavioral testing (avoidance testing on PND 23–32, water maze on PND 59–70) and sacrificed on PND 90–92, at which time blood was collected for TSH, T4, and T3 determinations; subset 3 was tested for motor activity on PND 14, 18, 22, and 59, and for auditory startle habituation on PND 23 and 60 and sacrificed on PND 67–69; and subset 4 was sacrificed on PND 80–86 and used for thyroid pathology and neurohistological examination and morphology. In addition, on PND 5, litters were culled to 8 pups, blood was collected for hormone analysis and the thyroid was processed for histopathology.

Treatment with perchlorate caused no maternal toxicity as judged by clinical observations, body and thyroid weights, and thyroid histology. Perchlorate did not significantly affect gestation length, litter size, number of stillborn, gestation index, pup viability index, or pup's weight. In the pups, there were no significant changes in body weight, absolute and relative food consumption, or exposure to perchlorate did not affect the day of vaginal latency or the day of preputial separation. Microscopic examination of the thyroid from pups culled on PND 5 revealed changes restricted mainly to high-dose males consisting of hypertrophy/hyperplasia of the follicular epithelium and decrease in follicle size. TSH was elevated only in pups born to dams treated with 8.5 mg/kg/day, but T3 levels were significantly reduced at 0.9 mg/kg/day and higher doses. T4 was significantly reduced at 2.6 mg/kg/day and higher doses. In pups sacrificed on PND 12 (subset 1), a significant increase in thickness of the corpus callosum was seen in high-dose females; this also was observed in high-dose males but the difference with controls was not statistically significant. Evaluation of the next lower dose group (2.6 mg/kg/day) revealed a significant

decrease in the hippocampal gyrus size in males, increase in the anterior to posterior cerebellum size and decrease in the caudate putamen in females, but no significant difference in the corpus callosum. Evaluations of subsets 2 and 3 revealed no behavioral effects in the offspring of dams exposed up to 8.5 mg perchlorate/kg/day (passive avoidance, swimming water maze, motor activity, and auditory startle). Also in subsets 2 and 3, there were no necropsy observations that seemed perchlorate-related, and terminal body weights and absolute and relative thyroid weights were comparable among the groups. In subset 4, there were no necropsy observations related to treatment, no significant effect on final body weight or thyroid weight, and no treatment-related neuropathological changes in the brain. However, morphometry evaluation of eight specific brain areas revealed a significant increase in mean thickness of the frontal cortex, caudate putamen, and corpus callosum from high-dose males. Based on the thyroid effects on pups culled on PND 5, the dose of 0.09 mg/kg/day can be considered a developmental NOAEL. The highest dose tested, 8.5 mg/kg/day is a maternal NOAEL. It should be mentioned that questions and concerns have been raised regarding the brain morphology findings. Issues that were raised by NAS (2005) include: "(1) apparent systematic differences in the plane of section among treatment groups, (2) lack of a clear and consistent dose-response relationship, (3) doubts about the biological plausibility of the changes that were observed, and (4) concerns that the measures that were used were relatively insensitive and would be unlikely to pick up subtle differences in neurodevelopment."

In a study of similar design, exposure began 2 weeks before mating and was terminated on PND 10 (Bekkedal et al. 2000, 2004). On PND 5, all of the pups were weighed and the litter culled to four males and four females. Tests of motor activity were conducted on one male and one female selected randomly on PND 14, 18, and 22. Nine measures of motor activity were monitored: frequency and time of ambulatory movements, frequency and time of stereotypic movements, frequency of movements in the horizontal plane, distance traveled in the horizontal plane, frequency of rears, total number of horizontal movements made while in rearing position, and time spent resting. Each measure of activity was recorded for 90 consecutive minutes on each test day. Data were divided into nine 10-minute blocks. The results showed that the main effect for perchlorate dose was not significant for any of the nine dependent variables, and there were no reliable interactions for treatment. The highest dose tested, 8.5 mg perchlorate/kg/day, is considered a NOAEL for neurodevelopmental effects in this study.

Because of EPA's concerns that the changes in motor activity in the rats in the two studies summarized above had biological significance, the results of both studies were re-analyzed (Dunson 2001). Each study was re-analyzed separately and combined using a Bayesian Hierarchical Modeling Approach. According to Dunson (2001), the re-analysis showed evidence of an increasing dose-response trend in

motor activity in both studies, though the effect in the York et al. (2004) study was less pronounced. After reviewing the two studies in question and the re-analysis by Dunson, NAS (2005) concluded that: "general motor activity is not necessarily the most relevant or most sensitive aspect of motor function to assess if neonatal hypothyroidism is the suspected mechanism of action".

A cross-fostering study in rats was conducted by Mahle et al. (2003). Pregnant Sprague-Dawley rats were administered ammonium perchlorate in the drinking water at doses of 0 or 1 mg perchlorate/kg/day from GD 2 to PND 21. Cross-fostering was done on PND 1 such that four groups of pups were formed: never exposed, exposed in utero and via maternal milk, exposed only in utero, and exposed only via maternal milk. Dams and pups were sacrificed on PND 10. There was no indication of maternal toxicity during the study. However, serum T4 was significantly decreased and TSH was significantly increased in exposed dams that nursed their own pups; TSH was also increased in dams that nursed unexposed pups. The two cross-fostered litters (exposed only *in utero* and exposed only via nursing) had significantly lower weight than control pups and than pups exposed both in utero and via milk. T3 was not significantly affected in any pup group (male or female). T4 was significantly reduced in female pups exposed only via milk and in females exposed in utero plus via milk; the decrease was more marked in the latter group. T4 was not significantly affected in male pups. TSH was increased significantly in male and female pups (more pronounced in females) from groups that received double exposure and in groups exposed only via milk; there was no significant difference between these two groups. The results suggest that: (1) exposure in utero to perchlorate at the dose tested had little or no impact on serum levels of thyroid hormone and TSH measured in pups on PND 10, (2) the changes in serum thyroid hormone and TSH levels seen in PND 10 pups exposed both in utero and via maternal milk appear to be completely due to postnatal exposure to perchlorate through lactation, and (3) perchlorate could be acting directly on the pups' thyroid and/or may be limiting the availability of iodide to nursing pups by inhibiting NIS in breast tissue.

The developmental effects of perchlorate were also examined in rabbits administered 0, 0.09, 0.85, 8.5, 25.5, or 85 mg perchlorate/kg/day in the drinking water on GD 6–28 (York et al. 2001b). Sacrifices were conducted on GD 29. There were no deaths attributed to treatment with the test material or chemical-related clinical signs, or effects on body weight or uterine weight. There were no compound-related effects on any of the litter parameters studied including litter averages for corpora lutea, implantations, litter sizes, live and dead fetuses, percent dead or resorbed fetuses, and fetal body weights. All placentae appeared normal. There were no treatment-related increases in gross alterations or in skeletal and soft

tissue anomalies. This study defined a maternal NOAEL of 0.85 mg/kg/day (see Endocrine Effects section for summary of maternal effects) and a developmental NOAEL of 85 mg/kg/day.

Developmental NOAEL and LOAEL values from these studies are shown in Table 3-2 and Figure 3-2.

3.2.2.7 Cancer

Limited information was located regarding exposure to perchlorate and cancer in humans. In the ecologic study by Li et al. (2001) described earlier, the prevalence of thyroid cancer was not significantly higher among residents from Clark County (Las Vegas), whose drinking water had 4–24 μ g/L of perchlorate (0.0001–0.0007 mg perchlorate/kg/day) than in residents from another urban area of similar size (Reno, Washoe County), but with no perchlorate in the water, or than those from all other counties, also with no perchlorate exposure.

Morgan and Cassady (2002) conducted an ecologic study among residents of 13 contiguous census tracts in Redlands, California, San Bernardino County. Residents had been exposed to various concentrations of trichloroethylene (TCE) and ammonium perchlorate. Testing for TCE began in 1980, whereas, testing for perchlorate began in 1997. The concentration of perchlorate in the wells in 2001 was reported to be in the range 5-98 ppb, with drinking water concentrations not exceeding 18 ppb. The concentration of TCE in the wells initially ranged from 0.09 to 97 ppb, but did not exceed 5 ppb in the drinking water since 1991 after the water underwent treatment or the highly contaminated wells were removed from service. The standardized incidence ratios (SIRs, observed/expected) for all cancers combined or for any specific cancer site was not significantly different than 1.00, except for colon and rectum (SIR, 0.86; 99% CI, 0.74–0.99) and lung and bronchus (SIR, 0.71; 99% CI, 0.61–0.81), which were lower than expected, and melanoma of the skin (SIR, 1.42; 99% CI, 1.13–1.77) and uterine cancer (SIR, 1.35; 99% CI, 1.06–1.70), which were higher than expected. The SIR for thyroid cancer was 1.0 (99% CI, 0.63-1.47) based on 40 observed cases. When the analysis was restricted to children, no cancers were observed more often than expected. NAS (2005) notes that limitations of the study include the fact that timing and duration of exposure to perchlorate is unclear, that there also was exposure to TCE, and that there was no adjustment for other potential confounding variables. NAS (2005) further notes that the expected numbers were derived from the four-county region as a whole, which included the exposed community, not from an unexposed area. The latter could have resulted in an underestimate of the SIR.

PERCHLORATES 70 3. HEALTH EFFECTS

Potassium and sodium perchlorates have been shown to produce thyroid tumors (papillary and/or follicular adenomas and/or carcinomas) in rats and mice with long-term exposure (1–24 months) to 1–1.2% concentrations in the feed or drinking water (Fernandez-Rodriguez et al. 1991; Florencio Vicente 1990; Gauss 1972; Kessler and Kruskemper 1966; Pajer and Kalisnik 1991; Toro Guillen 1991). Estimated doses in these studies ranged from 928 to 2,573 mg perchlorate/kg/day. The cancer effect levels from these studies are shown in Table 3-2 and Figure 3-2. In a related study in rats, Fernández-Santos et al. (2004) determined the incidence of Ki-ras oncogene mutations in follicular cell carcinomas of the thyroid induced by administration of radioactive iodine and potassium perchlorate (1% in drinking water) for up to 18 months. Direct sequencing showed no mutations in the amplified gene segment of any of the induced thyroid tumors. The results suggested that Ki-ras activation via mutations at codons 12 and 13 is neither a constant event nor an early event in the development of rat thyroid follicular cell carcinoma. An additional study found that low level exposure to potassium perchlorate (0.1% in the feed, corresponding to a dose of 64 mg perchlorate/kg/day) for 19 weeks promoted the development of thyroid tumors initiated by N-bis(2-hydroxypropyl)nitrosamine (Hiasa et al. 1987).

NAS (2005) noted that: "on the basis of the understanding of the biology of human and rodent thyroid tumors, it is unlikely that perchlorate poses a risk of thyroid cancer in humans". The EPA has concluded that perchlorate is not likely to pose a risk of thyroid cancer in humans, at least at doses below those necessary to alter thyroid hormone homeostasis, based on the hormonally-mediated mode of action in rodent studies and species differences in thyroid function (IRIS 2005).

3.2.3 Dermal Exposure

No studies were located regarding the following effects in humans or animals after dermal exposure to perchlorate:

- 3.2.3.1 Death
- 3.2.3.2 Systemic Effects
- 3.2.3.3 Immunological and Lymphoreticular Effects
- 3.2.3.4 Neurological Effects
- 3.2.3.5 Reproductive Effects

3.2.3.6 Developmental Effects

3.2.3.7 Cancer

3.3 GENOTOXICITY

No studies were located regarding genotoxic effects in humans after inhalation, oral, or dermal exposure to perchlorates. Limited information is available from studies in animals. Siglin et al. (2000) found no evidence of bone marrow erythrocyte micronucleus formation in male and female rats as a result of exposure to 8.5 mg perchlorate/kg/day as ammonium perchlorate in the drinking water for 90 days. Zeiger et al. (1998b) also reported no increase in micronucleus formation in bone marrow from mice injected intraperitoneally with 500 mg of ammonium perchlorate/kg/day for 3 consecutive days; higher doses were lethal to the mice. Cyclophosphamide was used as positive control in both studies.

Magnesium perchlorate was negative in a test for SOS-inducing activity in *Salmonella typhimurium* strain 1535 (Nakamura and Kosaka 1989) and in a test for production of deoxyribonucleic acid (DNA)-protein cross links in cultured human lymphocytes (Costa et al. 1996). Zeiger et al. (1998a) found no evidence of mutagenicity for ammonium perchlorate with or without metabolic activation in six different *Salmonella* strains. Ammonium perchlorate was not mutagenic in the mouse lymphoma assay with or without metabolic activation (San and Clarke 1999).

The available data suggest that perchlorate is not a mutagenic or clastogenic agent.

3.4 TOXICOKINETICS

Overview. Short-term studies on humans and animals demonstrate that perchlorate appears to be readily absorbed by the digestive system after oral exposure. Maximum blood levels appear within a few hours after ingestion. Perchlorate is rapidly taken up into the thyroid gland, by an active transport mechanism, and reaches a maximum level in the thyroid in approximately 4 hours in rats. Elimination of perchlorate from the thyroid is also rapid; half-times of 10–20 hours have been estimated in rats. Perchlorate does not appear to be modified in the body, either by degradation or covalent binding. Perchlorate is rapidly eliminated from the body in the urine with half-times of approximately 8–12 hours in humans and 10–20 hours in rats. No studies on the kinetics of long-term administration of perchlorate in humans or animals have been reported.

3.4.1 Absorption

3.4.1.1 Inhalation Exposure

No studies were found regarding quantitative absorption of perchlorate after inhalation exposure. Occupational studies have measured urinary perchlorate in workers, suggesting that pulmonary absorption may occur (Lamm et al. 1999), although swallowing of particles may have also occurred. Under normal ambient temperatures, the vapor pressure of a perchlorate salt solution is expected to be low, which would reduce the likelihood of exposure to perchlorate fumes or vapors from that source. However, if perchlorate particles were suspended in air, absorption by inhalation would be possible, depending on the particle size. Given the aqueous solubility of perchlorate salts, it is likely that small particles reaching the alveoli would dissolve and readily enter the systemic circulation.

3.4.1.2 Oral Exposure

Perchlorate has been shown, in both human and animal studies, to be readily absorbed after oral exposure. In human subjects who ingested 10 mg/day perchlorate as potassium perchlorate in drinking water for 14 days (0.14 mg/kg/day), urinary excretion rate of perchlorate was 77% of the dose/day, after 7 days of exposure, indicating that at least 77% of the ingested dosage had been absorbed (Lawrence et al. 2000). Evidence for rapid absorption in humans is provided by two studies of elimination patterns. Anbar et al. (1959) detected potassium perchlorate in urine samples collected from four subjects 3 hours after ingestion of 200 mg perchlorate. Durand (1938) gave sodium perchlorate in a single oral dose (784 mg perchlorate per person) to two individuals and found perchlorate in the urine as early as 10 minutes after ingestion. Approximately 30% of the ingested dose had been eliminated in the urine within 3 hours after the dose, and 95% was eliminated within 48 hours; these results suggest rapid and near complete absorption of perchlorate through the digestive system. Selinova et al. (1986) examined the absorption of ammonium perchlorate in rats, rabbits, and calves after a single oral dose (2, 20, 200, or 600 mg perchlorate/kg). In rats, a maximum concentration of perchlorate in blood was noted between 30 and 60 minutes after administration (suggesting entrance into the systemic circulation before 30 minutes); in cattle, the maximum blood concentration of perchlorate occurred at 5 hours. In this study, only 8.5% of the administered dose was excreted in feces, and the rest was excreted in the urine, suggesting that >90% of the administered oral dose was absorbed.

3.4.1.3 Dermal Exposure

No studies were found regarding absorption of perchlorate after dermal exposure. As a general rule, electrolytes applied from aqueous solutions do not readily penetrate the skin (Scheuplein and Bronaugh 1983). On this basis, dermal absorption of perchlorate is expected to be low.

3.4.2 Distribution

Perchlorate binds to bovine and human serum albumin (Carr 1952; Scatchard and Black 1949). Perchlorate binds only weakly to either of the two binding sites of transferrin (association constants 7 and 35 M) (Harris et al. 1998).

Studies conducted in rabbits and rats indicate that perchlorate concentrations in most soft tissues (e.g., kidney, liver, skeletal muscle) are similar to the serum concentrations; tissue:serum concentration ratios >1 have been found in thyroid (5–10) and skin (1–2) (Durand 1938; Yu et al. 2002). Accumulation of perchlorate in the thyroid occurs by a saturable, active transport process (see Section 3.5.1). As a result, thyroid serum concentrations and the amount of perchlorate in the thyroid as a fraction of the absorbed dose decrease with increasing dose (Chow and Woodbury 1970). Elimination of perchlorate from the thyroid gland is relatively rapid, with half-times in rats estimated to be approximately 10–20 hours (Fisher et al. 2000; Goldman and Stanbury 1973; Yu et al. 2002).

Studies conducted in rats administered intravenous injections of perchlorate indicate that perchlorate is secreted into the gastric lumen (Yu et al. 2002). Perchlorate secreted into the gastric lumen may be absorbed in the small intestine.

3.4.2.1 Inhalation Exposure

No studies were found in humans or in animals regarding distribution of perchlorate after inhalation exposure.

3.4.2.2 Oral Exposure

In a survey of 36 healthy lactating volunteers, perchlorate was detected in breast milk at a mean concentration of 10.5 ppb (range, 0.6–92. ppb) (Kirk et al. 2005). Exposure of the lactating women was presumed to have occurred mainly from perchlorate in food and drinking water. No correlation was apparent between the concentration of perchlorate in the breast milk and the water that the respective mothers consumed. Given the small number of samples analyzed, these results should be considered preliminary.

Studies conducted in rabbits and rats indicate that perchlorate concentrations in most soft tissues (e.g., kidney, liver, skeletal muscle) are similar to the serum concentrations; tissue:serum concentration ratios >1 have been found in thyroid (5–10) and skin (1–2) (Durand 1938; Yu et al. 2002). Accumulation of perchlorate in the thyroid occurs by a saturable, active transport process (see Section 3.5.1).

Perchlorate has been shown to cross the placenta of rats. In rats exposed to perchlorate in drinking water, fetal:maternal serum concentration ratios were approximately 1 when the maternal dosage was 1 mg/kg/day or lower, and were <1 when the maternal dosage was 10 mg/kg/day, suggesting the possibility of a dose-dependent limitation in the capacity of transplacental transfer (Clewell et al. 2003a).

3.4.2.3 Dermal Exposure

No studies were found regarding distribution of perchlorate after dermal exposure.

3.4.2.4 Other Routes of Exposure

Several studies have examined the distribution of perchlorate in animals after intravenous, intramuscular, or peritoneal injection (Anbar et al. 1959; Chow and Woodbury 1970; Chow et al. 1969; Durand 1938; Goldman and Stanbury 1973; Yu et al. 2002). These studies have shown that absorbed perchlorate, regardless of the route of exposure, will distribute to soft tissues, including adrenal, brain, kidney, liver, mammary gland, skeletal muscle, spleen, testes, and thyroid. The highest concentrations occur in the thyroid, where tissue:serum concentration ratios of 5–10 have been observed (Chow and Woodbury 1970). The elimination half-time for the thyroid was estimated in rats to be approximately 10–20 hours (Fisher et al. 2000; Goldman and Stanbury 1973; Yu et al. 2002).

Other tissues that appear to concentrate perchlorate are the salivary gland and skin, although not to the same degree as the thyroid (Anbar et al. 1959; Lazarus et al. 1974; Yu et al. 2002). Tissue:blood concentration ratios of 1.5–2 have been observed for the salivary gland (Anbar et al. 1959) and 1–2 for the skin (Yu et al. 2002).

3.4.3 Metabolism

There is no evidence that perchlorate is metabolized in the body. Anbar et al. (1959) assayed for potential metabolites of potassium perchlorate (radiolabeled with ³⁶Cl and ¹⁸O₄) in the urine of patients 3 hours after a single oral dose (200 mg perchlorate per person). They did not detect any isotopic exchange of the oxygen atoms in excreted perchlorate; furthermore, although they found that 1–3% of the excreted ³⁶Cl was chloride ion, this value was within experimental error. They concluded that the perchlorate excreted after 3 hours was unmodified. There has been no investigation as to whether perchlorate that is eliminated at later time points would exhibit the same isotopic pattern.

Goldman and Stanbury (1973) found that perchlorate reached a maximum concentration (>3% of the administered dose/g tissue) in the thyroid gland of rats 4 hours after an intraperitoneal injection of radiolabeled potassium perchlorate (K ³⁶ClO₄; 18 or 24 mg perchlorate/kg). However, trichloroacetic acid precipitates of thyroid homogenates contained only background levels of radioactivity, indicating that perchlorate is not covalently bound to thyroid protein.

3.4.4 Elimination and Excretion

The few studies of the elimination and excretion of perchlorate, described in the sections that follow, suggest that it is rapidly eliminated from the body through the urinary tract. Similar results have been obtained after oral exposure or after intravenous or intraperitoneal injection; the specific cation appears not to influence the pattern of excretion.

3.4.4.1 Inhalation Exposure

A study in two workers occupationally exposed to perchlorate found that the urinary perchlorate concentration increased over 3 days of perchlorate exposure, but there was a decrease between the

12-hour work shifts (Lamm et al. 1999). Excretion after the last exposure appeared to follow a first-order kinetics pattern, particularly when the urinary perchlorate concentration was between 0.1 and 10 mg/L. The average elimination half-life for the two workers was approximately 8 hours. No information was located regarding excretion of perchlorate in animals following inhalation exposure.

3.4.4.2 Oral Exposure

In adult human subjects who ingested potassium perchlorate in drinking water (0.14 mg/kg/day) for 14 days, urinary excretion rate of perchlorate was 77% of the dose/day after 7 days of exposure, indicating that urine is the main excretory pathway for absorbed perchlorate (Lawrence et al. 2000). The urinary excretion rate of perchlorate returned to control levels (<0.5 mg/day) within 14 days after exposure to perchlorate was terminated (Lawrence et al. 2000). Perchlorate was detected in the urine of two adults at 10 minutes after a single oral dose of sodium perchlorate (784 mg perchlorate per person); urinary excretion as a percentage of the dose was 30% at 3 hours, 50% in at 5 hours, 85% at 24 hours, and 95% at 48 hours (Durand 1938). This suggests an excretion half-time of approximately 12 hours. The latter estimate is consistent with the elimination kinetics of perchlorate from serum. The elimination half-time for perchlorate in serum was estimated to be approximately 8 hours in adult human subjects who ingested potassium perchlorate in drinking water (0.5 mg/kg/day) for 14 days (Greer et al. 2002). Thus, in humans, perchlorate is rapidly eliminated and would not be expected to accumulate in the body with prolonged exposure. Based on an elimination half-time of approximately 8–12 hours, a steady state would be achieved within 3–4 days of continuous exposure. The detection of perchlorate in breast milk from lactating women (Kirk et al. 2005) also indicates breast milk as an excretion route in humans.

Studies conducted in a variety of experimental animals, including rats, rabbits, and calves, have shown that absorbed perchlorate is rapidly and nearly completely excreted in the urine (Fisher et al. 2000; Selivanova et al. 1986; Yu et al. 2002).

Studies conducted in rats have shown that perchlorate is excreted in mammary milk (Clewell et al. 2003b). Perchlorate has also been detected in dairy milk (Howard et al. 1996; Kirk et al. 2005).

3.4.4.3 Dermal Exposure

No studies were found regarding elimination or excretion of perchlorates after dermal exposure.

3.4.4.4 Other Routes of Exposure

Studies in which rats received intravenous or intraperitoneal injections of perchlorate provide additional support for the rapid excretion of perchlorate in urine. Rats that received a single intravenous injection of 0.01, 0.1, 1.0, or 3.0 mg/kg perchlorate (as ammonium perchlorate) excreted 85, 86, 80, or 79% of the administered dose, respectively, in urine (Fisher et al. 2000). The elimination half-time for intravenously injected perchlorate (approximately 0.04 mg, 0.18–0.25 mg/kg, as potassium perchlorate) from serum, and the urinary excretion half-time were estimated in rats to be approximately 20 hours (Goldman and Stanbury 1973). Similarly, rats injected with sodium perchlorate (2, 8, or 49 mg perchlorate/kg) excreted 50% of the administered dose in urine during the first 6 hours and had excreted 93–97% of the dose by 60 hours (Eichler and Hackenthal 1962); in this study, higher doses of perchlorate were eliminated at a faster rate than lower doses. Similar results were obtained in rats that received a single intravenous dose of 3.3 mg/kg perchlorate as ammonium perchlorate; urinary excretion of perchlorate was essentially complete within 12 hours (Yu et al. 2002).

3.4.5 Physiologically Based Pharmacokinetic (PBPK)/Pharmacodynamic (PD) Models

Physiologically based pharmacokinetic (PBPK) models use mathematical descriptions of the uptake and disposition of chemical substances to quantitatively describe the relationships among critical biological processes (Krishnan et al. 1994). PBPK models are also called biologically based tissue dosimetry models. PBPK models are increasingly used in risk assessments, primarily to predict the concentration of potentially toxic moieties of a chemical that will be delivered to any given target tissue following various combinations of route, dose level, and test species (Clewell and Andersen 1985). Physiologically based pharmacodynamic (PBPD) models use mathematical descriptions of the dose-response function to quantitatively describe the relationship between target tissue dose and toxic end points.

PBPK/PD models refine our understanding of complex quantitative dose behaviors by helping to delineate and characterize the relationships between: (1) the external/exposure concentration and target tissue dose of the toxic moiety, and (2) the target tissue dose and observed responses (Andersen et al. 1987; Andersen and Krishnan 1994). These models are biologically and mechanistically based and can be used to extrapolate the pharmacokinetic behavior of chemical substances from high to low dose, from route to route, between species, and between subpopulations within a species. The biological basis of

PBPK models results in more meaningful extrapolations than those generated with the more conventional use of uncertainty factors.

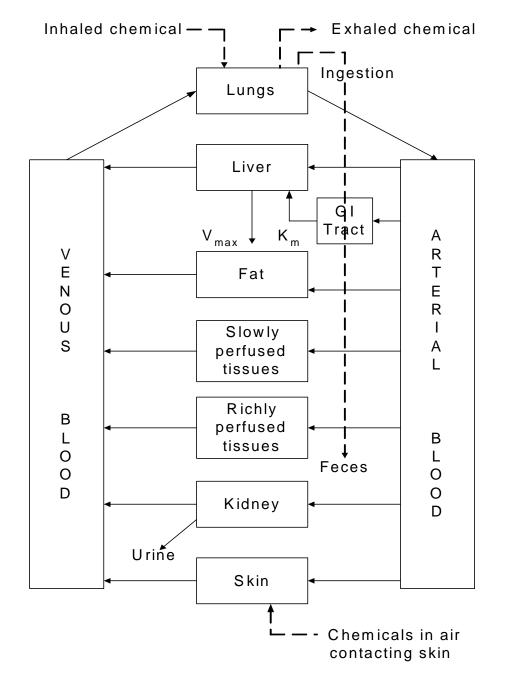
The PBPK model for a chemical substance is developed in four interconnected steps: (1) model representation, (2) model parameterization, (3) model simulation, and (4) model validation (Krishnan and Andersen 1994). In the early 1990s, validated PBPK models were developed for a number of toxicologically important chemical substances, both volatile and nonvolatile (Krishnan and Andersen 1994; Leung 1993). PBPK models for a particular substance require estimates of the chemical substance-specific physicochemical parameters, and species-specific physiological and biological parameters. The numerical estimates of these model parameters are incorporated within a set of differential and algebraic equations that describe the pharmacokinetic processes. Solving these differential and algebraic equations provides the predictions of tissue dose. Computers then provide process simulations based on these solutions.

The structure and mathematical expressions used in PBPK models significantly simplify the true complexities of biological systems. If the uptake and disposition of the chemical substance(s) are adequately described, however, this simplification is desirable because data are often unavailable for many biological processes. A simplified scheme reduces the magnitude of cumulative uncertainty. The adequacy of the model is, therefore, of great importance, and model validation is essential to the use of PBPK models in risk assessment.

PBPK models improve the pharmacokinetic extrapolations used in risk assessments that identify the maximal (i.e., the safe) levels for human exposure to chemical substances (Andersen and Krishnan 1994). PBPK models provide a scientifically sound means to predict the target tissue dose of chemicals in humans who are exposed to environmental levels (for example, levels that might occur at hazardous waste sites) based on the results of studies where doses were higher or were administered in different species. Figure 3-3 shows a conceptualized representation of a PBPK model.

If PBPK models for perchlorates exist, the overall results and individual models are discussed in this section in terms of their use in risk assessment, tissue dosimetry, and dose, route, and species extrapolations.

Figure 3-3. Conceptual Representation of a Physiologically Based Pharmacokinetic (PBPK) Model for a Hypothetical Chemical Substance



Source: adapted from Krishnan et al. 1994

Note: This is a conceptual representation of a physiologically based pharmacokinetic (PBPK) model for a hypothetical chemical substance. The chemical substance is shown to be absorbed via the skin, by inhalation, or by ingestion, metabolized in the liver, and excreted in the urine or by exhalation.

AFRL Perchlorate Biokinetics Models

Description of the models. The Air Force Research Laboratory (AFRL) developed PBPK models of the kinetics of ingested or injected perchlorate in rats and humans (Fisher et al. 2000; Merrill et al. 2003, 2005). The models were developed simultaneously with models of iodide biokinetics. When combined, the perchlorate and iodide models simulate the competitive inhibition of iodide transport by perchlorate in thyroid and other tissues that have NIS activity. The adult rat model has been extended to include pregnancy and maternal-fetal transfer of perchlorate, and lactation and maternal-pup perchlorate transfer through milk (Clewell et al. 2003a, 2003b).

The adult rat and human models have the same structure and differ only in values for physiological and some of perchlorate parameters (Table 3-3, Figure 3-4). Both models simulate nine tissue compartments: blood, kidney, liver, skin, stomach, thyroid, fat, other slowly perfused tissues, and other richly perfused tissues. Uptakes from blood into the tissue vascular compartments are simulated as flow-limited processes. Distributions within blood, skin, stomach, and thyroid are simulated as diffusion limited processes with first-order clearance terms. Transport of perchlorate within tissues that have NIS activity are simulated with tissue-specific affinity constants and maximum velocities. This includes uptake of perchlorate into thyroid follicle cells and secretion of perchlorate into the follicle lumen. Active transport of perchlorate into the stomach lumen and in skin is also simulated in the models. Excretion is simulated with a first-order clearance term for transfer of perchlorate from the kidney into urine. The assumption that perchlorate is transported into the thyroid gland by the NIS is based on observations of accumulation of perchlorate by the gland and on numerous studies that have shown that perchlorate competitively inhibits iodide uptake into thyroid tissue and thyroid follicle cells. However, studies of transport kinetics of NIS suggest that, while perchlorate may bind to and competitively inhibit iodide transport, it may not be a substrate for transport by the NIS (Eskandari et al. 1997; Yoshida et al. 1997).

Extensions of the adult rat model to simulate perchlorate kinetics during pregnancy include the addition of two additional compartments representing the mammary gland and placenta (Clewell et al. 2003a, 2003b). Uptake of perchlorate into the mammary gland tissue from the mammary tissue vascular space is simulated as a capacity-limited transport process, representing the activity of NIS in this tissue. Uptake of perchlorate into the placenta from blood is simulated as a flow-limited process. Exchanges of perchlorate between the placenta and fetus are simulated with first-order clearance terms. The fetal model is identical in structure to the adult (nonpregnant) model, with adjustments in the physiological and perchlorate parameters to reflect the fetus.

Table 3-3. Perchlorate and Iodide Parameter Values for the AFRL Adult Male Rat and Human PBPK Models

	Male rat		Human	
Parameter	Perchlorate	lodide	Perchlorate	Iodide
Partition coefficients (unitless)				
Slowly perfused/plasma	0.31	0.21	0.31	0.21
Richly perfused/plasma	0.56	0.40	0.56	0.40
Fat/plasma	0.05	0.05	0.05	0.05
Kidney/plasma	0.99	1.09	0.99	1.09
Liver/plasma	0.56	0.44	0.56	0.44
Gastric tissue/gastric blood	0.70	1.0	1.80	0.50
Gastric juice/gastric tissue	1.70	3.50	2.30	3.50
Skin tissue/skin blood	1.0	0.70	1.15	0.70
Thyroid follicle/thyroid stroma	0.15	0.15	0.13	0.15
Thyroid lumen/thyroid follicle	8.00	8.00	7.00	7.00
Red blood cells/plasma	0.73	1.00	0.80	1.00
Max capacity, (ng/hour-kg)				
Thyroid follicle	$1.0x10^3$	5.4x10 ⁴	2.5x10 ⁵	$2.5x10^{5}$
Thyroid lumen	$2.0x10^4$	$4.0x10^6$	5.0x10 ⁴	$6.0x10^5$
Skin	5.0x10 ⁵	5.0x10 ⁵	1.0x10 ⁶	9.0x10 ⁵
Gastric	$2.0x10^4$	$2.0x10^6$	1.0x10 ⁵	1.0x10 ⁵
Plasma binding	3.4x10 ⁻³	$1.0x10^{2}$	$5.0x10^2$	$2.0x10^{2}$
Affinity constants, (ng/L)				
Thyroid lumen	1.0x10 ⁸	1.0x10 ⁹	1.0x10 ⁸	1.0x10 ⁹
Thyroid follicle	1.8x10 ⁵	$4.0x10^6$	1.6x10 ⁵	$4.0x10^6$
Skin	1.8x10 ⁵	$4.0x10^6$	2.0x10 ⁵	$4.0x10^6$
Gastric	1.7x10 ⁵	$4.0x10^6$	2.0x10 ⁵	$4.0x10^{6}$
Plasma binding	1.1x10 ⁴	_	1.8x10 ⁴	$7.8x10^{5}$
Permeability area cross products (L/hour-kg)			
Gastric blood to gastric tissue	1.00	1.00	0.6	0.2
Gastric tissue to gastric juice	0.80	0.10	8.0	2.0
Skin blood to skin tissue	0.80	0.10	1.0	0.01
Plasma to red blood cells	1.00	1.00	1.0	1.0
Thyroid follicle to thyroid stroma	6.0x10 ⁻⁵	1.0x10 ⁻⁴	1.0x10 ⁻⁴	6.0x10 ⁻⁴
Thyroid lumen to thyroid follicle	0.01	4.0x10 ⁻⁷	0.01	1.0x10 ⁻⁴

^{***}DRAFT FOR PUBLIC COMMENT***

3. HEALTH EFFECTS

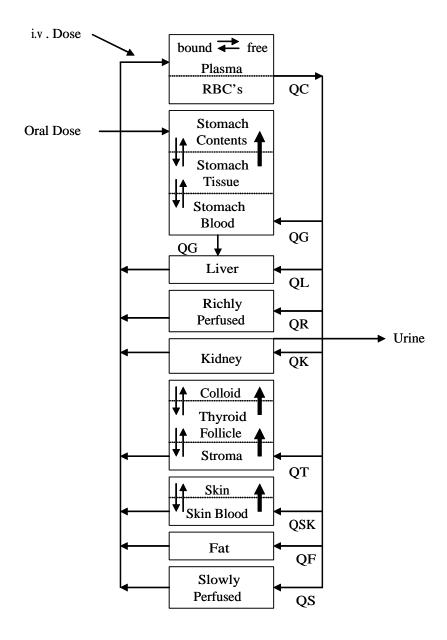
Table 3-3. Perchlorate and Iodide Parameter Values for the AFRL Adult Male Rat and Human PBPK Models

	Male	Male rat		Human	
Parameter	Perchlorate	lodide	Perchlorate	lodide	
Clearance values (L/hour-kg)					
Urinary excretion	0.07	0.05	0.13	0.11	
Plasma unbinding	0.032	_	0.025	_	
Hormone production	_	0.10	_	0.01	
Hormone secretion	_	1.2x10 ⁻⁶	_	1.2x10 ⁻⁶	
Hormone deiodination	-	_	_	9.0x10 ⁻⁴	

Adapted from Merrill et al. (2003, 2005).

AFRL = Air Force Research Laboratory; PBPK = physiologically based pharmacokinetic

Figure 3-4. Structure of AFRL PBPK Model of Perchlorate in Adult Male Humans and Rats



Thick arrows within tissue compartments indicate transfers that are affinity- and capacity-limited. Thin arrows within tissue compartments are diffusion limited transfers. Q indicates flow for flow-limited transfers.

AFRL= Air Force Research Laboratory; PBPK = physiologically based pharmacokinetic

The lactating rat model includes a milk compartment in mammary tissue and a first-order clearance term for describing secretion of perchlorate from mammary tissue into milk (Clewell et al. 2003b). Transfer of perchlorate from milk to the neonate is simulated as a first-order clearance process. The neonate model is identical in structure to the adult (nonpregnant) model, with adjustments to the physiological and perchlorate parameter values to reflect the neonate (Clewell et al. 2003a).

Validation of the models. The rat perchlorate model has been evaluated for predicting kidney, serum, gastric lumen, tissue (including thyroid), and urine perchlorate concentrations in adult rats that received acute intravenous injection of radiolabeled perchlorate (36ClO₄, Merrill et al. 2003a; Yu et al. 2002). In general, model predictions were within 1–2 standard deviations of observed values. When the same parameter values were used to predict perchlorate concentrations in the thyroid in rats that were exposed to repeated doses of perchlorate in drinking water, model predictions did not agree with observations for dosages ≥3 mg/kg/day; however, good correspondence with observations was achieved by adjusting the partition coefficient for transfer of perchlorate across the thyroid follicle cell membrane. Similarly, the model also predicted reasonably well the inhibition of radioiodine uptake in the thyroid produced by an acute intravenous dose of perchlorate; however, the model underpredicted thyroid iodide uptake in rats that received perchlorate in drinking water for 14 days at doses >1 mg/kg/day. Thus, the model simulated a greater inhibition of thyroid uptake of iodide in animals that received repeated doses of perchlorate than was actually observed. The inability of the model to accurately predict the effect of repeated exposures to perchlorate on thyroid iodide uptake is not surprising, since the model does not simulate the hormonal regulation of NIS activity and organification of iodide in the thyroid. In animals that received repeated exposures to perchlorate, induction of NIS and thyroid hormone production are likely to have occurred secondary to elevations in serum TSH (Uyttersprot et al. 1997; Yu et al. 2002). Such a response could have partially restored thyroid iodide uptake to higher levels than would be predicted if induction were not taken into account.

The adult human model also predicted reasonably well (i.e., within 1–2 standard deviations of observations) perchlorate concentrations in plasma and urine in subjects who received oral doses of perchlorate (Durand 1938; Eichler 1929; Kamm and Drescher 1973; Greer et al. 2002; Merrill et al. 2005). Model predictions of radioiodide in gastric juice, serum, thyroid, and urine following an intravenous dose of radioiodide also corresponded with observations made in healthy adults (Hays and Solomon 1965). Model predictions of thyroid radioiodine uptake in subjects who received oral doses of perchlorate agreed with observations when the kinetic parameters for iodide in the thyroid (i.e., maximum

transport into the thyroid follicle) were adjusted to achieve good correspondence to the observations (Greer et al. 2002; Merrill et al. 2005). When the model was calibrated by adjusting the maximum transport rate for iodide into the thyroid follicle, it accurately predicted the observed time course for radioiodine uptake in a Graves' disease patient who received a single tracer dose of radioiodine (Stanbury and Wyngaarden 1952); however, the model substantially overpredicted iodide uptake after the same patient received a dose of perchlorate. Here again, the error in predictions of the effect of perchlorate on iodide uptake may reflect humoral regulation of iodide transport and organification mechanisms or a response to perchlorate in Graves' disease patients that is not simulated in the model.

The rat maternal/fetal model was evaluated by comparing predictions of perchlorate concentrations in maternal and fetal serum and maternal thyroid in rats exposed to perchlorate in drinking water (Clewell et al. 2001, 2003a). Model predictions agreed well (within 1–2 standard deviations of observations) with observations. Predictions of maternal and fetal radioiodine uptakes in thyroid were also in reasonable agreement with observations in rats that received single injections of iodine with or without single injections or oral gavage doses of perchlorate, or at the conclusion of 18 days of exposures to perchlorate in drinking water (Brown-Grant 1966; Clewell et al. 2001, 2003a; Sztanyik and Turai 1988).

Similar outcomes occurred in evaluations of the lactating dam/neonate model (Clewell et al. 2003b). The model accurately predicted serum and thyroid iodide concentrations in the dam and neonate following single intravenous injections of radioactive iodine, with or without concurrent injection of perchlorate, and in maternal thyroid following an 18-day exposure to perchlorate in drinking water (Clewell et al. 2003b). Model predictions of radioiodide levels in mammary gland and milk, in rats that did or did not receive single doses of perchlorate, corresponded with observations (Clewell et al. 2003b).

Risk assessment. The rat and human models have been used to calculate human equivalent exposure levels for perchlorate that would be expected to produce the same degree of inhibition of iodide uptake in to the thyroid gland (Clewell et al. 2001). These estimates have been used to extrapolate doseresponse relationships for perchlorate observed in rats to humans.

Target tissues. The perchlorate models are designed to calculate perchlorate concentrations in serum and thyroid and inhibition of iodide uptake into the thyroid produced by exposures to perchlorate.

Species extrapolation. The models are designed for applications to rat or human dosimetry and cannot be applied to other species without modification.

Interroute extrapolation. The models are designed to simulate intravenous or oral exposures to perchlorate and cannot be applied to other routes of exposure without modification.

3.5 MECHANISMS OF ACTION

3.5.1 Pharmacokinetic Mechanisms

Perchlorate is readily soluble in water and is quickly absorbed through the digestive tract. The mechanism by which perchlorate is transferred from the digestive system to the blood has not been investigated. Since Durand (1938) detected perchlorate in the urine of subjects 10 minutes after oral administration, it seems likely that absorption of perchlorate may begin in the stomach and continue in the small intestine. Anbar et al. (1959) determined that perchlorate eliminated in the urine 3 hours after an oral dose had not been metabolized (see Section 3.4.1). Whether microflora of the gut or intestinal enzymes modify perchlorate that is finally eliminated in the feces has not been investigated.

Whatever the mechanism of absorption, perchlorate is distributed throughout the body via the circulation (see Section 3.4.2). It apparently is not metabolized (Anbar et al. 1959) and it binds only weakly to cations. Concentrations of perchlorate rise above serum levels only for those tissues that are equipped with the anion transporter mechanism that normally takes up iodide. The effects of perchlorate on the thyroid gland are known from studies on humans and animals (see Section 3.2); perchlorate levels in the thyroid reach a maximum several hours after administration. Chow and co-workers (Chow and Woodbury 1970; Chow et al. 1969) determined that perchlorate is taken up from interstitial fluid by active transport at the base of thyroid follicular cells, which then actively transport it out into the follicular lumen. The effects of perchlorate on the transfer of maternal iodide in milk have been studied in rats and cattle (Clewell et al. 2003b; Howard et al. 1996; Kirk et al. 2005). The accumulation of perchlorate in ducts of the salivary gland has been described in mice (Lazarus et al. 1974). Studies on rodents have demonstrated that perchlorate can cross the placental barrier and affect the thyroid gland of the fetus (see Section 3.2).

The mechanism of transport of perchlorate into the thyroid gland and other tissues is unknown. Perchlorate is accumulated in thyroid follicle cells and lumen against an electrochemical gradient, indicating an active transport mechanism, and possibly different mechanisms at the basolateral and luminal membranes (Chow et al. 1969; Chow and Woodbury 1970; Clewell et al. 2004; Goldman and

Stanbury 1973). Thyroid uptake of perchlorate in hypophysectomized rats is stimulated by administration of TSH (Chow et al. 1969). Perchlorate competitively inhibits iodide transport in thyroid slices, cultured thyrocytes, cultured cells transformed to express thyroid NIS, and thyrocyte membrane vesicles (Eskandari et al. 1997; O'Neill et al. 1987; Wolff and Maurey 1962, 1963; Yoshida et al. 1997). The above observations suggest that perchlorate transport into thyroid follicle cells, and possibly into other tissues, may be mediated by NIS (Wolff 1998); however, efforts to obtain more direct evidence of NISmediated transport of perchlorate have not been successful. Chinese hamster ovary (CHO) cells transfected with the rat thyroid NIS gene and Xenopus oocytes transfected with rat thyroid NIS mRNA express active NIS that exhibits a Na(2):I(1) stoichiometry, is electrogenic (inward directed current), occurs against an electrochemical gradient for iodide in the presence of an inward electrochemical gradient for sodium, and is inhibited by perchlorate (Eskandari et al. 1997; Yoshida et al. 1997). Both systems, when clamped at an interior negative potential (40–50 mV), exhibit sodium-dependent inward currents in the presence of I and SCN; the transfected oocyte exhibits sodium-dependent inward currents in the presence of a variety of anions, including I, ClO₃, SCN, SeCN, NO₃, Br, BF₄, IO₄, BrO₃, SO₄-2, F⁻, and HPO₄⁻². However, these systems do not show perchlorate-stimulated currents in the presence or absence of a favorable inward-directed Na gradient. This suggests that either perchlorate is not transported by NIS or that NIS transport of perchlorate is electroneutral (e.g., Na+:I- stoichiometry of 1:1). Evidence for perchlorate being a transport substrate for NIS is the observation that COS-7 cells transfected to express NIS and FRTL5 cells transport the structural tetrahedral oxyanion analogs of perchlorate, perrhenate (RoO₄⁻) and pertechnetate (TcO₄⁻).

Studies of the kinetics of excretion and elimination of perchlorate from serum in humans indicate that absorbed perchlorate is excreted in urine with an elimination half-time of 8–14 hours (Durand 1938; Greer et al. 2002; Lawrence et al. 2000). Thus, in humans, perchlorate would not be expected to accumulate in the body with prolonged exposure. Based on an elimination half-time of approximately 8–12 hours, a steady state would be achieved within 3–4 days of continuous exposure. Rapid elimination of perchlorate (half-time of 20 hours) has also been observed in rats (Eichler and Hackenthal 1962; Fisher et al. 2000; Goldman and Stanbury 1973; Yu et al. 2002). The mechanisms of renal excretion of perchlorate are not understood.

3.5.2 Mechanisms of Toxicity

Perchlorate is an inhibitor of NIS, the primary mechanism by which iodide enters thyroid follicle cells from the blood, and the first step in the uptake of iodide into the thyroid and formation of thyroid

hormones (Figure 3-5; Carrasco 1993; Taurog 2000; Wolff 1998). All toxic effects of perchlorate on the thyroid hormone system derive directly or secondarily from this mechanism.

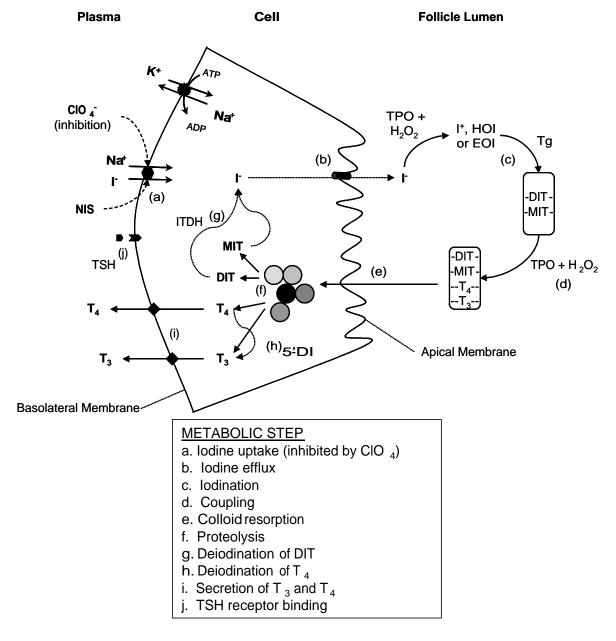
Thyroid Hormone. The thyroid hormone, T3, is essential for normal development of the nervous system and for the regulation of metabolism of cells in nearly all tissues of the body. Adverse effects on a wide variety of organ systems can result from disruption in the availability of T3 to target tissues. Organ systems affected by disturbances in T3 levels include the skin, cardiovascular system, pulmonary system, kidneys, gastrointestinal tract, liver, blood, neuromuscular system, central nervous system, skeleton, male and female reproductive systems, and numerous endocrine organs, including the pituitary and adrenal glands.

T3 exerts its wide range of actions by binding to thyroid hormone receptors (TRs) in the cell nucleus, which, when bound with hormone, modulate the transcription of a variety of genes (Anderson et al. 2000). TRs consist of a family of structurally similar proteins within the so-called *steroid receptor superfamily* that includes receptors for steroid hormones, vitamin D, retinoic acid, and peroxisomal proliferator activators (Lazar 1993). Each receptor has DNA binding domains capable of forming two zinc fingers; the sequence of the latter determine hormone receptor specificity to response elements on DNA that modulate gene transcription of hormone-sensitive genes. A ligand binding domain is responsible for conferring specificity for hormone binding.

Modulation of gene expression occurs when the T3–TR complex binds to a region of DNA associated with a thyroid hormone response element (TRE). Studies in humans and experimental animals have identified TREs associated with a variety of genes, including growth hormone, myelin basic protein, α -myosin heavy chain, malic enzyme and protein S14 (important in lipogenesis), sarcoplasmic reticulum Ca²– ATPase, Pcp-2 (in Purkinje cells), Na⁺/K⁺-ATPase, and TSH (Anderson et al. 2000; Klein and Levey 2000; Schwartz et al. 1994).

Adverse effects on cell metabolism and growth can result from either understimulation or overstimulation of target tissues by T3. The amount of T3 available to target tissues is highly controlled by feedback regulation of the production, secretion, and elimination of both T3 and its metabolic precursor, T4 (Figure 3-6). Major components of this mechanism include negative feedback control mediated by T4 and T3 of the synthesis and release of thyrotropin-releasing hormone (TRH) in the hypothalamus and of TSH in the pituitary. TRH stimulates the synthesis and secretion of TSH in the pituitary and modulates

Figure 3-5. Pathways Uptake and Metabolism of Iodide in the Thyroid Gland

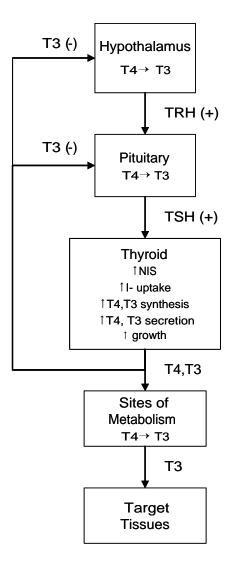


Source: Adapted from Taurog 2000

The diagram depicts a single thyroid follicle cell, with the plasma side of the follicle on the left and the follicle lumen on the right. lodide uptake (a) occurs through a Na+/I- symporter (NIS) in the basolateral membrane; the perchlorate ion competitively inhibits the NIS, preventing uptake of iodide into the follicle cell. Efflux into the follicle lumen (b) is thought to occur through an I- channel in the apical membrane. Iodination occurs in the follicle lumen (c). The enzyme thyroid peroxidase (TPO), depicted in the follicle lumen, is actually located in the apical membrane. Deiodination of iodotyrosines (g) is catalyzed by a microsomal enzyme, iodotyrosine dehalogenase (ITDH); monodeiodination of T4 (h) is catalyzed by the microsomal enzyme, 5'-diodinase. All steps in the uptake of iodine and synthesis of thyroid hormones (a–h) are stimulated by binding of thyroid stimulating hormone (TSH) to a receptor in the basolateral membrane.

DIT = diiodotyrosine; EOI = enzyme-linked species; HOI = hypoidous acid; ITDH = iodotyrosine dehalogenase; MMI = methimazole; MIT = monoiodotyrosine; PTU = propylthiouracil; T3 = triiodothyronine; T4 = thyronine; Tg = thyroglobulin; TPO = thyroid peroxidase; TSH = thyroid stimulating hormone

Figure 3-6. Hypothalamic-pituitary-thyroid (HPT) Feedback Pathways for Regulation of Thyroid Hormone Production and Secretion



T3 inhibits the synthesis and secretion of thyroid releasing hormone (TRH) in the paraventricular nuclei of the hypothalamus and the synthesis and secretion of thyroid stimulating hormone (TSH) in the thyrotrophs of the anterior pituitary. Most of the T3 in these tissues derives from local deiodination of T4; as a result, TRH and TSH synthesis and secretion are sensitive to circulating levels of both T3 and T4. Doses of perchlorate that decrease circulating levels of T3 or T4 can trigger the HPT feedback mechanism to stimulate thyroid growth, including hypertrophy and hyperplasia of follicle cells. Chronic stimulation of thyroid growth is thought to be contributor to the development of thyroid tumors in rats exposed to perchlorate.

the biologic potency of TSH. The latter is thought to result from effects of TRH on posttranslational glycosylation of TSH (Cohen et al. 2000; Scanlon and Toft 2000). TSH promotes growth of the thyroid gland follicle cells and stimulates thyroid iodide uptake and synthesis and secretion of T4 and T3 (Spaulding 2000).

The metabolism of T4 and T3 is also regulated by feedback control mechanisms (Darras et al. 1999). T3 is synthesized from the deiodination of T4 in a reaction catalyzed by selenium requiring, microsomal enzymes known as iodothyronine deiodinases. Although some production of T3 occurs in the thyroid, most of the T3 that is available to extrathyroidal target tissues derives from deiodination of T4 that occurs outside of the thyroid (Figure 3-6). The liver and kidney are thought to be major sites of production of T3 in the circulation; however, local tissue production of T3 from T4 is thought to be the predominant source of T3 in the brain and pituitary. Iodothyronine deoidinases also catalyze the inactivation of T4 and T3. The activities of deiodinases are under feedback control, mediated by T3, T4, and reverseT3 (rT3) an inactive deiodination product of T4 (Darras et al. 1999).

Mechanism of Uptake of Iodide into the Thyroid. Synthesis of T4 and T3 in the thyroid is dependent on delivery of iodide into the thyroid follicle where the iodination of thyroglobulin occurs in the first steps of hormone synthesis (Figure 3-5). Uptake of iodide into the thyroid is facilitated by a membrane carrier in the basolateral membrane of the thyroid follicle cell (Carrasco 1993; Levy et al. 1998a; Shen et al. 2001). The carrier, or NIS, catalyzes the simultaneous transfer of Na⁺ and Γ across the basolateral membrane (Chambard et al. 1983; Iff and Wilbrandt 1963; Nilsson et al. 1990). The NIS enables the follicle cell to achieve intracellular/extracellular concentration ratios of 10–50 for iodide (Andros and Wollman 1991; Bagchi and Fawcett 1973; Shimura et al. 1997; Vroye et al. 1998; Weiss et al. 1984b; Wolff 1964).

The NIS has been studied extensively in several *in vitro* preparations, including isolated plasma membrane vesicles of mammalian thyroid (O'Neill et al. 1987), FRTL-5 cells, a cell line derived from normal rat thyroid (Weiss et al. 1984b), *Xenopus lavis* oocytes transformed by intracellular injection of FRTL-5 RNA to express NIS (Eskandari et al. 1997), and other mammalian cells cultures transformed to express NIS (Levy et al. 1997; Nakamura et al. 1990; Smanik et al. 1996; Yoshida et al. 1997). Iodide transport by the NIS is inhibited by other anions, most notably, thiocyanate (SCN⁻) and perchlorate (ClO₄⁻) (Carrasco 1993; Wolff 1964). Thiocyanate is one of several anions other than I- that can be transported by the NIS, including SeCN⁻, NO₃⁻, ClO₃⁻, Br⁻, BF₄⁻, IO₄⁻, BrO₃⁻, ReO₄⁻, and TcO₄⁻ (Eskandari et al. 1997; Van Sande et al. 2003). Direct evidence for perchlorate transport by NIS (i.e., measurement of radioperchlorate flux through the NIS) is lacking; however, transport activity is likely given that the

NIS transports the structural analogs, perrhenate and pertechnetate. Perchlorate transport by NIS may be electroneutral, preventing its detection from measurements of ion currents or electrochemical gradients (Eskandari et al. 1997; Yoshida et al. 1997).

The NIS is expressed in a variety of other tissues, including breast tissue where it is thought to function in the transport of iodide into breast milk (Levy et al. 1997; Smanik et al. 1997; Spitzweg et al. 1998). In the rat, expression of the NIS, or a structurally similar membrane protein, increases during nursing and decreases after weaning (Levy et al. 1998a). In the mouse, expression of NIS in mammary tissue appears to be stimulated by prolactin (Perron et al. 2001; Rillema and Rowady 1997; Rillema et al. 2000).

Inhibition of Thyroid NIS and Thyroid Hormone Production by Perchlorate. Perchlorate inhibition of NIS can limit the availability of iodide needed for the production of T4 and T3 in the thyroid. The degree of inhibition required to actually impair the production of T4 and T3 to the extent that it effects circulating levels of T4 or T3 appears to vary with species. Rats appear to be more sensitive than humans (see Section 3.5.3). This difference is thought to derive from the rat thyroid having a smaller store of iodinated thyroglobulin that is more easily depleted when the availability of iodide is limited, and from a more rapid clearance of T4 from the rat circulation; the latter resulting from rats not having a high affinity binding protein for T4 in serum analogous to thyroid-binding globulin (THBG) in humans (Capen 1997). In humans, THBG functions as an important storage depot for circulating T4 and a buffer for homeostatic regulation of free T4 levels in serum (Robbins 2000). If the production of T4 and T3 is impaired sufficiently to deplete the thyroid of stored iodinated thyroglobulin, the thyroid cannot produce or secrete amounts of T4 and T3 needed to support physiological demands, circulating levels of T4 (fT4) and T3 decrease, and a state of thyroid hormone insufficiency ensues. A decrease in the levels of circulating thyroid hormones triggers HPT feedback control mechanisms that serve to adjust thyroidal iodide transport and hormone production in response to changes in circulating levels of T4, T3, and iodide. Major components of this mechanism include inhibition of the secretion of TRH from the hypothalamus, TRH-stimulated secretion of TSH from the pituitary, TSH-stimulated induction of thyroid follicle cell NIS (i.e., upregulation) and other thyroid cell proteins, increased capacity for transport of iodide into thyroid follicle cells, and increased synthesis and secretion of T4 and T3 from the thyroid. This system normally maintains circulating levels of T4 and T3 within narrow individual limits (Andersen et al. 2002). Doses of perchlorate that are sufficient to decrease circulating levels of thyroid hormones outside of these individual limits will result in increased secretion of TSH. Thus, an increase in the circulating levels of TSH is a sign that perchlorate has perturbed circulating levels of thyroid hormones. In humans, intraindividual variation in T4 levels is less than inter-individual variation, suggesting that the HPT feedback

mechanism can detect relatively small changes in thyroid hormone levels that are well within the range of variation expected in populations (Andersen et al. 2002).

Perchlorate-induced Hypothyroidism. Inhibition of thyroid iodide uptake can potentially deplete stores of T4 and T3 in the thyroid and lower serum T4 and T3 levels. Thus, perchlorate has the potential for producing hypothyroidism or for aggravating an ongoing hypothyroid condition. The term hypothyroidism refers to a state of suppressed production and/or secretion of thyroid hormones. The term clinical hypothyroidism refers to a condition in which the circulating levels of T4 and/or T3 are depressed below their normal ranges (usually accompanied with elevated serum TSH levels above the normal range) and in which there are clinical symptoms of thyroid hormone insufficiency (Ladenson 2000). Typical normal ranges for hormone levels are shown in Table 3-4. Subclinical hypothyroidism refers to a decrease in circulating T4 or T3 concentrations, usually accompanied by an increase in serum TSH concentration, within their respective normal ranges. An important question is whether small changes in circulating levels of thyroid hormones that trigger the HPT feedback mechanism, but do not fall outside of the normal population range, are detected as thyroid hormone insufficiency in tissues other than the hypothalamus or pituitary, including the embryo or fetal brain.

In humans, relatively large doses of perchlorate (600–900 mg/day, 8–13 mg/kg/day) are required to deplete thyroidal iodine stores sufficiently to decrease serum levels of T4 (Brabant et al. 1992; Bürgi et al. 1974). A 4-week oral exposure to 900 mg/day (approximately 13 mg/kg/day) did not produce clinical hypothyroidism in healthy adults (Brabant et al. 1992); however, a dosage considerably lower, 0.5 mg/kg/day for 14 days, produced a 70% inhibition of thyroid iodide uptake with no effects on the levels of circulating T4, T3, or TSH in serum, at least over the 14-day dosing period (Greer et al. 2002). In these short-term studies, it is possible that thyroid hormone production could have been suppressed by perchlorate inhibition of thyroid NIS without changing serum thyroid hormone levels. This could occur because the human adult thyroid contains a surplus of T4 to support normal levels of serum levels for several months (Greer et al. 2002). The ability of high dosages of perchlorate to lower T4 and T3 levels in serum is the basis for use of perchlorate in the pharmacological management of thyrotoxicosis, the clinical manifestation of abnormally elevated circulating levels of T4 and/or T3 (Soldin et al. 2001).

Perchlorate-induced Thyroid Enlargement and Cancer. Although there is no direct evidence of perchlorate causing cancer in humans, perchlorate has produced thyroid cell hyperplasia and papillary and/or follicular adenomas and/or carcinomas in rats and mice (see Section 3.2.2.7). Perchlorate itself

3. HEALTH EFFECTS

Table 3-4. Typical Reference Ranges for Serum Thyroid Hormones and TSH in **Humans**

	Reference range		
Hormone	Metric	SI unit	
Total T4	4–11 μg/dL	60–140 nM ^a	
Free T4	0.7–2.1 ng/dL	10–25 pM ^a	
Total T3	75–175 ng/dL	1.1–2.7 nM ^a	
Free T3	0.2–0.5 ng/dL	3–8 pM	
Reverse T3	15–45 ng/dL	0.2–0.7 nM	
TSH	0.3 – $4.0 \text{ mU/L}^{b,c}$	1–15 pM	

Source: Stockigt (2000) and Marcocci and Chiovato (2000)

SI = Systems Integration; T3 = 3,5,3'-triiodo-L-thyronine; T4 = 3,5,3',5'-tetraiodo-L-thyronine (thyroxine); TSH = thyroid stimulating hormone

^aChildren may have higher levels ^bAssumes a biologic potency of 7–15 mU/mg ^cHigher in neonates (de Zegher et al. 1994)

does not appear to be genotoxic (see Section 3.3). Production of thyroid tumors appears to be related to perchlorate-induced inhibition of thyroid iodide uptake and the resulting triggering of the HPT feedback control mechanism that elevates serum TSH levels. Persistent stimulation of the thyroid by TSH results in hypertrophy and hyperplasia of thyroid follicle cells, which are reflected in an increase in the size and weight of the thyroid (goiter). Tumors appear to be a progression of this hyperplasia. The mechanism by which gland enlargement leads to thyroid tumors is not completely understood. Thyroid gland proliferation may increase the fixation of mutations in the thyroid and promote the development of autonomous nodules, regions of thyroid follicle tissue that are less responsive or unresponsive to serum TSH concentrations (Corvilain et al. 2000; Fagin 2000). Consistent with the concept that TSH-stimulation and the resulting thyroid cell hypertrophy and hyperplasia are contributing factors to thyroid tumorigenesis are the observations that thyroid tumors can be produced in rats by a variety of different treatments that chronically elevate serum TSH levels, including maintaining the animals on a diet deficient in iodide, or by exposing the animals to chemical agents (e.g., thiouracil compounds, sulfonamides) that disrupt thyroid hormone production (Capen 1997).

Developmental Effects of Perchlorate. Thyroid hormones are essential for normal development of the nervous system, lung, skeletal muscle, and possibly other organ systems (Forhead et al. 2002; Hume et al. 2001; Porterfield and Hendrich 1993). The fetus is dependent on maternal thyroid hormones at least until the fetal thyroid begins to produce T4 and T3 (Zoeller and Crofton 2000). In humans, this occurs at approximately 16-20 weeks of gestation. Brain development begins in humans prior to the onset of fetal thyroid hormone production, with a major growth spurt occurring between 12 and 18 weeks of gestation with the beginning of neuron multiplication (Pintar 2000). This is followed by glial cell multiplication, myelination, and formation of dendritic extensions and synapses, which begin at approximately 18 weeks, reaching their peak near the end of gestation and continuing through postnatal years 1 and 2 (Boyages 2000; Fisher and Brown 2000; Oppenheimer and Schwartz 1997). Thyroid hormones are present in human amniotic fluid at 8 weeks of gestation prior to the onset of fetal thyroid hormone production (Contempre et al. 1993; Thorpe-Beeston et al. 1991). Thyroid hormone receptors are present and occupied by hormone at this time as well, suggesting that the fetus is capable of responding to maternal thyroid hormones (Bernal and Pekonen 1984; Ferreiro et al. 1988). The contribution of maternal thyroid hormones to the fetal thyroid hormone status is also evident from infants who have an inherited disorder that abolishes T4 production but are born, nevertheless, with normal serum thyroid hormone levels (i.e., euthyroid) and become hypothyroid after birth if not administered thyroid hormones within 2 weeks after birth (Larsen 1989; Vulsma et al. 1989). This suggests that, in the complete absence of fetal thyroid function, the maternal thyroid is able to maintain adequate levels of thyroid hormone in the fetus at late

term. However, athyrotic babies, although born euthyroid, show retarded skeletal maturation at birth, suggesting that fetal thyroid function during earlier phases of gestation may be necessary for normal skeletal development (Rovet et al. 1987; Wolter et al. 1979). Uncorrected maternal hypothyroidism, on the other hand, may result in impaired neurodevelopment of the fetus (Haddow et al. 1999; Pop et al. 1999; Soldin et al. 2001). In addition, as discussed by NAS (2005), preterm infants are more susceptible to thyroid hormone perturbations than term infants.

Studies in rats provide further support for the importance of maternal thyroid hormones in development. Both T4 and T3 are present in rat fetal tissues prior to the onset of hormone production by the fetal thyroid on approximately day 17 of gestation, and maternal hormones appear to make a significant contribution to hormone levels in the fetus in late gestation as well (Calvo et al. 1990; Escobar del Rey et al. 1986; Morreale de Escobar et al. 1990; Zoeller and Crofton 2000). Furthermore, thyroid hormoneresponsive genes that are important in early development of the brain are expressed in the rat fetus prior to fetal thyroid hormone production, and expression of these genes is sensitive to the maternal thyroid hormone status (Dowling and Zoeller 2000; Dowling et al. 2001). Disruption of the maternal thyroid hormone system of rats by removal of the maternal thyroid or by maternal iodide deficiency results in decreased levels of thyroid hormones in the fetus and congenital hypothyroidism (Escobar del Rey et al. 1986; Morreale de Escobar et al. 1985). These observations suggest an important role of maternal thyroid hormones in development of the rat fetus and that, by limiting the availability of thyroid hormones to the early fetus, suppression of maternal thyroid hormone production by perchlorate could translate into disruptions of fetal development. The availability of maternal T4 to the fetus appears to be particularly important for maintenance of T3 levels in the fetal rat brain. Treatment of pregnant rats with methimazole, an inhibitor of thyroid hormone synthesis, resulted in decreased levels of both T4 and T3 in fetal tissues, including fetal brain (Calvo et al. 1990). Maternal infusions of T4 restored brain T3 levels; however, maternal infusion of T3 had little restorative effect on brain T3 levels, although it was able to restore T3 levels in other fetal tissues. Studies in which radiolabelled T4 was administered to neonatal rats made hypothyroid by maternal or neonatal treatment with methimazole provide direct evidence for the enhanced production of brain T3 from T4 (Silva and Mathews 1984). These observations are consistent with an important role of local generation of T3 from T4 in the brain by the action of brain iodothyronine deiodinases in maintaining brain T3 levels (Darras et al. 1999; Zoeller and Crofton 2000). From a toxicological perspective, these observations also suggest that, at least in the rat, a decrease in maternal serum T4 levels, even in the absence of changes in maternal serum T3 levels may have adverse consequences on fetal brain development.

The above observations suggest that perchlorate could potentially disrupt fetal thyroid hormone status by three mechanisms. Perchlorate inhibition of maternal thyroid iodide uptake, and the resulting suppression in production and levels of maternal thyroid hormones, could limit the availability of thyroid hormones needed for normal fetal development prior to the onset of fetal thyroid hormone production if thyroid function in the mother is compromised. Perchlorate can also cross the placenta and may directly inhibit fetal thyroid iodide uptake and, secondarily, fetal thyroid hormone production. By inhibiting NIS in breast tissue, perchlorate may also limit the availability of iodide to nursing infants, who depend entirely on breast milk for the iodide needed to produce thyroid hormone (Agency for Toxic Substances and Disease Registry 2004). No information is available on the doses in humans that might decrease iodide uptake into breast milk. Radioiodine uptake into mammary milk was decreased in rats exposed to 1 or 10 mg/kg/day perchlorate in drinking water (Clewell et al. 2003b). Studies conducted in cows and goats have also shown that perchlorate can decrease radioiodine uptake into mammary milk (Howard et al. 1996).

Thyroid suppression at birth has been observed in infants born to mothers who received potassium perchlorate during pregnancy for treatment of hyperthyroidism (Crooks and Wayne 1960; Fisher et al. 1962). Direct evidence of maternal-fetal transfer of perchlorate and suppression of fetal thyroid iodide uptake and hormone production has been provided from studies of rats and guinea pigs (Clewell et al. 2003a; Postel 1957; Schröder-van der Eslst et al. 2001; York et al. 2001b). Several epidemiological studies have explored the strength of possible associations between perchlorate exposures and neonatal thyroid hormone status. Although some of these studies are suggestive of a possible association between perchlorate exposures and elevated serum TSH levels in infants, the findings of the currently available epidemiological literature, taken *in toto*, is inconclusive regarding effects of perchlorate on neonatal thyroid hormone status (Brechner et al. 2000; Crump et al. 2000; Lamm and Doemland 1999; Li et al. 2000a, 2000b; Schwartz 2001). Studies conducted in rats provide direct evidence that perchlorate exposures during pregnancy or lactation can disturb thyroid hormone status in the neonate (Brown-Grant 1966; Brown-Grant and Sherwood 1971; Clewell et al. 2003a, 2003b; Golstein et al. 1988; Lampe 1967; Mahle et al. 2003; York et al. 2001a, 2003, 2004). The mechanisms for these effects have not been elucidated.

3.5.3 Animal-to-Human Extrapolations

The ability of perchlorate to inhibit thyroid uptake of iodide in both humans (Bürgi et al. 1974; DeGroot and Buhler 1971; Faure and Dussault 1975; Greer et al. 2002; Lawrence et al. 2000, 2001; Stanbury and

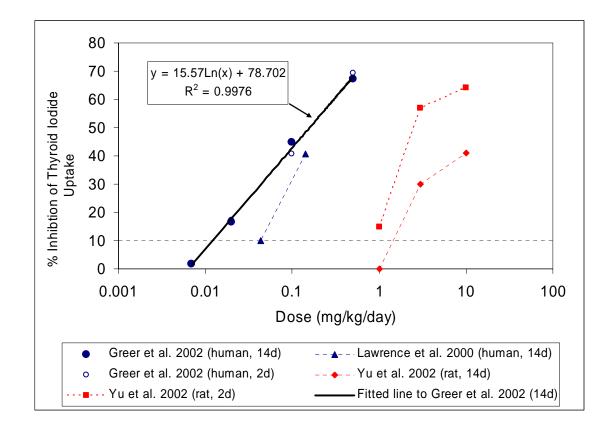
Wyngaarden 1952) and animals (Kapitola et al. 1971; Ortiz-Caro et al. 1983; Schonbaum et al. 1965; Wyngaarden et al. 1952) is well established. Based on this ability, potassium perchlorate was widely used for a time as treatment to restore normal thyroid activity in patients with hyperactive thyroids (e.g., Crooks and Wayne 1960; Morgans and Trotter 1960). Therapeutically effective doses in humans were in the range of 5–20 mg perchlorate/kg/day.

Although abundant evidence exists to show that perchlorate can inhibit thyroid iodide uptake in humans, direct evidence that perchlorate can disrupt thyroid hormone levels and produce changes in thyroid morphology (in the absence of underlying thyroid disorder, such as Graves' disease or other causes such as thyrotoxicosis) derives largely from animal studies. Changes in serum levels of thyroid hormones, indicative of suppressed hormone production, and thyroid hypertrophy, indicative of stimulation of the thyroid gland by TSH, have been shown to occur in rats and mice exposed to perchlorate by the oral route (see Section 3.2.2.2, Endocrine Effects). Evidence that perchlorate can produce thyroid tumors also derives from the results of studies conducted in mice and rats (see Section 3.2.2.7). In humans and other mammals, a limitation in the availability of iodide for thyroid hormone production, regardless of the cause of the limitation, triggers an HPT feedback mechanism, which serves to maintain serum hormones at sufficient levels to satisfy physiological requirements. Extrapolation of dose levels that disrupt thyroid hormone status in animals to pharmacodynamically equivalent doses in humans must take into account not only potential species differences in perchlorate biokinetics, but also potential species differences in the compensatory response to a limitation in iodide availability to the thyroid.

Few studies have been reported that allow direct comparisons of the dose-response relationships for the effects of perchlorate on thyroid hormone status in humans and experimental animals. However, these studies indicate that whereas human adults are more sensitive than rats to the inhibitory effects of oral dosages (mg/kg/day) of perchlorate on thyroid iodide uptake (Figure 3-7), rats are far more sensitive than humans to the acute effects of inhibition of iodide uptake on circulating thyroid hormone levels (see below). Inhibition of thyroid iodide uptake has been observed in healthy euthyroid adults who were exposed to dosages exceeding 0.007 mg/kg/day in drinking water (Greer et al. 2002; Lawrence et al. 2000). The dosage that produced a 50% inhibition of 24-hour thyroid iodide uptake was approximately 0.15 mg/kg/day when the exposure duration was either for 2 or 14 days (Figure 3-7, Greer et al. 2002). Substantially higher dosages were required to inhibit thyroid iodide uptake in rats; 1–3 mg/kg/day in drinking water for 14 days produced only a 3% inhibition (Yu et al. 2002, Figure 3-7). Rats also exhibited a pronounced attenuation of the inhibition in thyroid iodide uptake when the exposure duration

Figure 3-7. Comparison of Dose-Response Relationships for the Inhibitory Effect of Perchlorate on 24-hour Thyroid Iodide Uptake in Humans and Rats

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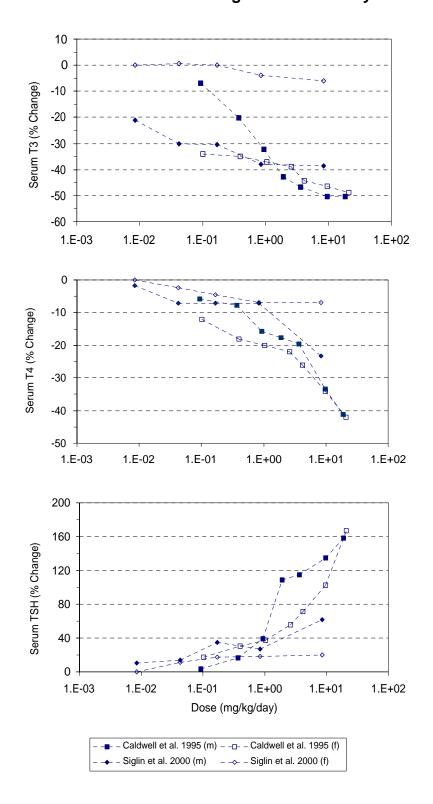
PERCHLORATES 100 3. HEALTH EFFECTS

was increased from 2 to 14 days; this was not observed in humans, at least in the dosage and duration ranges studied (Figure 3-7). The more pronounced attenuation of the response to perchlorate in the rat may reflect the triggering of HPT feedback control mechanisms and induction of NIS, which serve to regulate thyroid iodide transport and hormone production in response to a decrease in serum thyroid hormones and iodide levels. The involvement of the HPT control mechanism in the attenuation of the response to perchlorate in the rat is consistent with the observed dose-response relationship for changes in serum T3, T4, and TSH levels (Figure 3-8). Perchlorate dosages of 1-5 mg/kg/day for 14 days in drinking water depressed serum levels of T3 and T4 and increased levels of TSH at dosage levels that had minimal effects on thyroid iodide uptake (Caldwell et al. 1995; Siglin et al. 2000). By contrast, serum hormone levels were unchanged in human adults who received dosages of up to 0.5 mg/kg/day for the same duration and who exhibited as much as a 70% inhibition of thyroid iodide uptake (Greer et al. 2002; Lawrence et al. 2000). These observations suggest that dosages of perchlorate that inhibit thyroid iodide uptake must occur over a longer duration to produce effects on circulating levels of thyroid hormones in healthy, euthyroid adult humans than in healthy, euthyroid adult rats. This is thought to be related to a smaller and more rapid turnover of the hormone pool in the rat thyroid, and to a more rapid clearance of secreted hormone in the rat; that latter being, in part, related to the absence of thyroid binding globulin (TBG) in rats (Capen 1997; Greer et al. 2002).

Less is known about the relative sensitivities of humans and experimental animals to developmental effects of perchlorate. Outstanding uncertainties include potential differences in kinetics of maternal-fetal and maternal-infant transfer of perchlorate, as well as potential differences in the degree to which the fetus of the human, in comparison to experimental animals, is dependent on maternal thyroid hormone for development, particularly during the period of gestation prior to the onset of fetal hormone production.

NAS (2005) reviewed the human and animal data and concluded that the human data provided a more reliable point of departure for the risk assessment than the animal data. In agreement with the above discussion, NAS (2005) further noted that: "the rat is a good quantitative model for assessing inhibition of iodide uptake by the thyroid caused by perchlorate exposure, but it is only a good *qualitative* model for the effects of that inhibition."

Figure 3-8. Changes in Serum Thyroid Hormone Levels in Rats Exposed to Perchlorate in Drinking Water for 14 Days



3.6 TOXICITIES MEDIATED THROUGH THE NEUROENDOCRINE AXIS

Recently, attention has focused on the potential hazardous effects of certain chemicals on the endocrine system because of the ability of these chemicals to mimic or block endogenous hormones. Chemicals with this type of activity are most commonly referred to as endocrine disruptors. However, appropriate terminology to describe such effects remains controversial. The terminology endocrine disruptors, initially used by Colborn and Clement (1992), was also used in 1996 when Congress mandated the Environmental Protection Agency (EPA) to develop a screening program for "...certain substances [which] may have an effect produced by a naturally occurring estrogen, or other such endocrine effect[s]...". To meet this mandate, EPA convened a panel called the Endocrine Disruptors Screening and Testing Advisory Committee (EDSTAC), which in 1998 completed its deliberations and made recommendations to EPA concerning endocrine disruptors. In 1999, the National Academy of Sciences released a report that referred to these same types of chemicals as hormonally active agents. The terminology endocrine modulators has also been used to convey the fact that effects caused by such chemicals may not necessarily be adverse. Many scientists agree that chemicals with the ability to disrupt or modulate the endocrine system are a potential threat to the health of humans, aquatic animals, and wildlife. However, others think that endocrine-active chemicals do not pose a significant health risk, particularly in view of the fact that hormone mimics exist in the natural environment. Examples of natural hormone mimics are the isoflavinoid phytoestrogens (Adlercreutz 1995; Livingston 1978; Mayr et al. 1992). These chemicals are derived from plants and are similar in structure and action to endogenous estrogen. Although the public health significance and descriptive terminology of substances capable of affecting the endocrine system remains controversial, scientists agree that these chemicals may affect the synthesis, secretion, transport, binding, action, or elimination of natural hormones in the body responsible for maintaining homeostasis, reproduction, development, and/or behavior (EPA 1997). Stated differently, such compounds may cause toxicities that are mediated through the neuroendocrine axis. As a result, these chemicals may play a role in altering, for example, metabolic, sexual, immune, and neurobehavioral function. Such chemicals are also thought to be involved in inducing breast, testicular, and prostate cancers, as well as endometriosis (Berger 1994; Giwercman et al. 1993; Hoel et al. 1992).

Perchlorate can impair thyroid hormone production and, therefore, can be classified as an *endocrine disruptor*. As discussed in Sections 3.2.2.2 (Systemic-Endocrine Effects) and 3.5.2 (Mechanisms of Toxicity), at sufficiently high dosages, perchlorate can limit the availability of iodide needed for the production of the hormones in the thyroid and can depress serum levels of the thyroid hormones, T4 and

T3. The latter effect triggers HPT feedback control mechanisms to produce TSH, which stimulates growth of the thyroid and induces NIS, the primary mechanism by which iodide enters thyroid follicle cells from the blood and the first step in the uptake of iodide into the thyroid and formation of thyroid hormones. Thus, perchlorate has the potential for producing all of the adverse consequences of hypothyroidism including impairments in the development of the nervous systems and other organ systems, thyroid gland enlargement, and follicular cell hyperplasia and neoplasia.

3.7 CHILDREN'S SUSCEPTIBILITY

This section discusses potential health effects from exposures during the period from conception to maturity at 18 years of age in humans, when all biological systems will have fully developed. Potential effects on offspring resulting from exposures of parental germ cells are considered, as well as any indirect effects on the fetus and neonate resulting from maternal exposure during gestation and lactation. Relevant animal and *in vitro* models are also discussed.

Children are not small adults. They differ from adults in their exposures and may differ in their susceptibility to hazardous chemicals. Children's unique physiology and behavior can influence the extent of their exposure. Exposures of children are discussed in Section 6.6, Exposures of Children.

Children sometimes differ from adults in their susceptibility to hazardous chemicals, but whether there is a difference depends on the chemical (Guzelian et al. 1992; NRC 1993). Children may be more or less susceptible than adults to health effects, and the relationship may change with developmental age (Guzelian et al. 1992; NRC 1993). Vulnerability often depends on developmental stage. There are critical periods of structural and functional development during both prenatal and postnatal life and a particular structure or function will be most sensitive to disruption during its critical period(s). Damage may not be evident until a later stage of development. There are often differences in pharmacokinetics and metabolism between children and adults. For example, absorption may be different in neonates because of the immaturity of their gastrointestinal tract and their larger skin surface area in proportion to body weight (Morselli et al. 1980; NRC 1993); the gastrointestinal absorption of lead is greatest in infants and young children (Ziegler et al. 1978). Distribution of xenobiotics may be different; for example, infants have a larger proportion of their bodies as extracellular water and their brains and livers are proportionately larger (Altman and Dittmer 1974; Fomon 1966; Fomon et al. 1982; Owen and Brozek 1966; Widdowson and Dickerson 1964). The infant also has an immature blood-brain barrier (Adinolfi 1985; Johanson 1980) and probably an immature blood-testis barrier (Setchell and Waites 1975). Many

xenobiotic metabolizing enzymes have distinctive developmental patterns. At various stages of growth and development, levels of particular enzymes may be higher or lower than those of adults, and sometimes unique enzymes may exist at particular developmental stages (Komori et al. 1990; Leeder and Kearns 1997; NRC 1993; Vieira et al. 1996). Whether differences in xenobiotic metabolism make the child more or less susceptible also depends on whether the relevant enzymes are involved in activation of the parent compound to its toxic form or in detoxification. There may also be differences in excretion, particularly in newborns who all have a low glomerular filtration rate and have not developed efficient tubular secretion and resorption capacities (Altman and Dittmer 1974; NRC 1993; West et al. 1948). Children and adults may differ in their capacity to repair damage from chemical insults. Children also have a longer remaining lifetime in which to express damage from chemicals; this potential is particularly relevant to cancer.

Certain characteristics of the developing human may increase exposure or susceptibility, whereas others may decrease susceptibility to the same chemical. For example, although infants breathe more air per kilogram of body weight than adults breathe, this difference might be somewhat counterbalanced by their alveoli being less developed, which results in a disproportionately smaller surface area for alveolar absorption (NRC 1993).

Fetuses, infants and children may be especially susceptible to the thyroid effects of perchlorate. Thyroid hormones regulate cell proliferation, migration, and differentiation during development, and maintenance of normal levels is essential to normal growth and development. Disruption of circulating hormone levels can have markedly different effects, depending on the stage of development. Effects can include mental retardation, impaired motor skills, and hearing and speech impediments (Boyages 2000; Fisher and Brown 2000). Several factors may contribute to a high vulnerability of the fetus and neonate to perchlorate. In addition, as discussed by NAS (2005), preterm infants are more sensitive to thyroid hormone perturbations than term infants.

Thus far, there is no conclusive evidence that exposure to perchlorate produces developmental effects in humans, although two studies of newborns in Arizona and California reported that neonates from women whose drinking water contained perchlorate had higher TSH values than those from women with no exposure to perchlorate (Brechner et al. 2000; Schwartz 2001). However, the methods used in the two latter studies have been questioned. Other similar studies in the United States have found no significant associations between maternal exposure to perchlorate via the drinking water and T4 levels (Li et al. 2000a), TSH levels (Li et al. 2000b), and incidence of congenital hypothyroidism (Kelsch et al. 2003;

Lamm and Doemland 1999). Two studies of Chilean neonates whose mothers may have been chronically exposed to up to 100–120 μg/L (ppb) of perchlorate in the drinking water found no evidence of adverse thyroid effects among the neonates (Crump et al. 2000; Téllez et al. 2005). Studies in experimental animals have shown that exposure of the mother to perchlorate during gestation, or even during lactation, can lead to reduced thyroid hormone levels and associated thyroid effects in the offspring (Brown-Grant and Sherwood 1971; Golstein et al. 1988; Lampe 1967; Postel 1957; York et al. 2001a, 2003, 2004). Evaluation of a series of neurobehavioral parameters in rat pups exposed to perchlorate *in utero* (maternal exposure up to 8.5 mg perchlorate/kg/day) and through maternal milk revealed no significant treatment-related effects (Bekkedal et al 2000; York et al. 2004). Microscopic examination of the brain from 12-day-old pups showed a significant increase in the thickness of the corpus callosum from females in the highest dose group, 8.5 mg/kg/day (York et al. 2004). The toxicological significance of this finding is controversial and its biological significance has been questioned (NAS 2005). High-dose male offspring sacrificed at about 80 days of age had a significant increase in brain weight and in the weight of the prefrontal cortex and corpus callosum (York et al 2004). Exposure to perchlorate has not caused teratogenic effects in animals.

Perchlorate has been shown to cross the placenta of rats (Clewell et al. 2003a; Schröeder-van der Eslst et al. 2001). Thus, in addition to the potential for perchlorate to exert effects on fetal development by depressing levels of maternal thyroid hormones, perchlorate may exert direct effects on the fetal thyroid. Thyroid suppression at birth has been observed in infants born to mothers who received potassium perchlorate during pregnancy for treatment of hyperthyroidism (Crooks and Wayne 1960; Fisher et al. 1962). Direct evidence of maternal-fetal transfer of perchlorate and suppression of fetal thyroid hormone production has been provided from studies of rats and guinea pigs (Postel 1957; York et al. 2001b; Yu et al. 2002).

Studies conducted in experimental animals have shown that perchlorate enters mammary milk (Clewell et al. 2003b). Perchlorate has also been detected in human breast milk at a mean concentration of 10 ppb (Kirk et al. 2005). Whereas, this indicates that nursing infants may be exposed to perchlorate in breast milk, whether the amount of perchlorate in the breast milk is great enough to affect thyroid function of the infant has not been demonstrated.

Models of the biokinetics of perchlorate in adult humans and rats have been developed (Fisher et al. 2000; Merrill et al. 2003, 2005). An adult rat model has been extended to include pregnancy and maternal-fetal

transfer of perchlorate, and lactation and maternal-pup perchlorate transfer through milk (Clewell et al. 2003a, 2003b).

3.8 BIOMARKERS OF EXPOSURE AND EFFECT

Biomarkers are broadly defined as indicators signaling events in biologic systems or samples. They have been classified as markers of exposure, markers of effect, and markers of susceptibility (NAS/NRC 1989).

Due to a nascent understanding of the use and interpretation of biomarkers, implementation of biomarkers as tools of exposure in the general population is very limited. A biomarker of exposure is a xenobiotic substance or its metabolite(s) or the product of an interaction between a xenobiotic agent and some target molecule(s) or cell(s) that is measured within a compartment of an organism (NAS/NRC 1989). The preferred biomarkers of exposure are generally the substance itself or substance-specific metabolites in readily obtainable body fluid(s) or excreta. However, several factors can confound the use and interpretation of biomarkers of exposure. The body burden of a substance may be the result of exposures from more than one source. The substance being measured may be a metabolite of another xenobiotic substance (e.g., high urinary levels of phenol can result from exposure to several different aromatic compounds). Depending on the properties of the substance (e.g., biologic half-life) and environmental conditions (e.g., duration and route of exposure), the substance and all of its metabolites may have left the body by the time samples can be taken. It may be difficult to identify individuals exposed to hazardous substances that are commonly found in body tissues and fluids (e.g., essential mineral nutrients such as copper, zinc, and selenium). Biomarkers of exposure to perchlorates are discussed in Section 3.8.1.

Biomarkers of effect are defined as any measurable biochemical, physiologic, or other alteration within an organism that, depending on magnitude, can be recognized as an established or potential health impairment or disease (NAS/NRC 1989). This definition encompasses biochemical or cellular signals of tissue dysfunction (e.g., increased liver enzyme activity or pathologic changes in female genital epithelial cells), as well as physiologic signs of dysfunction such as increased blood pressure or decreased lung capacity. Note that these markers are not often substance specific. They also may not be directly adverse, but can indicate potential health impairment (e.g., DNA adducts). Biomarkers of effects caused by perchlorates are discussed in Section 3.8.2.

A biomarker of susceptibility is an indicator of an inherent or acquired limitation of an organism's ability to respond to the challenge of exposure to a specific xenobiotic substance. It can be an intrinsic genetic or other characteristic or a preexisting disease that results in an increase in absorbed dose, a decrease in the biologically effective dose, or a target tissue response. If biomarkers of susceptibility exist, they are discussed in Section 3.10 "Populations that are Unusually Susceptible."

3.8.1 Biomarkers Used to Identify or Quantify Exposure to Perchlorates

Studies in workers exposed to perchlorate (Lamm et al. 1999) and in volunteers who ingested daily doses of perchlorate for 14 days (Lawrence et al. 2000) indicate that perchlorate is rapidly eliminated unchanged in the urine (see Section 3.4.4). Urine, therefore, is a convenient testing medium for perchlorate. However, the excretion of perchlorate is so rapid that an acute exposure might be detectable for only a few days after exposure. Both occupational studies and studies with volunteers have estimated an elimination half-life for perchlorate of approximately 8–12 hours (Greer et al. 2002; Lamm et al. 1999; Lawrence et al. 2000). The methods used for measuring perchlorate in urine have not been standardized (see also Section 7.1).

Using a highly selective analytical method of coupled ion chromatography and electrospray tandem mass spectrometry, Valentín-Blasini et al. (2005) found an association between urinary levels of perchlorate with the concentrations of perchlorate in drinking water. In a population with no known perchlorate drinking water contamination, concentrations of perchlorate adjusted for urinary creatinine showed a median level of 7.8 µg perchlorate/g creatinine, with a range of 1.0–35 µg perchlorate/g creatinine. (The authors speculated that this population may have had exposure to perchlorate via contaminated food and/or tobacco.) When the urine samples of the pregnant women in three cities in the study in Chile (Crump et al. 2000) were analyzed by this method, the women in the city with a drinking water concentration of perchlorate of about 0.4 ng/mL had a median urinary concentration of 21 µg perchlorate/g creatinine, those with a drinking water concentration of about 5.8 ng/mL had a median urinary concentration of 37 µg perchlorate/g creatinine, and those with a drinking water concentration of about 114 ng/mL had a median urinary concentration of 120 µg perchlorate/g creatinine (Valentín-Blasini et al. 2005).

Perchlorate has also been detected in human breast milk, but the concentrations were not well-correlated with the water consumed by the lactating women. As perchlorate is fairly rapidly cleared when exposure

ceases, the presence of perchlorate in breast milk may be highly variable with time and recent dietary history (Kirk et al. 2005).

Other potential biomarkers of exposure to perchlorate relate to their effect on the thyroid gland. As described in Section 3.5.2, Mechanisms of Toxicity, perchlorate blocks uptake of iodide into the thyroid, leading to an increase in the serum level of free iodine (i.e., not bound to T4), which is then excreted in urine. No study has developed a correlation between exposure to particular dose levels of perchlorate and specific relative increases of free iodine in serum or urine. In serum, the normal level of free iodine ranges from 1.0 to 5.2 mg/L and the level of protein-bound iodine ranges from 32 to 72 mg/L. Iodine is excreted in urine at a rate that is nearly equal to the rate of intake, or approximately 100–200 mg/24 hours (Agency for Toxic Substances and Disease Registry 2004). Perchlorate also produces a decrease in the levels of T3 and T4 in serum, while increasing the serum level of TSH. Again, specific correlations between levels and duration of exposure to perchlorate and alterations in serum levels of T3, T4, or TSH have not been developed. Furthermore, these potential biomarkers are not specific to perchlorate; other antithyroid agents, such as carbimazole, can have similar effects.

3.8.2 Biomarkers Used to Characterize Effects Caused by Perchlorates

The thyroid is the critical target for perchlorate. Perchlorate blocks uptake of iodide into the thyroid, leading to an increase in the serum level of free iodine (i.e., not bound to T4), which is then excreted in urine. If the dosage is sufficient to limit the availability of iodide for the production of thyroid hormones, then perchlorate can also produce a decrease in the levels of T3 and T4 in serum, while increasing the serum level of TSH. Therefore, levels of iodide in serum or urine, and levels of T3, T4, and TSH in serum, can all be considered to be biomarkers of effect for perchlorate. It should be noted that none of these biomarkers is specific to perchlorate; other antithyroid agents such as carbimazole can have similar effects. Although the specific amount of change in these biomarkers associated with a demonstrably adverse effect has not been established, changes in these parameters can be considered to indicate potential impairment of health. Typical normal ranges for hormone levels are shown in Table 3-4.

3.9 INTERACTIONS WITH OTHER CHEMICALS

No studies were located regarding health effects in humans or animals exposed to perchlorate in combination with other chemicals that are likely to be found with perchlorate in the environment, in the workplace, or at hazardous waste sites. However, Tonacchera et al. (2004) investigated the simultaneous joint effects of perchlorate and other competitive inhibitors of iodide uptake (thiocyanate, nitrate, and non-radioactive iodide) in inhibiting RAIU in an *in vitro* test system in which human NIS was stably transfected into a Chinese hamster ovary (CHO) cell line. The relative potency of perchlorate to inhibit 125I⁻ uptake at the NIS was 15, 30, and 240 times that of thiocyanate, non-radioactive iodide, and nitrate, respectively, on a molar concentration basis. The results showed that the anions interact in a simple additive fashion and that the concentration response for each inhibitor was indistinguishable from that of each of the others after adjusting for differences in inhibition potencies.

3.10 POPULATIONS THAT ARE UNUSUALLY SUSCEPTIBLE

A susceptible population will exhibit a different or enhanced response to perchlorates than will most persons exposed to the same level of perchlorates in the environment. Reasons may include genetic makeup, age, health and nutritional status, and exposure to other toxic substances (e.g., cigarette smoke). These parameters result in reduced excretion of perchlorates, or compromised function of organs affected by perchlorates. Populations who are at greater risk due to their unusually high exposure to perchlorates are discussed in Section 6.7, Populations with Potentially High Exposures.

As discussed in NAS (2005), fetuses and preterm newborns constitute the most sensitive populations, although infants and developing children are also considered sensitive populations. The expected high sensitivity of developing organisms is due to the important role played by thyroid hormones during development (see Section 3.5.2). Perchlorate may reduce the level of circulating thyroid hormones in the blood, and low thyroid hormone levels during embryonic or fetal development can lead to effects such as mental retardation, impaired motor skills, and hearing and speech impediments (Haddow et al. 1999; Pop et al. 1999; Soldin et al. 2001). Hypothyroid pregnant women also constitute a susceptible group who may also put the fetus at higher risk if maternal hypothyroidism is present before the onset of fetal thyroid function at 10–12 weeks of gestation (Zoeller and Crofton 2000).

People with reduced thyroid activity from other causes may also be an unusually susceptible population. This includes people living in endemic goiter areas with low iodine intake, people with exposure to other anti-thyroid drugs (e.g., lithium) (Green 1996; Spaulding et al. 1972), and people with Hashimoto's disease (autoimmune hypothyroidism [Weetman 2000]) or other diseases that reduce thyroid hormone levels. Exposure to perchlorate may produce additional deficiencies in these people beyond those due to their pre-existing conditions, potentially increasing the severity of their conditions.

3.11 METHODS FOR REDUCING TOXIC EFFECTS

This section will describe clinical practice and research concerning methods for reducing toxic effects of exposure to perchlorates. However, because some of the treatments discussed may be experimental and unproven, this section should not be used as a guide for treatment of exposures to perchlorates. When specific exposures have occurred, poison control centers and medical toxicologists should be consulted for medical advice. No texts were located that provided specific information about treatment following exposures to perchlorate.

3.11.1 Reducing Peak Absorption Following Exposure

There are no established methods for managing initial exposure to perchlorate or for reducing peak absorption.

Since perchlorate is readily excreted in the urine, it is reasonable to assume that increasing the water uptake would help the body eliminate perchlorate. No studies have investigated this issue.

3.11.2 Reducing Body Burden

There are no published studies on reducing the body burden of perchlorate. Where perchlorate, used clinically as an antithyroid drug, has caused side-effects, treatments have relied on natural elimination by excretion in the urine.

3.11.3 Interfering with the Mechanism of Action for Toxic Effects

There are no published studies on the treatment of perchlorate exposure by interfering with its mechanism of toxicity.

3.12 ADEQUACY OF THE DATABASE

Section 104(i)(5) of CERCLA, as amended, directs the Administrator of ATSDR (in consultation with the Administrator of EPA and agencies and programs of the Public Health Service) to assess whether adequate information on the health effects of perchlorates is available. Where adequate information is not available, ATSDR, in conjunction with the National Toxicology Program (NTP), is required to assure the initiation of a program of research designed to determine the health effects (and techniques for developing methods to determine such health effects) of perchlorates.

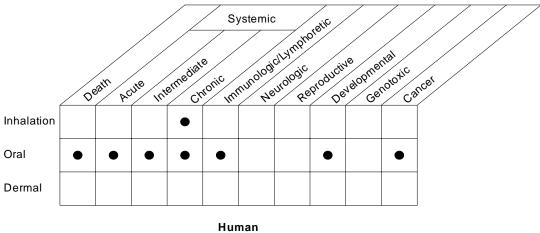
The following categories of possible data needs have been identified by a joint team of scientists from ATSDR, NTP, and EPA. They are defined as substance-specific informational needs that if met would reduce the uncertainties of human health assessment. This definition should not be interpreted to mean that all data needs discussed in this section must be filled. In the future, the identified data needs will be evaluated and prioritized, and a substance-specific research agenda will be proposed.

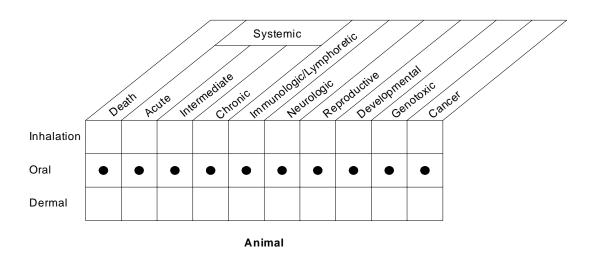
3.12.1 Existing Information on Health Effects of Perchlorates

The existing data on health effects of inhalation, oral, and dermal exposure of humans and animals to perchlorates are summarized in Figure 3-9. The purpose of this figure is to illustrate the existing information concerning the health effects of perchlorates. Each dot in the figure indicates that one or more studies provide information associated with that particular effect. The dot does not necessarily imply anything about the quality of the study or studies, nor should missing information in this figure be interpreted as a "data need". A data need, as defined in ATSDR's Decision Guide for Identifying Substance-Specific Data Needs Related to Toxicological Profiles (Agency for Toxic Substances and Disease Registry 1989), is substance-specific information necessary to conduct comprehensive public health assessments. Generally, ATSDR defines a data gap more broadly as any substance-specific information missing from the scientific literature.

Human studies on perchlorate include cross-sectional epidemiology studies of perchlorate workers with inhalation exposure, short-term oral experimental studies on healthy and hyperthyroid subjects, general population studies of adults, school-age children and neonates, and case reports of hyperthyroid patients with intermediate- or chronic-duration oral treatment with perchlorate.

Figure 3-9. Existing Information on Health Effects of Perchlorate





Existing Studies

Animal studies for perchlorate are available only by the oral route. The available studies have included investigation of systemic effects by acute, intermediate, and chronic exposure, as well as immunological, reproductive, and developmental effects, lethality, and cancer. Limited information is available regarding effects of perchlorate on the nervous system in adult animals. No experimental studies have been conducted that examine the interactions of perchlorate exposure and dietary iodine levels.

3.12.2 Identification of Data Needs

Acute-Duration Exposure. Acute-duration studies are available for healthy humans (Bürgi et al. 1974; DeGroot and Buhler 1971; Faure and Dussault 1975; Greer et al. 2002; Lawrence et al. 2000, 2001) and animals (Arieli and Chinet 1985; BRT 2000; Caldwell et al. 1995; DoD 1999; Kapitola et al. 1971; Mannisto et al. 1979; Matsuzaki and Suzuki 1981; Schonbaum et al. 1965; Siglin et al. 2000; Spreca and Musey 1974) orally exposed to perchlorate. These studies suggest that the thyroid is the main target for acute exposure to perchlorate. The study by Greer et al. (2002) identified a NOEL for radioactive iodine uptake by the thyroid of 0.0007 mg/kg/day. NAS (2005) recently completed an evaluation of the literature on perchlorate and derived an RfD of 0.007 mg/kg/day based on the findings of Greer et al. (2002). ATSDR has adopted the NAS (2005) chronic RfD for the chronic oral MRL. Normally, ATSDR would derive an acute MRL based on a 14-day study. It is not clear, however, if adopting the NAS chronic RfD for the acute (and the intermediate) MRL would be appropriate; ATSDR is soliciting public comments to this effect. The highest dose level tested, 0.5 mg/kg/day, produced an inhibition of iodine uptake of approximately 69%. Conduction of acute studies by the inhalation or dermal routes does not seem warranted since these routes of exposure do not play a significant role in environmental exposures to perchlorate. In its review of perchlorate toxicity, NAS (2005) notes that further studies of perchlorate in rats would be of limited utility for clarifying the health effects of perchlorate in humans. Instead, NAS recommends conducting a series of in vitro studies using human tissues and animal studies to determine the role of NIS in placental iodide transport, the susceptibility of breast NIS to perchlorate inhibition, the role of iodide status in these effects, and the effects of perchlorate on development independently of effects on iodide transport. NAS (2005) further notes studies on the effects of perchlorate on other tissues that contain NIS could be conducted.

Intermediate-Duration. Intermediate-duration studies are available for humans (Brabant et al. 1992) and animals (Bekkedal et al. 2000a; BRT 2000; DoD 1999; Eskin et al. 1975; Gauss 1972; Hiasa et al. 1987; Logonder-Mlinsek et al. 1985; MacDermott 1992; Ortiz-Caro et al. 1983; Pajer and Kalisnik 1991;

Postel 1957; Sangan and Motlag 1986, 1987; Selivanova and Vorobieva 1969; Shevtsova et al. 1994; Siglin et al. 2000; Tarin-Remohi and Jolin 1972; Vijayalakshmi and Motlag 1989a, 1989b, 1990, 1992; York et al. 2001a, 2001b, 2003, 2004) orally exposed to perchlorates. Brabant et al. (1992) studied the effect of administration of perchlorate for 4 weeks on thyroid hormone levels in healthy volunteers and their findings suggested that a period of iodide deficiency may increase the sensitivity of the thyroid to TSH. The studies in animals provided information on systemic, immunologic, reproductive, and developmental effects of perchlorate. The thyroid gland was shown to be the most sensitive target. Almost all of the earlier studies used only one dose level, usually quite high. Studies conducted in the past few years have used much lower doses, which allow the construction of dose-response relationships and the identification of NOAELs and LOAELs. In two of these studies (Siglin et al. 2000; York et al. 2003), ATSDR identified LOAELs of 0.009 mg/kg/day. Additional studies in animals would be useful to clarify some controversial findings of the studies recently available. These findings concern effects of perchlorate on neurobehavioral effects and brain morphometry in young animals exposed in utero. As indicated above, additional studies in rats for the purpose of clarifying the health effects of perchlorate in humans seem unnecessary given the difference in the manner rats and humans handle the inhibition of iodide uptake by the thyroid. As mentioned above, studies by inhalation and dermal exposure do not seem necessary at this time since the most relevant route of exposure for perchlorate is the oral route, specifically drinking water.

Chronic-Duration Exposure and Cancer. Humans with chronic inhalation exposure to perchlorate at work were the subject of cross-sectional epidemiology studies (Braverman et al. 2005; Gibbs et al. 1998; Lamm et al. 1999). These studies found no significant effects associated with perchlorate exposure. Cancer has not been reported in humans with exposure to perchlorate. A study by Li et al. (2001) found no significant association between perchlorate in drinking water and the prevalence of thyroid diseases or thyroid cancer residents in Nevada counties. However, no information was available on age, gender, ethnicity, iodine intake, and other risk factors. An additional study of cancer among residents of San Bernardino County, California, was limited by mixed exposures and lack of adjustment for potential confounding variables (Morgan and Cassidy 2002). A few chronic-duration studies of toxicity and carcinogenicity are available in animals exposed to perchlorate orally (Kessler and Kruskemper 1966; Toro Guillen 1991). A few intermediate-duration carcinogenicity studies by the oral route are also available (Florencio Vicente 1990; Gauss 1972; Pajer and Kalisnik 1991). The data from these studies are limited for assessment of toxicity or carcinogenicity, however, because only a high dose level was used. Still, it was evident that the main target of toxicity was the thyroid gland, as thyroid adenomas and/or carcinomas were observed in the animals. It is unclear whether additional studies would

provide new key information. As noted above, ATSDR has adopted the NAS (2005) RfD for the chronic oral MRL. NAS (2005) recommended a clinical study designed to provide information on the potential chronic effects of low-dose perchlorate exposure on thyroid function, with a special focus on the ability and mechanisms of thyroid compensation. If for ethical or other reasons it is not possible to conduct studies in humans, NAS (2005) suggested that chronic studies in nonhuman primates could provide useful information.

Genotoxicity. No genotoxicity studies were located for perchlorate in humans, but two studies were located in animals *in vivo*. Siglin et al. (2000) found no evidence of bone marrow micronucleus formation in male and female rats exposed to perchlorate in the drinking water for 90 days. Similar negative results were reported by Zeiger et al. (1998b) in mice treated with perchlorate intraperitoneally for 3 days. *In vitro* studies were limited to a test for SOS-inducing activity in *S. typhimurium* (Nakamura and Kosaka 1989), a test for production of DNA-protein cross links in cultured human lymphocytes (Costa et al. 1996), tests for mutagenicity in various *Salmonella* strains (Zeiger et al. 1998a), and a mouse lymphoma assay (San and Clarke 1999). Perchlorate gave negative results in all tests. Additional testing does not seem necessary at this time.

Reproductive Toxicity. No data on reproductive effects of perchlorate in humans were located. Earlier reproductive toxicity studies available for perchlorate in animals (Brown-Grant 1966; Brown-Grant and Sherwood 1971) were of limited utility for assessing reproductive toxicity because they included only brief exposure of females during gestation, limited investigation of reproductive end points, and single dose levels. A two-generation reproduction study was recently published (York et al. 2001b). In that study, exposure of rats to up to 25.5 mg perchlorate/kg/day did not affect any reproductive index. In addition, a 90-day drinking water in rats did not find gross or microscopic alterations in the testes, prostate, epididymis, uterus, ovaries, or mammary glands (Siglin et al. 2000). That study also reported no significant effects on sperm motility, concentration, count, or morphology. Further studies do not appear to be necessary at this time.

Developmental Toxicity. There are several oral studies available that evaluated the effects of perchlorate on thyroid parameters in human newborns (Brechner et al. 2000; Crump et al. 2000; Kelsh et al. 2003; Lamm and Doemland 1999; Li et al. 2000a, 2000b; Schwartz 2001; Téllez et al. 2005). Crump et al. (2000) also examined the effects of perchlorate on thyroid function in school-age children. For the most part, no significant alterations were reported, although Brechner et al. (2000) and Schwartz (2001) reported an association between high levels of perchlorate in the drinking water and elevated serum levels

of TSH, but the methods used in the two latter studies have been criticized. Developmental toxicity studies are available for perchlorate in animals (Bekkedal et al. 2000; Brown-Grant and Sherwood 1971; Golstein et al. 1988; Lampe 1967; Mahle et al. 2003; Postel 1957; York et al. 2001a, 2001b, 2003, 2004). Although the earlier studies were of limited utility because standard developmental toxicity end points were not monitored and only single high dose levels were administered, the studies conducted in the past few years have been able to establish dose-response relationships using relatively low exposure levels of perchlorate. Exposure of pregnant rats to perchlorate has resulted in thyroid alterations in the pups at maternal doses as low as 0.009 mg/kg/day (York et al. 2003). In addition to evaluating thyroid effects in the offspring, some recent studies have conducted neurobehavioral testing in the offspring at various ages and have also conducted histological and morphometric evaluations of pups' brains (Bekkedal et al. 2000; York et al. 2003, 2004). However, these studies have not evaluated known thyroid hormone-responsive end points in brain; for example, expression of genes that are known to respond to thyroid hormone or maturation specific brain structures (i.e., Purkinje cells) that respond to thyroid hormone (Porterfield and Hendrich 1993; Zoeller and Crofton 2000). Furthermore, there is no evidence that the linear measures of specific brain areas that have been evaluated in animal studies are responsive to changes in circulating levels of thyroid hormone. Studies directed at characterizing the reaction of thyroid hormone responsive end points in brain to small changes in thyroid hormone have not been conducted. NAS (2005) noted that studies of pregnant monkeys could provide useful information on the effects of perchlorate on fetal and neonatal development. The neurodevelopmental progress of children presumed to have been exposed to perchlorate in utero could be followed in longitudinal studies in search of possible long-term effects.

Immunotoxicity. Immune system and lymphoreticular effects due to perchlorate have not been systematically studied in healthy humans. Lymphoreticular effects were reported in one case series of human thyrotoxicosis patients treated with potassium perchlorate (Morgans and Trotter 1960). Immune effects in animals treated with very high doses of perchlorate (300–2,600 mg/kg/day) were reported as increases in the number and degranulation of mast cells in the thyroid and other tissues (Logonder-Mlinsek et al. 1985; Spreca and Musy 1974). More recent acute- and intermediate-duration studies assessed indices of humoral- and cell-mediated immunocompetence in mice (0.1–50 mg/kg/day), but there were deficiencies in the studies (BRT 2000; DoD 1999). It would be helpful if some other end points were studied, such as the determination of whether perchlorate increases the sensitizing response to other chemicals or whether perchlorate is a sensitizer itself.

Neurotoxicity. No data were located regarding neurological effects in humans or animals exposed to perchlorate. Neither 14-day nor 90-day studies in animals observed any signs indicative of neurotoxicant

in adult animals. However, since thyroid hormone insufficiency is known to affect brain function in adult humans, and perchlorate can produce decreased circulating levels of thyroid hormone, it is likely that perchlorate can, at some dose, impair brain function in adults. The dose-response relationship for these effects has not been characterized. As mentioned above under Developmental Effects, some studies in rats have suggested that exposure to perchlorate during pregnancy can cause neurodevelopmental alterations in the offspring (York et al. 2004). The biological significance of some of the neurodevelopmental alterations, particularly the changes in morphometry of some brain regions (York et al. 2004), has been questioned (NAS 2005).

Epidemiological and Human Dosimetry Studies. Information on effects of exposure to perchlorate in healthy humans is derived from occupational studies (Braverman et al. 2005; Gibbs et al. 1998; Lamm et al. 1999), studies of the general population, including adults, children, and neonates (Brechner et al. 2000; Chang et al. 2003; Crump et al. 2000; Kelsh et al. 2003; Li et al. 2000a, 2000b, 2001; Schwartz 2001; Téllez et al. 2005) and controlled exposures in volunteers (Greer et al. 2002; Lawrence et al. 2000, 2001). All of these studies provide information on the effects of perchlorate on thyroid parameters, but a few of them provide additional information on hematological, hepatic, and renal effects. Although it is well known that the thyroid is the target organ for perchlorate, the existing studies of the general population have had design limitations that preclude establishing with confidence the levels of environmental exposure that may induce clinically significant alterations in thyroid function and, therefore, represent a health risk. Well-designed epidemiological studies of environmentally exposed populations could provide valuable information and decrease the uncertainty of using data collected in acute studies in volunteers to establish long-term safe exposure levels. NAS (2005) identified pregnant women, their fetuses, and newborns as populations at greatest risk of the adverse effects of iodide deficiency and recommended that epidemiologic research should focus on assessing possible health effects of perchlorate exposure in these populations. These studies should use direct methods of exposure to perchlorate in individuals and methods more suitable for examining potentially causal associations. NAS (2005) further suggested that future research could be organized into additional analysis of existing data, new studies of health effects in selected populations, and monitoring of the frequencies of specific conditions in communities affected by the continuing efforts to reduce perchlorate in drinking water. Regarding a study in Chile (Téllez et al. 2005), NAS (2005) recommended incorporating more extensive neurodevelopmental assessments of the children beginning in infancy and continuing through school age. Specific end points that should be assessed in follow-up studies include auditory function, including measures of otoacoustic emissions; visual attention; cognitive function, including tests for executive function and memory; and tests of motor function, particularly balance, coordination, and rapid finger

movements (NAS 2005). Increasing the sample sizes of the birth cohorts from the cities studied in Chile would be desirable to achieve appropriate statistical power for detecting possible differences among exposure groups on the developmental assessments. NAS (2005) also suggested that the question of whether or not exposures to perchlorate at concentrations present in municipal drinking water are related to an increased incidence of maternal hypothyroidism during gestation could be addressed in the study of pregnant women in Chile (Téllez et al. 2005) and that using larger samples would improve the precision of the estimates.

Biomarkers of Exposure and Effect.

Exposure. Potential biomarkers of exposure include perchlorate in urine, breast milk, iodide in blood and urine, and thyroid (T4, T3) and pituitary (TSH) hormones in blood. Perchlorate in urine is a biomarker that is specific for exposure to perchlorate; however, the biomarkers for iodine and thyroid hormones are not exclusive to perchlorate (changes may be produced by other anti-thyroid compounds and may be influenced by diet). One study found an association between urinary levels of perchlorate and drinking water concentrations of perchlorate (Valentín-Blasini et al. 2005). Further studies designed to correlate levels of one or more of these potential biomarkers with exposure levels would be useful to facilitate medical surveillance that can lead to early detection of exposure to perchlorate.

Effect. Biomarkers of effect for perchlorate also include levels of iodine in blood and urine, and thyroid (T4, T3) and pituitary (TSH) hormones in blood. Dosimetry has not been established to relate specific degrees of change in these markers to demonstrably adverse effects. Studies designed to perform this dosimetry would be useful to determine whether exposed populations may be experiencing adverse health effects due to perchlorate exposure.

Absorption, Distribution, Metabolism, and Excretion. Existing studies of absorption, distribution, metabolism, and excretion of perchlorate in humans provide information about the extent of absorption of ingested perchlorate and the extent and kinetics of urinary excretion of absorbed perchlorate (Anbar et al. 1959; Durand, 1938; Greer et al. 2002; Lawrence et al. 2000). These studies lend support to estimates of elimination half-time of absorbed perchlorate of approximately 8–12 hours. Studies in animals provide support for the above estimates as well as information about the tissue distribution and kinetics of elimination of perchlorate from various tissues after intravenous or oral exposures (Chow and Woodbury 1970; Chow et al. 1969; Clewell et al. 2003a, 2003b; Durand 1938; Fisher et al. 2000; Goldman and Stanbury 1973; Lazarus et al. 1974; Selivanova et al. 1986; Yu et al. 2002). The above

information has been used to support the development of PBPK models of perchlorate in adult humans, adult rats, pregnant rats and rat fetus, and lactating rats and rat pups (Clewell et al. 2003a, 2003b; Fisher et al. 2000; Merrill et al. 2003, 2005).

All of the above studies were by the oral or parenteral routes; no information is available regarding absorption following inhalation or dermal exposure, which although not relevant routes of exposure for the general population, are potential routes of exposure for perchlorate workers.

Comparative Toxicokinetics. Existing studies in humans and rats provide comparative information on the extent of absorption of ingested perchlorate, the routes of excretion of absorbed perchlorate, and the kinetics of excretion of absorbed perchlorate. PBPK models have been developed for the adult human and rat for the purpose of species extrapolation of oral or intravenous dosages of perchlorate (Clewell et al. 2003a, 2003b; Merrill et al. 2003, 2005). However, further research is necessary to determine how differences in thyroid physiology between humans and rats may affect the use of these models for human risk characterization and risk assessment. NAS (2005) identified a number of issues as potential data gaps with existing rat PBPK models including the need to: "(1) develop a more biologically-based description of placental transfer of perchlorate and iodide in rats, (2) determine whether perchlorate is transported by thyroid NIS if analytic methods of sufficient sensitivity can be developed or radiolabeled perchlorate with high radiochemical purity can be synthesized, (3) modify the adult human model to include the physiology of pregnancy and lactation to incorporate data from the recommended human clinical studies (if they are conducted), and (4) modify models to incorporate dietary iodide measurements from biomonitoring studies in pregnant or lactating women."

Methods of Reducing Toxic Effects. There are no established methods for reducing the toxic effects of perchlorate. Removal of the individual from exposure would also be effective since perchlorate is rapidly eliminated from the body (elimination half-time 8–12 hours) (Greer et al. 2002). Research into methods for reducing the toxic effects of perchlorate would enable treatment for individuals experiencing adverse health effects due to perchlorate exposure.

Children's Susceptibility. Data needs relating to both prenatal and childhood exposures, and developmental effects expressed either prenatally or during childhood, are discussed in detail in the Developmental Toxicity subsection above.

Children, infants, and the developing embryo and fetus may be especially susceptible to the thyroid effects of perchlorate because thyroid hormones are essential to normal growth and development. The embryo and fetus is dependent on maternal thyroid hormones prior to the onset of fetal thyroid hormone production at mid-gestation. Perchlorate studies in animals have shown that exposure of the mother to perchlorate during gestation, or even during lactation, can lead to reduced thyroid hormone levels and associated thyroid effects in the offspring (Brown-Grant and Sherwood 1971; Golstein et al. 1988; Lampe 1967; Postel 1957; York et al. 2001b, 2003, 2004). Further characterization of the toxicokinetics of perchlorate during pregnancy and lactation as well as comparisons of neurobehavioral tests between young animals exposed only *in utero* with animals exposed solely through lactation would provide valuable information.

Child health data needs relating to exposure are discussed in Section 6.8.1, Identification of Data Needs: Exposures of Children.

3.12.3 Ongoing Studies

The following ongoing research has been identified in the Federal Research in Progress database (FEDRIP 2005):

Dr. L.E. Braverman, from Brigham & Women's Hospital in Boston, Massachusetts, is conducting research aimed at determining the effects of low doses of oral potassium perchlorate on thyroid function in euthyroid male volunteers. His research is sponsored by the National Center for Research Resources.

Dr. N. Carrasco, from Yeshiva University in New York City, Albert Einstein School of Medicine, New York, is conducting research to characterize the molecular structure and function of the NIS that mediates the active translocation of iodide in the thyroid. Specific aims that Dr. Carrasco is pursuing include:

(1) to elucidate the regulatory mechanisms of NIS; (2) to establish structure-function relations in NIS; and (3) to assess NIS expression in the stomach under physiological and pathological conditions.

Dr. Carrasco's research is sponsored by the National Institute of Diabetes and Digestive and Kidney Diseases.

Dr. A.F. Firek, from the Department of Veterans Affairs, Medical Center in Loma Linda, California, intends to examine the effect of low-level perchlorate exposure for a period of 6 months in healthy euthyroid subjects. Specific end points that will be evaluated include incorporation of radioactive iodine

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into the thyroid, serum thyroid hormone and TSH levels, and thyroid gland morphology as assessed by physical examination. The research is sponsored by the Department of Veterans Affairs, Research and Development.